

... for a brighter future







A U.S. Department of Energy laboratory managed by The University of Chicago Your Synchrotron Powder Diffraction Instrument: 11-BM at Argonne's Advanced Photon Source

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# This document was presented at PPXRD -Pharmaceutical Powder X-ray Diffraction Symposium

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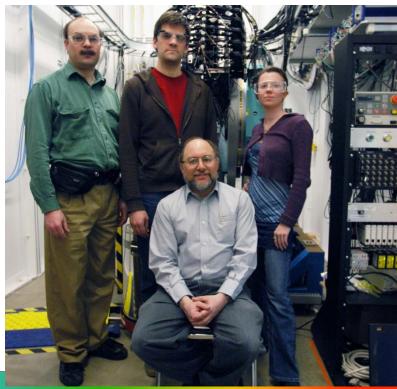


PPXRD Website – <u>www.icdd.com/ppxrd</u>

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- Instrument control: Xuesong Jiao & Tim Mooney
- Linux: Bill Sheehan & Dave Cyl
- Databases: Don Dohan & Yu Huang
- Web programming; David Carroll
- John Mitchell for launching the project
  - Ray Orbach for listening to him





#### Talk outline

What do you gain from powder diffraction at user facilities

- Why mail-in support?
- What does this require?
- How do you get access?
- What can be done with these data?

I I-BM: High-throughput powder diffraction without compromize



#### Part 1: High resolution powder diffraction, the next generation





## What do you gain from diffraction experiments at User Facilities?

- Neutrons: sensitive to many light atoms & provides excellent structural information
  - Few systematic errors
  - powder diffraction usually requires deuterium substitution
  - Short wavelength (TOF) high Q
- Synchrotron:
  - High resolution
  - Short wavelength
  - High sensitivity
- Less overlap, more observations: Better for structure solution; more accurate structures; subtle symmetry
- More observations, fewer corrections: more accurate...
- Subtle symmetry; minor phases

- or
- Very rapid measurements (10 patterns/sec)
- or
- Small beams (microns)
- Both: *in situ* & extreme temperature conditions



#### **Advantages of High-Resolution Powder Diffraction**

High-resolution diffraction - allows peaks to be resolved: essential for indexing and structure solution. Provides more observations: more reliable results

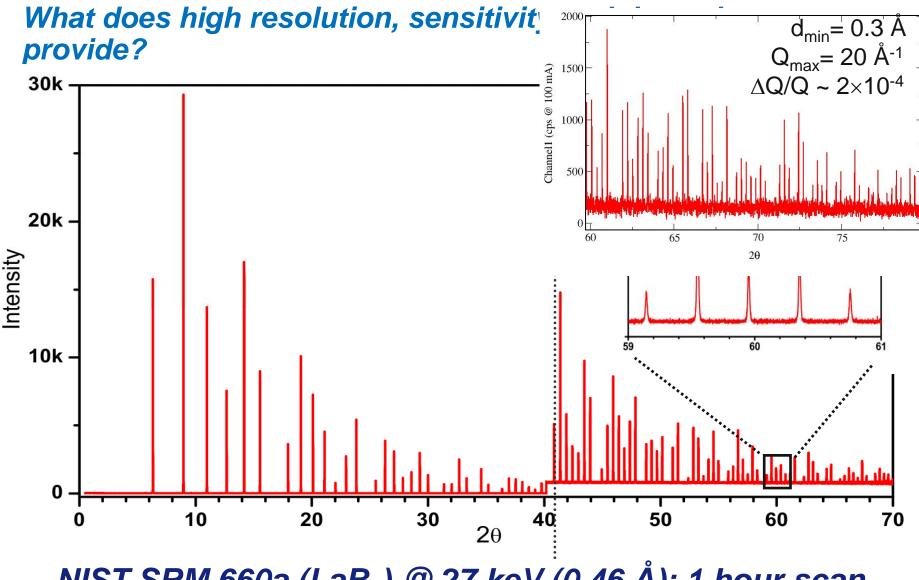
□**High-sensitivity diffraction** - allows weak peaks to be seen above background: essential for structural details (also more observations)

■Short wavelength – fewer systematic errors (absorption, extinction,...) – *more accurate* data. Wider Q range energy - *more observations*.

□No sample offset error – *no zero corrections*. Great for indexing.

□**High-Throughput diffraction -** allows these capabilities to be made available to the appropriate research communities in chemistry, materials, condensed matter physics, geosciences, pharmaceutical science, structural biology...





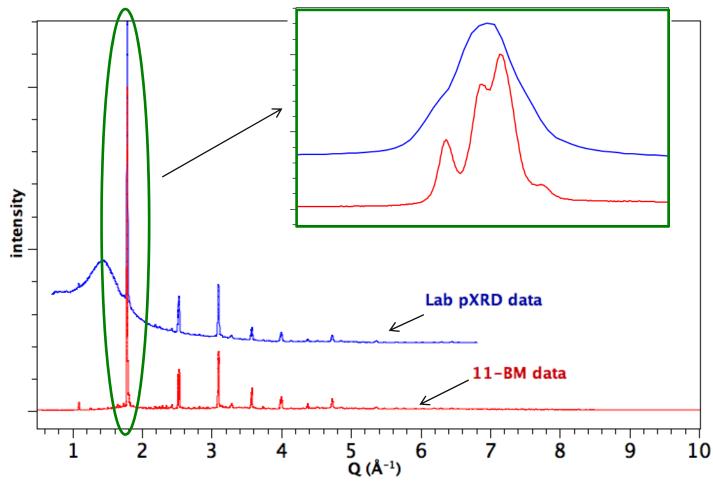
NIST SRM 660a (LaB<sub>6</sub>) @ 27 keV (0.46 Å): 1 hour scan



#### 11-BM: "Real World" examples I

Air-sensitive Rare-Earth Fluoride; Lab pXRD vs 11BM (in sealed glass capillary)

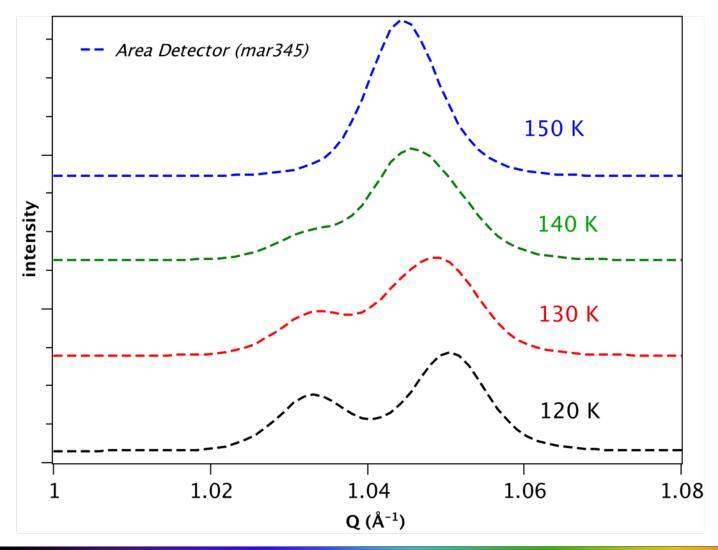
→ better background, sensitivity, Q-range & resolution !





#### 11-BM: "Real World" examples III

Structural Phase Transition in Organic-Inorganic Hybrid Material...





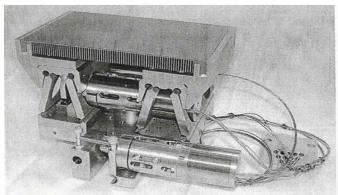
## Part 2: Design of an instrument to support routine high resolution, high throughput and high sensitivity diffraction

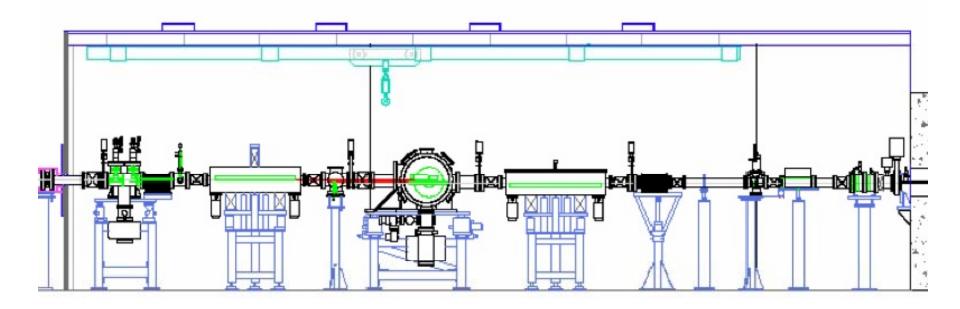




### **Beamline optics for high intensity and resolution**

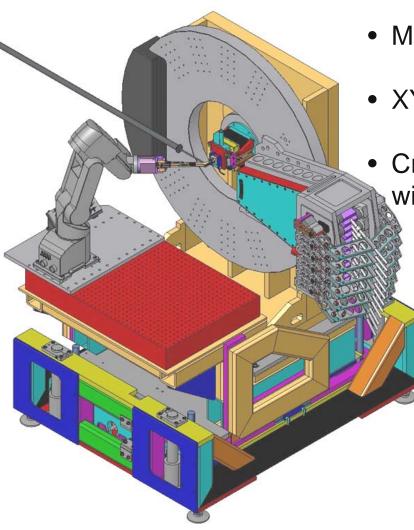
- Energy range 5.5-39keV (2.5 0.3Å)
- Energy resolution  $\Delta E/E \sim 10^{-4}$
- 1:1 sagittal focusing (optimal resolution)







### **Instrument:**

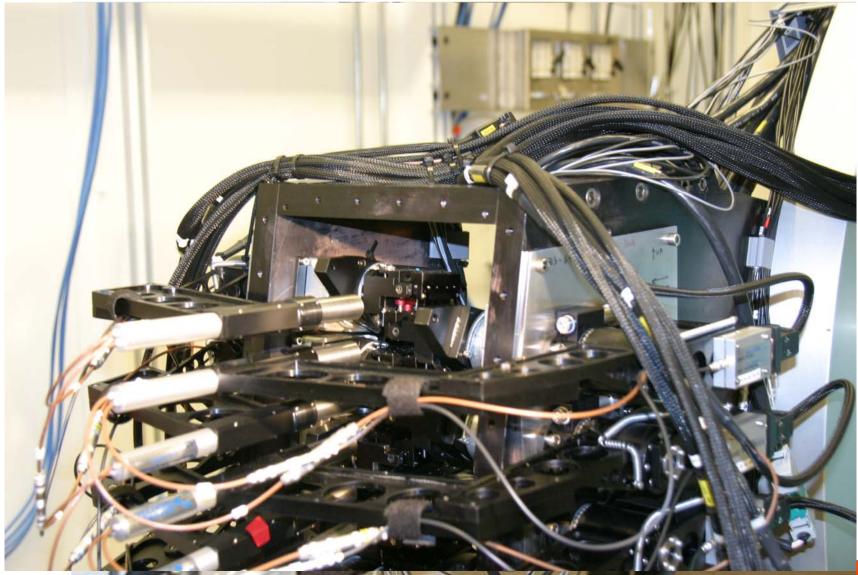


- 12 crystal two-axis analyzer detector speed data collection
- Mitsubishi robot unattended operation
- XYZ stage w/high speed spinner
- Cryostream 700+ -- unattended operation
   with data collection at 80 to 500 K



### 12 Analyzer/Detector System

#### As assembled



#### Make it easy to use

### For facile and routine access automation requires:

- 1. Must get many (>20) samples through per day
- 2. Resolution must be best available
- 3. Software must track samples and manage data
- 4. Software must be web-based and easy to use

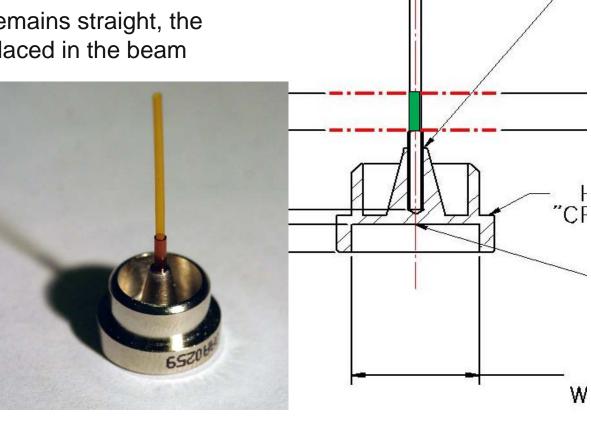




### **Standardized Sample Mounting**

- We supply sample mounting kits with polyimide tubing and barcode.
  - Users fill tube and ensure the their sample fills and is fixed in the green area.
  - As long as sample tube remains straight, the sample is automatically placed in the beam





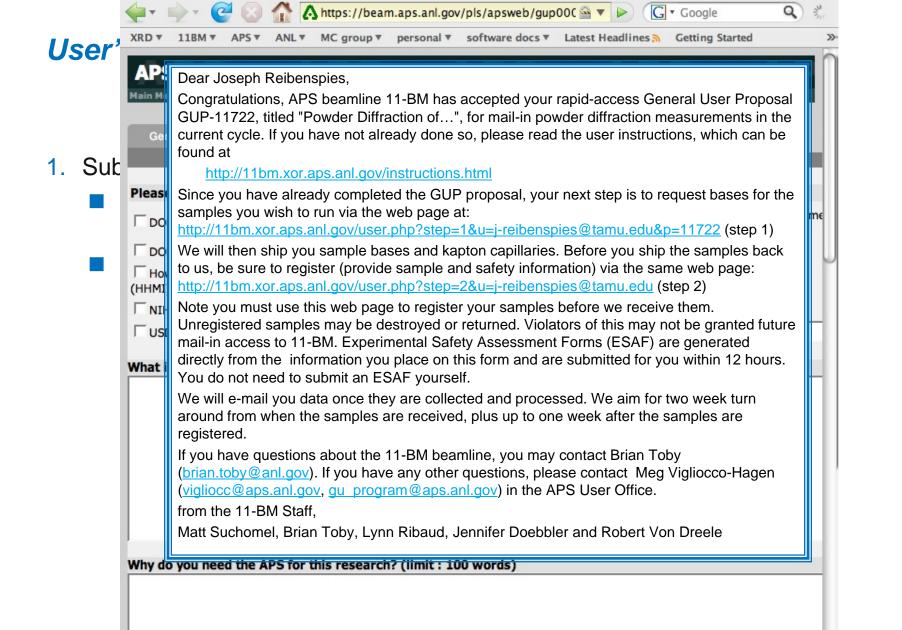
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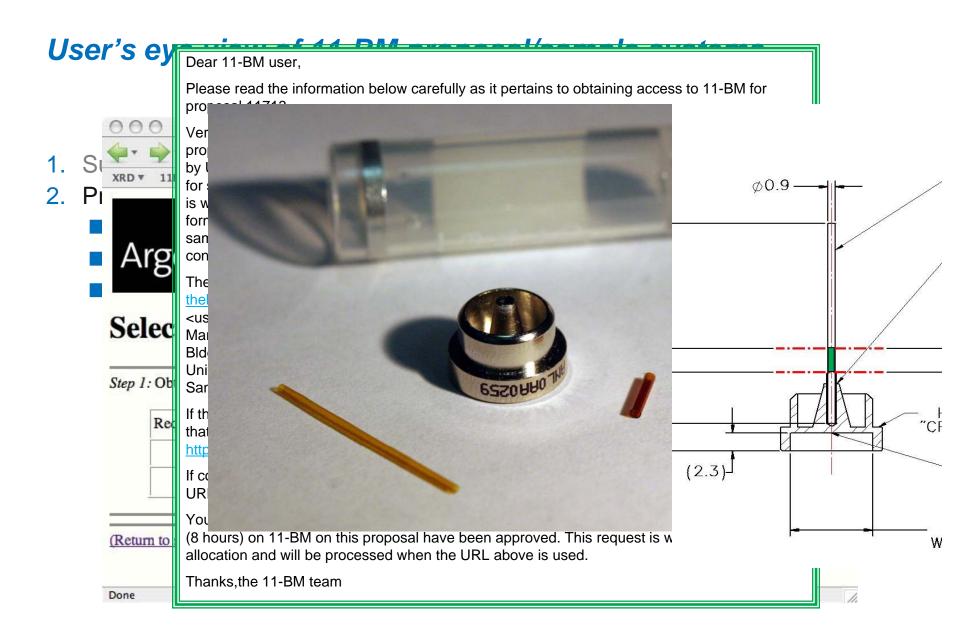
#### Part 3: Getting Access to 11-BM: The Infomercial



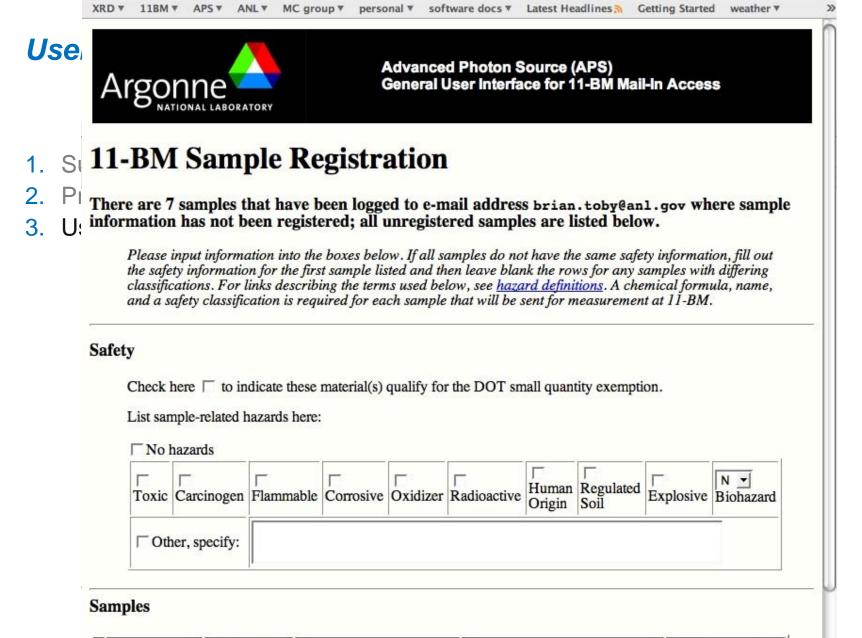












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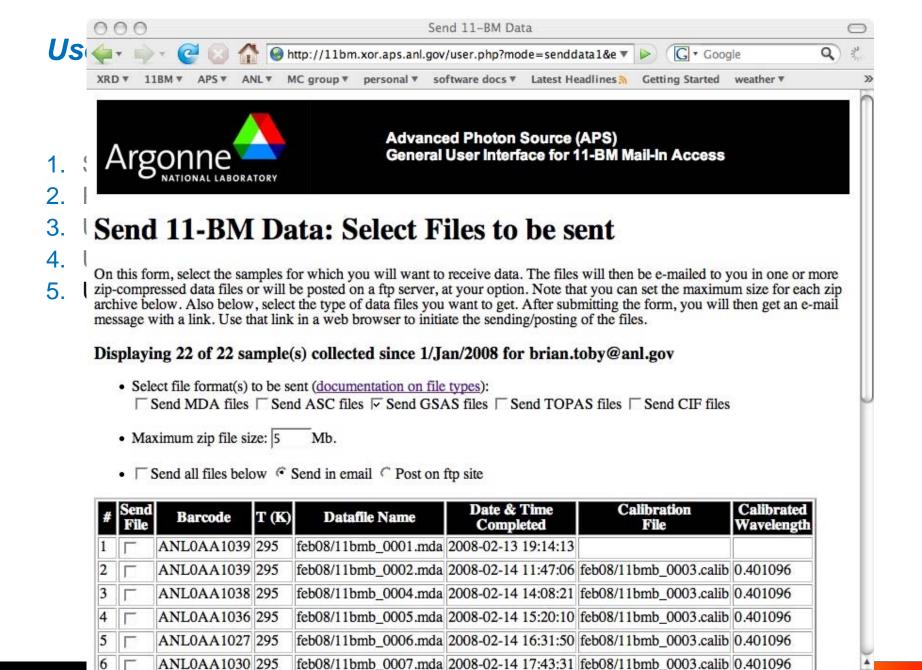
#	Barcode	Collection Temperature	Chemical Formula	Chemical Name	Sample ID (optional)
1	ANLOAA0534	Ambient 💌			
2	ANLOAA0535	Ambient 💌			

#### User's eye view of 11-BM proposal/sample systems

- 1. Submit an APS general user proposal.
- 2. Proposal is accepted
- 3. User supplies sample & safety information for each barcode
- 4. User sends mounted samples to APS







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#### User's eye view of 11-BM proposal/sample systems

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#### What does it cost?

- Non-proprietary work: **Free**
- Proprietary: US law requires cost recovery, <\$400/hour</p>

#### How long does it take?

- Most users receive sample bases within a week of proposal submission
- Most data is collected within a week (almost always within two weeks) of when the samples reach us.

Note that completion of paperwork (DOE User Agreement) is required for proposal submission. This can take some time, so this is good to start this in advance.



#### What's new: On-Site Use

Experimenters who have more extensive measurements, or ones where direct control of data collection

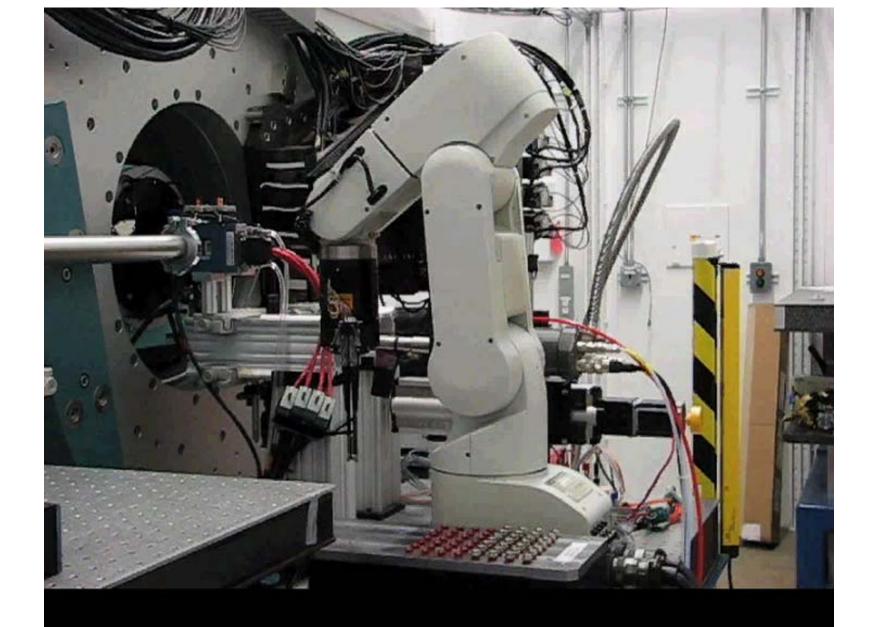
- High temperature (>200 C) or complex measurement protocols
  - When available: low T (<90 K)</li>
- Large numbers of samples
- Non-routine sample mounts
- In situ experiments



#### Coming soon:

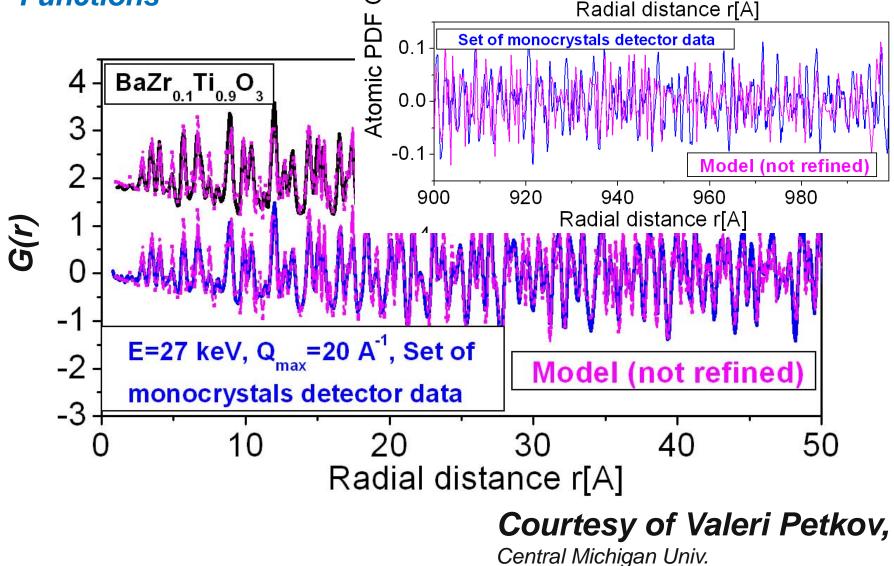
- Wider temperature range for on-site work: soon up to 1000 C, later down to ~10 K. (Note: 90 K – 500 K for mail-in)
- Facile wavelength changes for resonant scattering
- Area detection data collection systems
- Enhanced data security for proprietary measurements need some help from this community







## High-resolution Powder Diffraction and Pair-Distribution Functions Radial distance r[A]





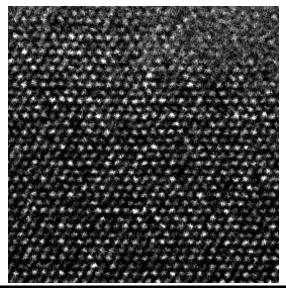
## What can be done with high-resolution powder diffraction?

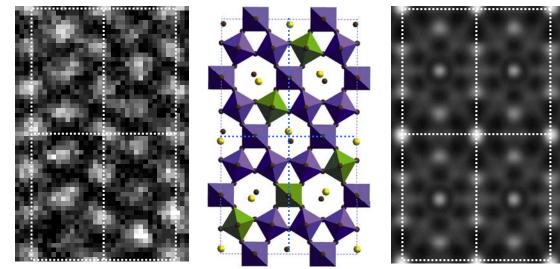


#### Propane Amoxidation: a decade+ mystery unraveled

Poorly understood, but commercially very important catalyst mixture, even ~10 years after discovery

- Breakthrough by group of Doug Buttrey (U. of DE) *et al.* by applying best-inworld research tools:
- Combinatorial synthesis: enhanced concentration of each phase (Symex)
- Atomic imaging by TEM
- High resolution X-ray powder diffraction (X7A @ BNL) allowed indexing of unit cells (correctly)





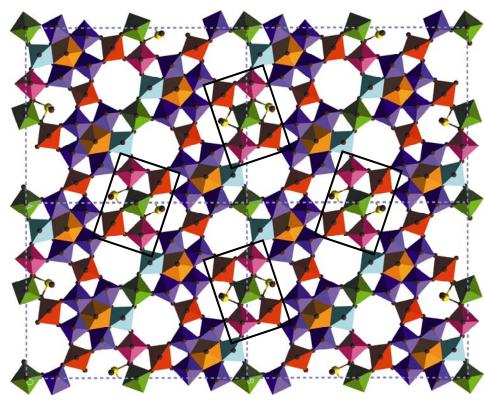


## Armed with world-class tools, the structure & function could be explored

### Crystallographic analysis:

- X-ray: metal sites
- Neutron: oxygen sites
- Combined fit:
  - Accurate M-O distances
  - Allows assignment of metal siting and *valences*
- Full structure: 16 metal sites (Mo,V,Te,Nb), 30 O atoms

Based on this detailed structure, a formal catalysis mechanism was proposed!



P. DeSanto Jr, D. J. Buttrey, R. K. Grasselli,C. G. Lugmair, A. F. Volpe Jr., B. H. Toby,& T. Vogt, *Topics in Catalysis*, 2003.



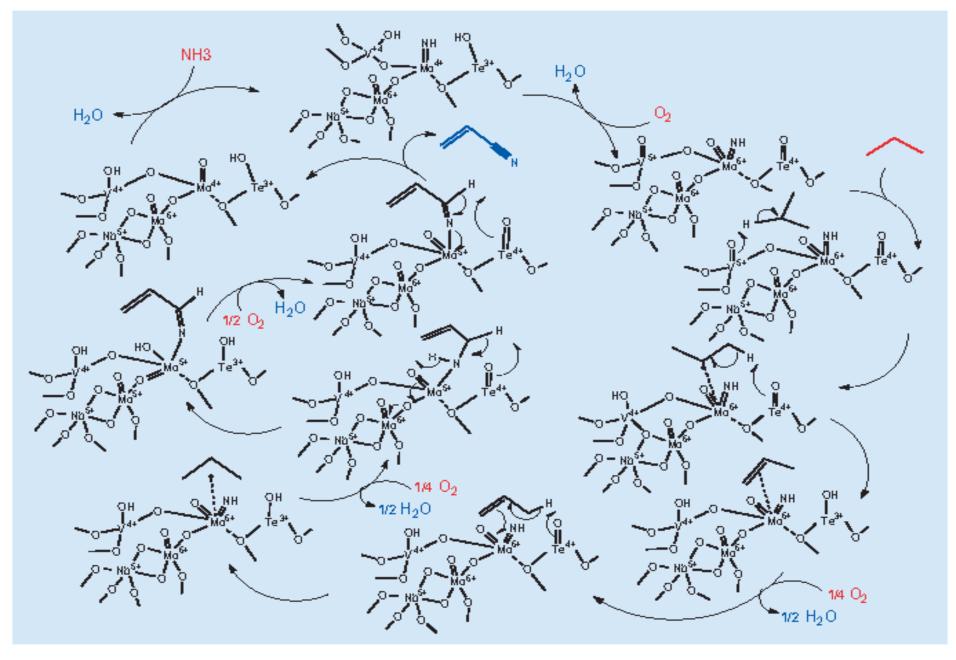
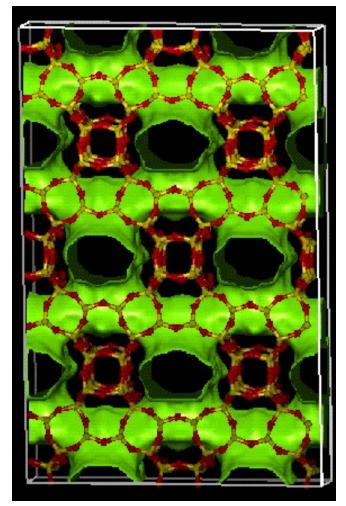


Fig. 2. A schematic mechanism for propane ammoxidation by the M1 catalyst that details the cation centers believed to be responsible for reactive process [3].

#### **Powder Diffraction: Principle research tool for Zeolitic Materials**

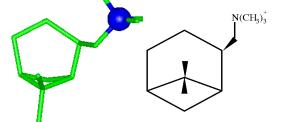
- Zeolites are <u>porous</u> aluminosilicate minerals [naturally occurring]
- ...built from tetrahedral SiO<sub>4</sub> and AlO<sub>4</sub> units
- Charge balance requires extra-framework cations:  $M^{+n}_{x/n} (AlO_2)^{-}_x (SiO_2)_{1-x}$
- Many other framework atoms can be used to create *zeolitic* materials
- Used for catalysis, selective adsorption, ion exchange...





#### CIT-1: Unraveling how zeolites form

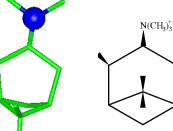
Many zeolitic materials are synthesized using organic "structure directing agent" (SDA) cations. This example illustrates how structural analysis validates modeling and explains the nature of zeolite-SDA interactions and how they drive synthesis.



▲ <u>1</u> Makes CIT-1 (~1% stacking faults)

**<u>2</u>** Makes SSZ-33 (similar to CIT-1, but with >30% stacking faults) ►





•  $\underline{3}$  Cannot be used to make either CIT-1 or SSZ-33

There are no obvious differences. Research problem: Why do the cations behave differently?



#### **CIT-1: Conclusions**

- Use of combined synchrotron and neutron diffraction gave cation siting: contradicts Monte-Carlo "docking" results
- Confirmed as unique solution by molecular modeling
- Explained why SDA <u>3</u> does not make CIT-1
- Stacking faults likely a kinetic phenomenon

Toby, Brian H., Khosrovani, Nazy, Dartt, Christopher B., Davis, Mark E., and Parise, John B., "Structure-directing Agents and Stacking Faults in the CON System: A Combined Crystallographic and Computer Simulation Study.", *Microporous and Mesoporous Materials* <u>39</u>, 77 (2000).

