



A Novel Method to Determine Amorphous Phase Concentrations Using the Rietveld Refinement Technique

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$y_{ci}(2\theta_i) = \sum_{k} \sum_{i=1}^{2} \phi(2\theta_i - 2\theta_k) L_k P_k A + y_{bi}$

1. Structure factor, F_K: Crystallography 2. Phase Scale, S: Phase concentration 3. ϕ Sample Microstructure and instrument: optics, L, A 4. Background, **y**_h:



 $Sy = \sum w_i (y_i - y_{ci})^2; w_i = 1/y_i$





Line Broadening



Peak diffraction profile ϕ (2 θ_{κ})

Convolution of

Instrumental Function (F_{instrument})

and Sample Function (F_{sample})

 ϕ (2 θ_{K}) = F_{instrument} \otimes F_{sample}





Sample function: F_{sample}

 $\mathbf{F}_{\text{sample}} = \mathbf{F}_{\text{crystallite size}} \otimes \mathbf{F}_{\text{microstrain}} \otimes \mathbf{F}_{\text{size}_\text{distribution}}$

Crystallite size (Domain coherence) Debye-Scherrer \Rightarrow fwhm α 1/Cos θ

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Microstrain \Rightarrow fwhm \alpha Tan \theta
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Crystallite size distribution
Log-normal distribution: the logarithm of the
crystallite size distribution follow a Gaussian
distribution \Rightarrow fwhm \alpha \sigma; (independent of \theta)
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Scale factor -> Phase concentration

W_i (%) = S_i (ZMV)_i / Σ_p S_p (ZMV)_p

W_i weight percent of phase i
S_i Phase scale factor
ZM Unit cell mass
V Unit cell volume





 $H_{L} = F_{h}(\alpha_{z})/\cos\theta$

$F_h(\alpha_z) = \lambda \sum a_{lmp} Y_{lmp}(\theta, \Phi)$







The Rietveld technique can be used to study nanocrystalline materials



Can the Rietveld technique be used to analyze amorphous materials?





Silicon oxide has several low temperature polymorphs:

Stishovite, Tetragonal

Beta-Cristobalite, Cubic

Alpha-Cristobalite, Tetragonal



Modeled with Beta Cristobalite



Modeled with alpha cristobalite







Nominal concentrations: 50 wt % Amorphous silica 50 wt % Rutile



$y_{ci}(2\theta_i) = S \sum_{k} |F_{k}|^2 \phi (2\theta_i - 2\theta_k) L_{k} P_{k} A + y_{bi}$

Improving Instrument modeling or instrument optic

Background: Simple model Chevychev Polynomial

It is necessary to take into account the Inelastic Compton scattering



Crystallization of amorphous lactose

Lactose crystallizes in three polymorphs:

Alpha Lactose hydrated, Monoclinic

Beta Lactose, Monoclinic

1/1 Alpha/Beta phase, Triclinic

The Amorphous phase was modeled with the

Crystallography of the Triclinic phase







Amorphous Indomethacin:

Crystalline Alpha-Form

Crystalline Gamma-Form

From the NMR experiments: The local atomic order of the amorphous Phase is similar to that in the Alpha-form





- The Rietveld refinement technique can be used to obtain Information about the concentrations of the phases in a sample inclusive when one of them is amorphous.
- The amorphous phase should be modeled with the crystalline structure of the phase that crystallizes first.
- The analysis requires to improve the model for the background and the optic of the diffractometer.



Some codes to refine crystalline structures

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