

An internal standard for pharmaceuticals

The Art of dealing with compromise

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MOTIVATION

Why **small traces**?

(Early detection contaminants/degradation products/ crystalline seed/highly potent)

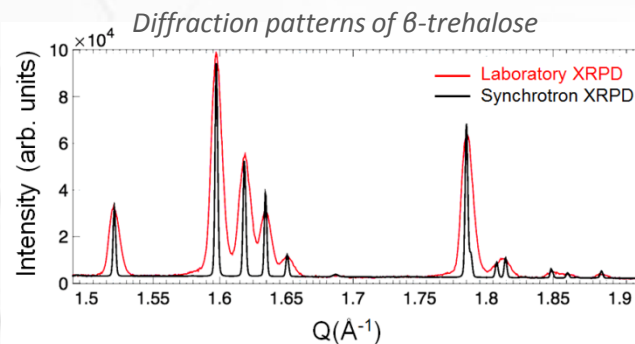
Actual status of QPA of **pharmaceuticals**?

Why quantifying on an **absolute** scale?



CHALLENGES

Pushing **instrumentation** limits



Tailoring methodology to characteristics of pharmaceuticals

Choice of quantification method

Why **small traces**?

Actual status of QPA of **pharmaceuticals**?

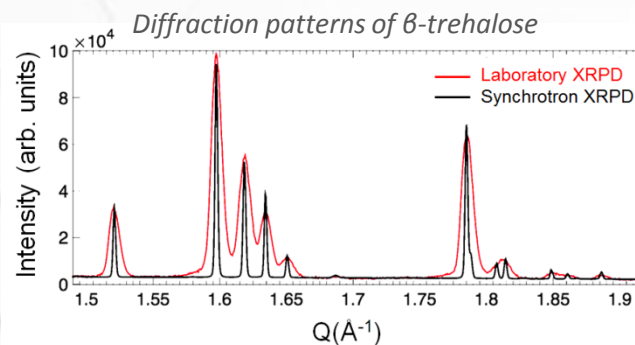
(Tailored properties, Alumina?,
unique vs. set of standards, PONKCS)

Why quantifying on
an **absolute** scale?



N. V. Y. Scarlett & I. C. Madsen,
Powder Diffraction, **21**, 4, 278-284 (2006). PONKCS method

Pushing **instrumentation** limits



Tailoring methodology to
characteristics of pharmaceuticals

Choice of quantification method

MOTIVATION

Why **small traces**?

Actual status of QPA of **pharmaceuticals**?

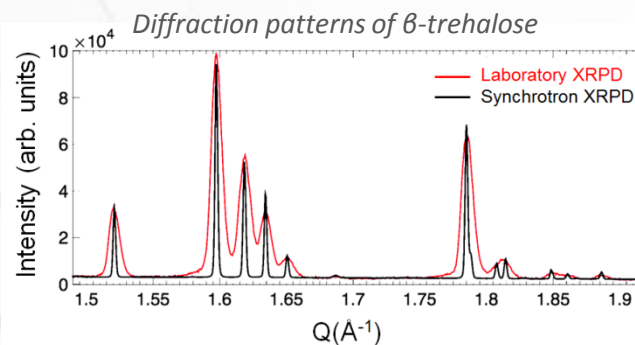
Why quantifying on an **absolute scale**?

(relative scale: invisible amorphous/unknown, ex: interconversion to amorphous)



CHALLENGES

Pushing **instrumentation limits**



Tailoring methodology to characteristics of pharmaceuticals

Choice of quantification method

MOTIVATION

Why **small traces**?

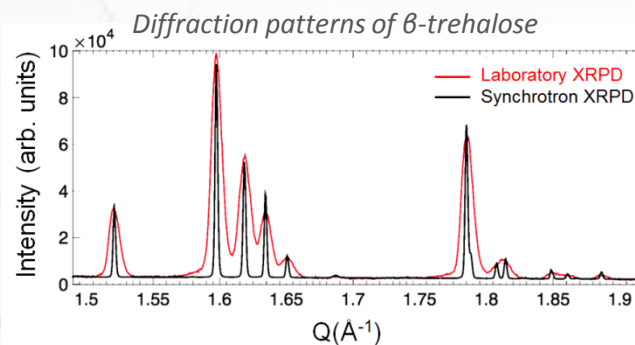
Actual status of QPA of **pharmaceuticals**?

Why quantifying on an **absolute** scale?



CHALLENGES

Pushing **instrumentation** limits



Tailoring methodology to characteristics of pharmaceuticals

Choice of quantification method

Pushing instrument limits

Synchrotron radiation +
Position sensitive detector +
capillary geometry

LoQ < 0.05 wt%, LoD < 0.01 wt%

High angular (FWHM) resolution

Data collection efficiency

Data modelling

No transparency effect

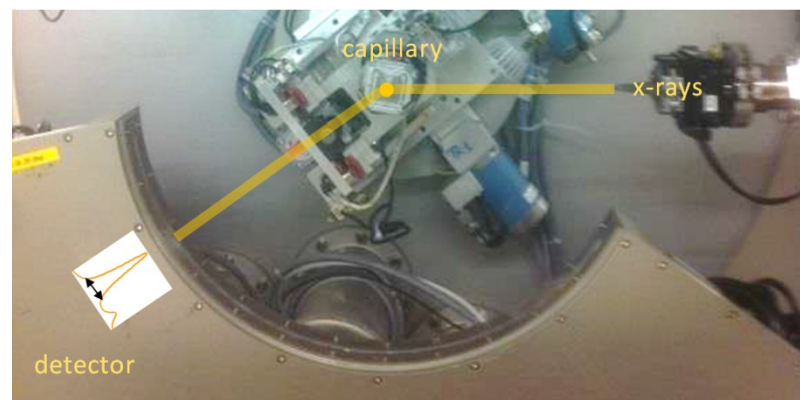
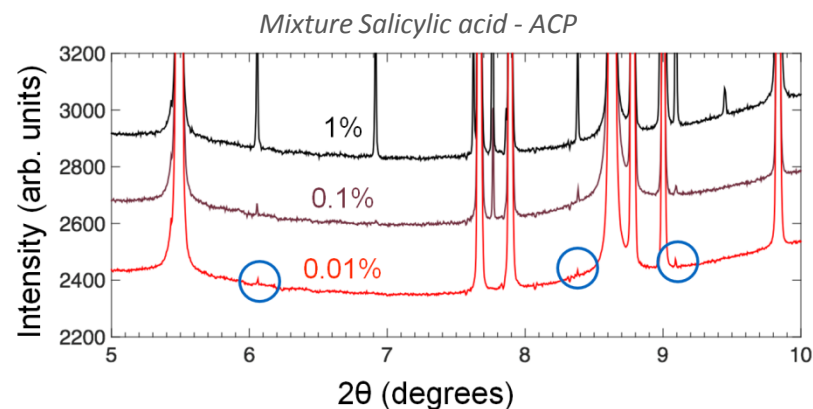
Tunable wavelength



Particle statistics/homogeneity
of distribution

Low scattered intensity

Reproducibility cap. patterns



VS



Tailoring methodology

Know your sample: **organics**

Light molecules, poor scattering power, large unit cells, low symmetry, peak overlapping, radiation sensitive, low absorption...

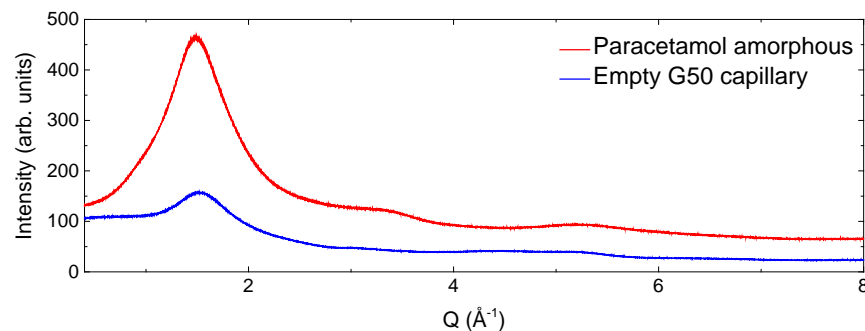
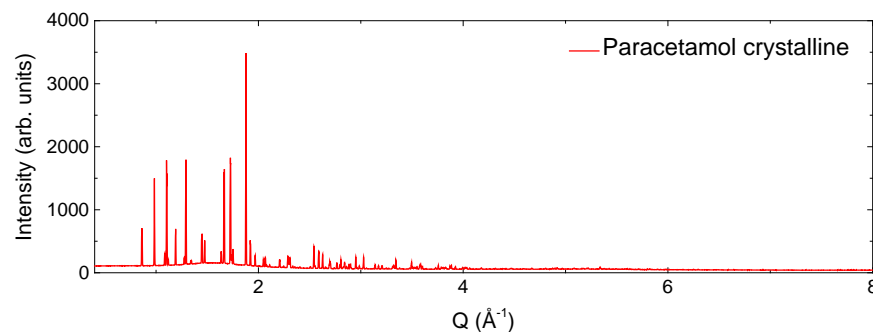
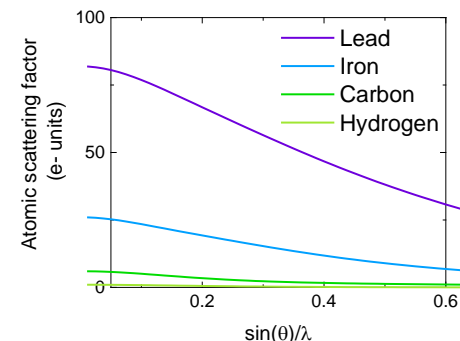
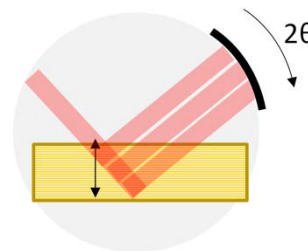
Spatial inhomogeneities

⇒ sample more powder volumes

Semi-crystalline materials

Degree of Crystallinity (DoC)

Correlation amorphous contributions sample vs container



Choice of quantification method

Internal standard method

Calibration curve: Y/N

- Direct correction for instrumental effects
- Comparable matrix effects
- Unknown compounds
- Amorphous quantification, Absolute scale
- Universal



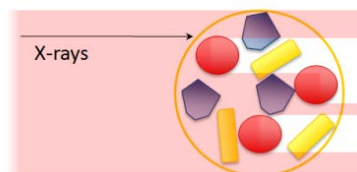
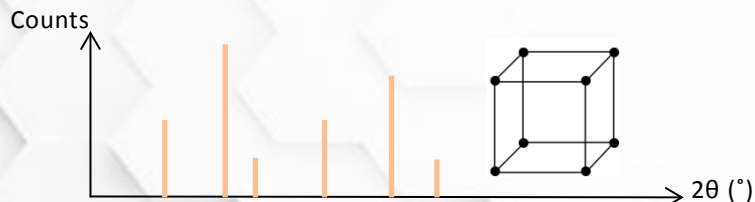
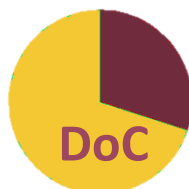
- Internal standard tailored to analyte
- Time consuming powder processing
- Powder samples only
- Analyte mixture contamination



Which internal standard?

```

Structure of Sucrose
:
_journal_name_full      'Personal communication to COD'
_journal_year          2013
_chemical_formula_moiety 'C12 H22 O11'
_chemical_formula_sum   'C12 H22 O11'
_chemical_formula_weight 342.30
_space_group_IT_number  4
_symmetry_cell_setting  monoclinic
_symmetry_int_tables_number 4
_symmetry_space_group_name_Hall 'P 2y#'
_symmetry_space_group_name_H-M 'P 1 21 1'
_atom_sites_solution_hydrogens geom
_atom_sites_solution_primary direct
    
```



Crystallite diameter (μm)	40	10	1
Crystallites (20 nm^3)	5.97×10^5	3.82×10^7	3.82×10^{10}
Number diffracting	12	760	38 000
σ_{ps}	0.289	0.036	0.005

C Madsen and N. V. Y. Scarlett in *Powder Diffraction: Theory and Practice*, 2008
 Robert E. Dinnebier and Simon J. L. Billinge, Print ISBN: 978-0-85404-231-9, DOI:10.1039/978184755823
 Deane K. Smith Powder Diffraction, 16, pp 186-191, (2001), doi:10.1154/1.1423285

Compromises ...

Alumina

Nice peak shape

Known DoC (NIST standard
SRM676 series)

					2 He
5 B	6 C	7 N	8 O	9 F	10 Ne
13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br
					36 Kr

Density ca. 3.9 g.cm⁻³

LAC (12.4 keV) = 37.3 cm⁻¹

Small wt% ⇒ weighting errors

Compromises ...

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Density ca. 3.9 g.cm^{-3}

LAC (12.4 keV) = 37.3 cm^{-1}

Small wt% \Rightarrow weighting errors

Beyond Alumina

Diamond

LAC (12.4 keV) = 3.2 cm^{-1}

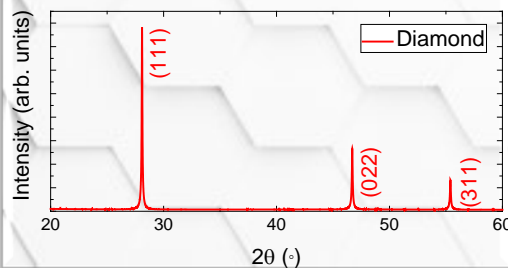
Density ca. 3.5 g.cm^{-3}

Small wt% \Rightarrow weighting errors

Very few peaks $< 60^\circ 2\theta$

\Rightarrow limited redundancy for peak overlap

Purity of grinding media

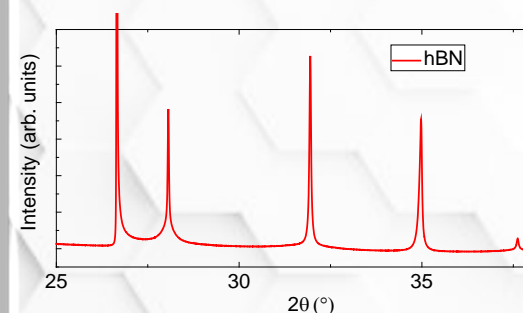


hBN

Density ca. 2.1 g.cm^{-3}

LAC (12.4 keV) = 2.9 cm^{-1}

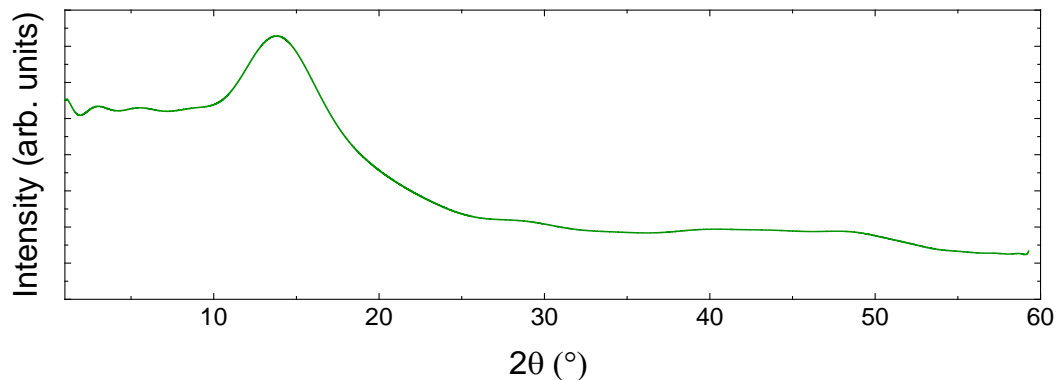
Helps to homogenize blends



Flakes \Rightarrow difficult peak shape
Standard Rietveld refinement won't work

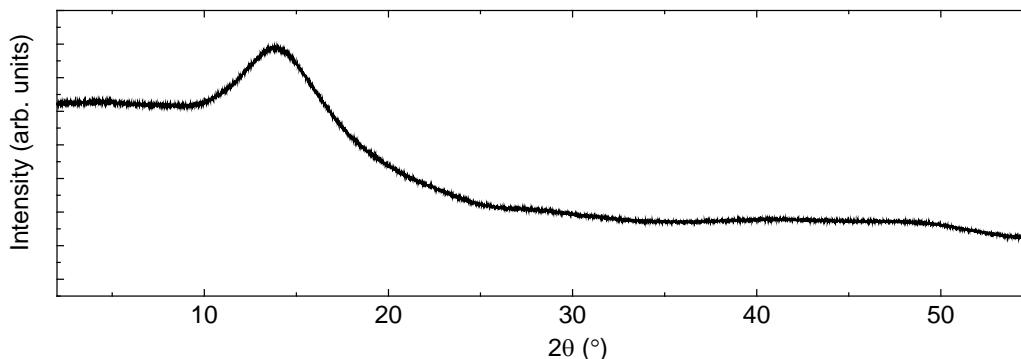
Alternative analysis strategy

High order polynomial:



Main separate contributions (physically-based background):

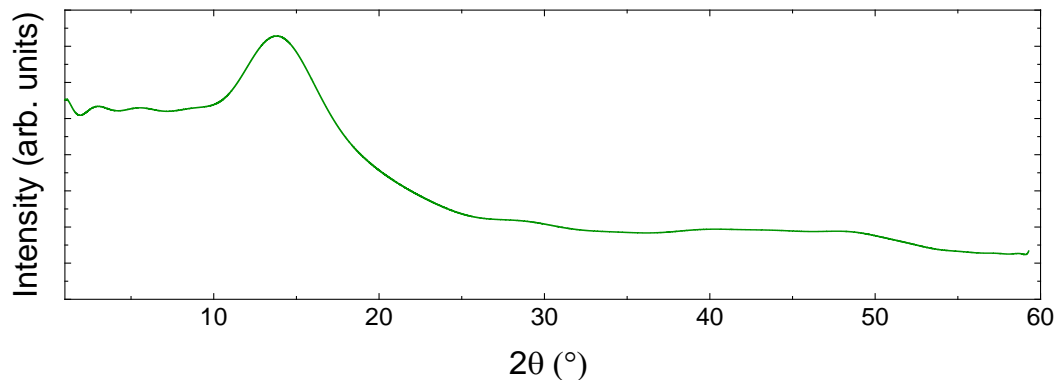
Empty



Modelling or subtracting?

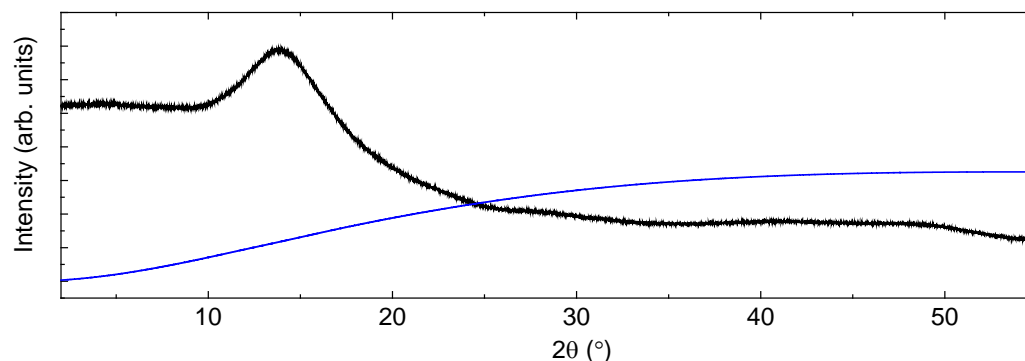
REAL samples: Amorphous pattern **not always** available
(i.e. calibration curve amorphous/crystalline not an option)

High order polynomial:



Main separate contributions (physically-based background):

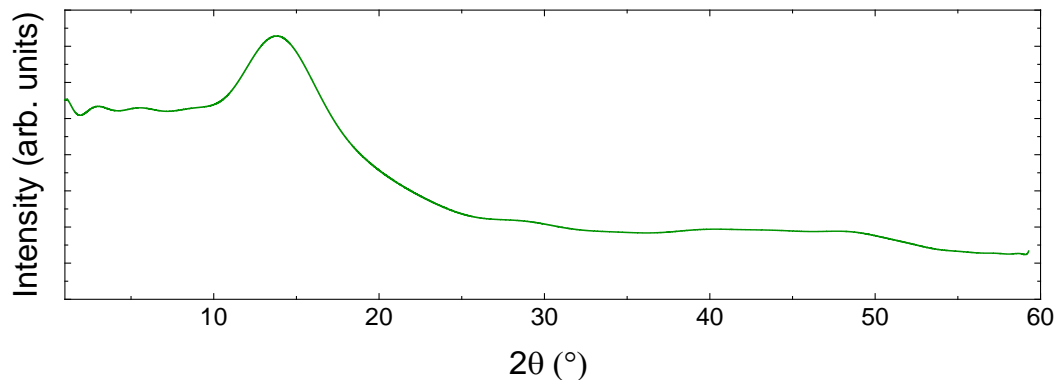
Empty
Compton



Modelling or subtracting?

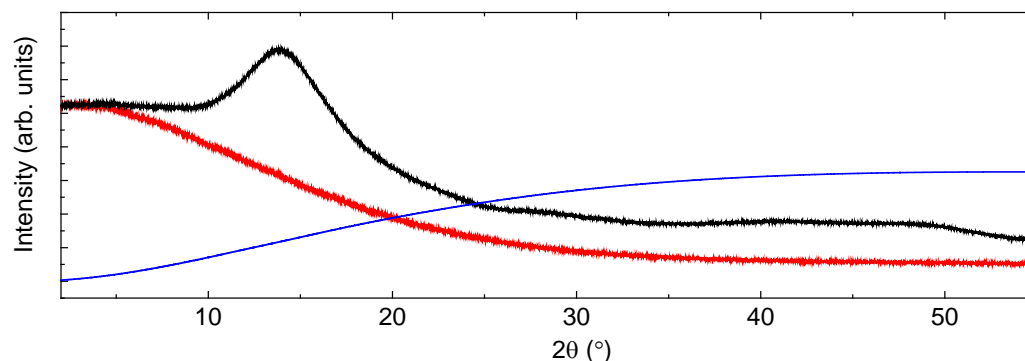
REAL samples: Amorphous pattern **not always** available
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High order polynomial:



Main separate contributions (physically-based background):

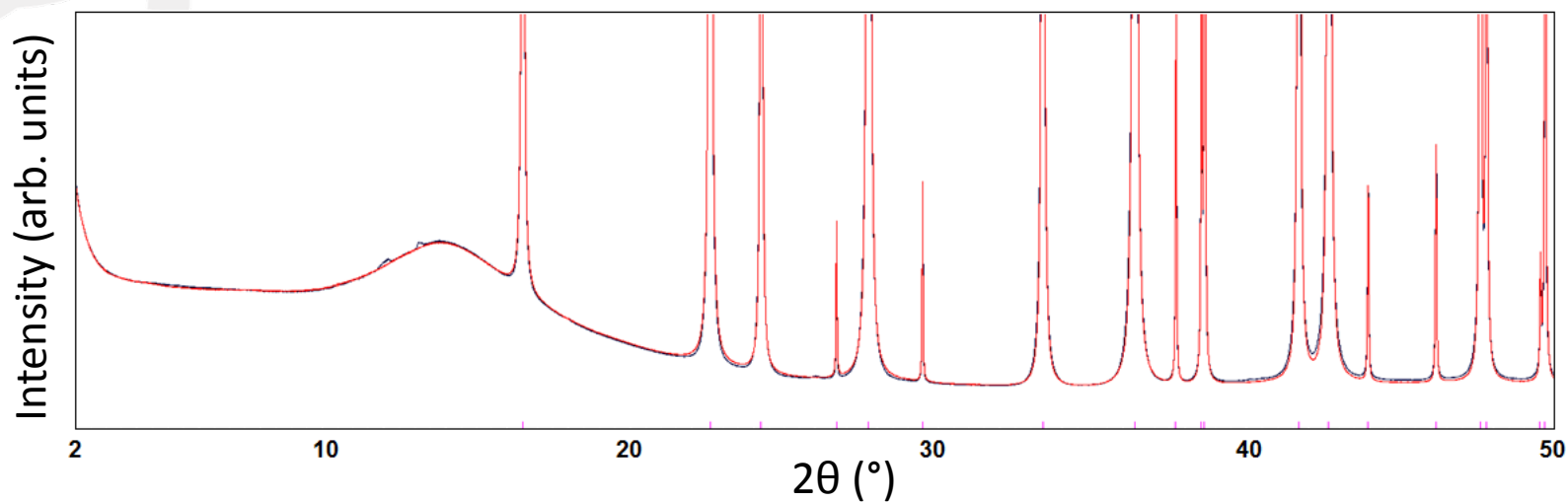
Empty
Compton
Air



Modelling or subtracting?

REAL samples: Amorphous pattern **not always** available
(i.e. calibration curve amorphous/crystalline not an option)

Alumina DoC



Background fits nicely with identified diffuse contributions

Expected DoC between 99 and 100 wt%:
we are NOT sensitive to <1 wt% amorphous content

- Test method against *ad-hoc physical mixtures* of amorphous /crystalline lactose

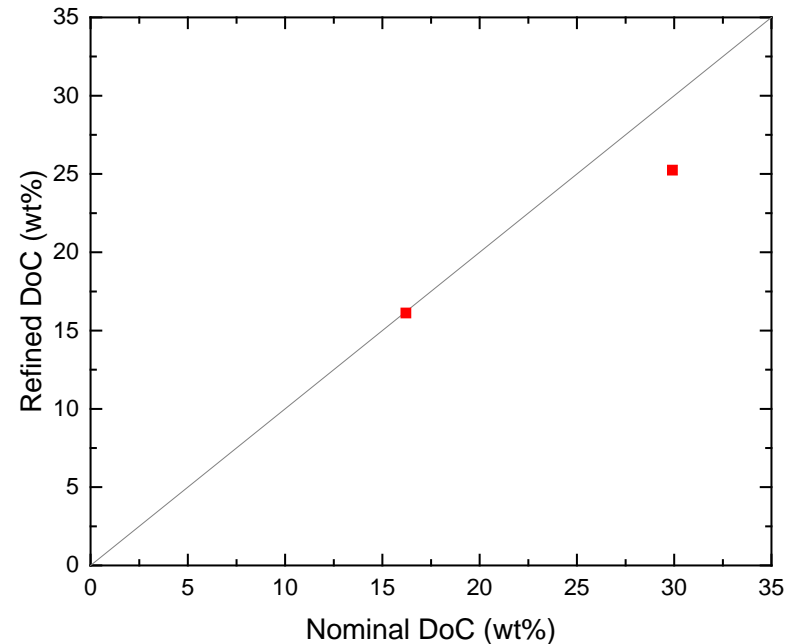
Step 1:

DoC of pure ‘crystalline’
lactose: 97% DoC

Step 2:

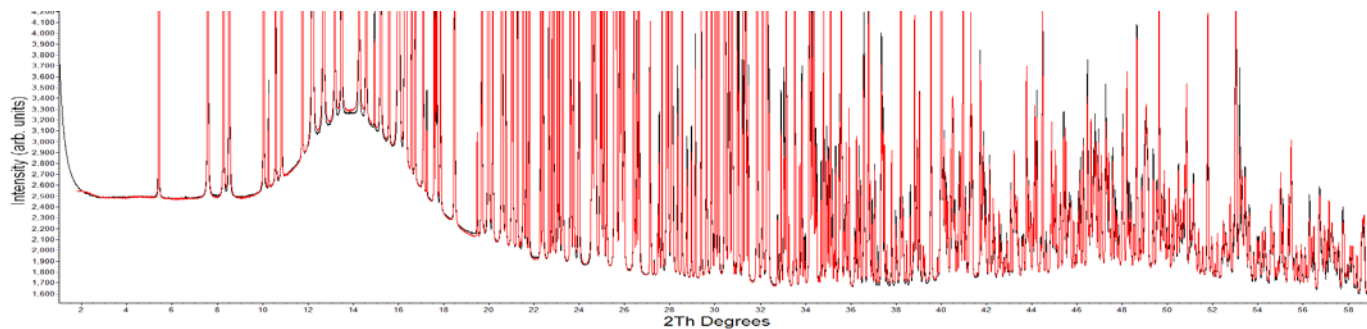
Refinement of the amorphous phase ‘by difference’

Ad-hoc mixtures of lactose: relative error between weighted and refined ratio in the range 0.5 – 16 %



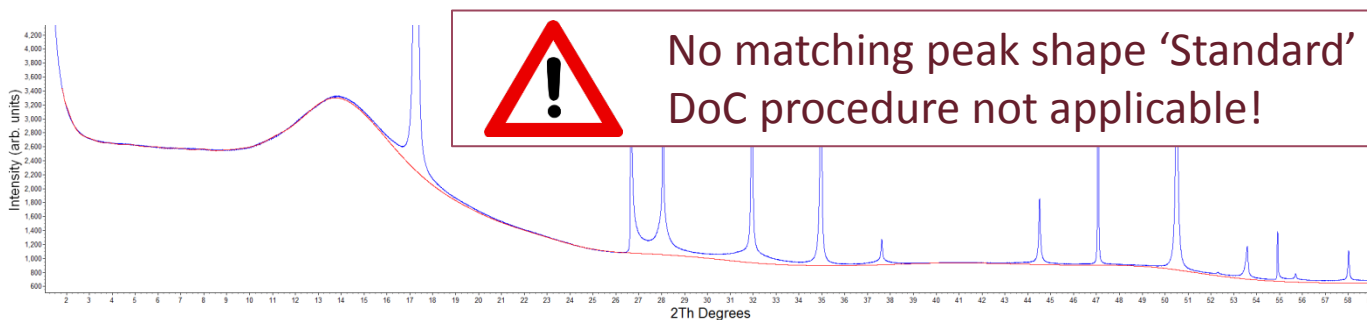
Sucrose

DoC ca. **96.4 wt%**



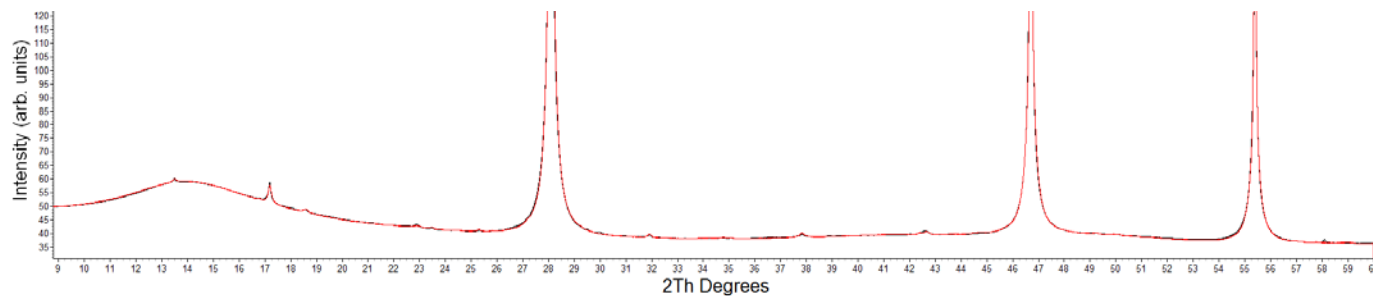
hBN

DoC ca. **99 wt%**

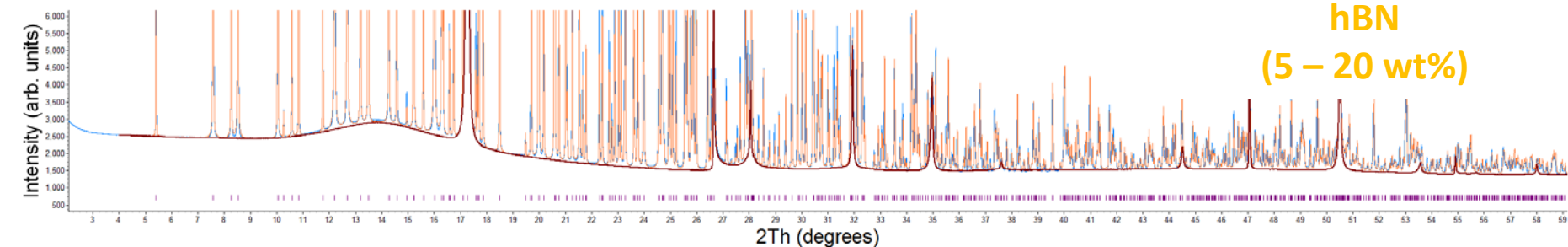
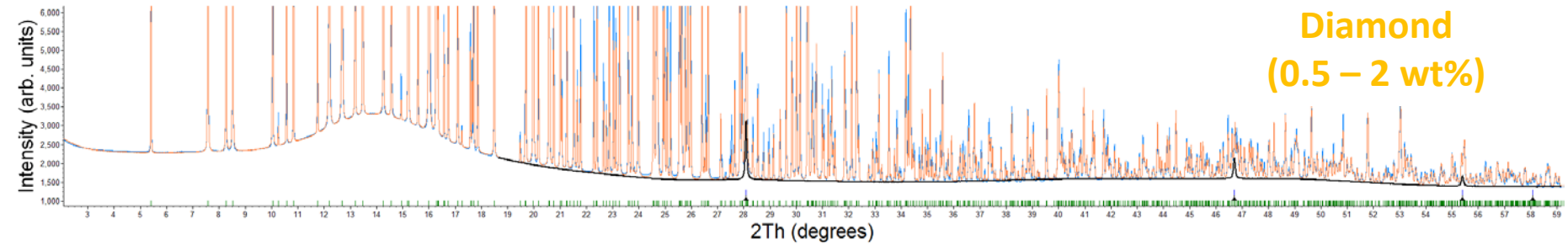
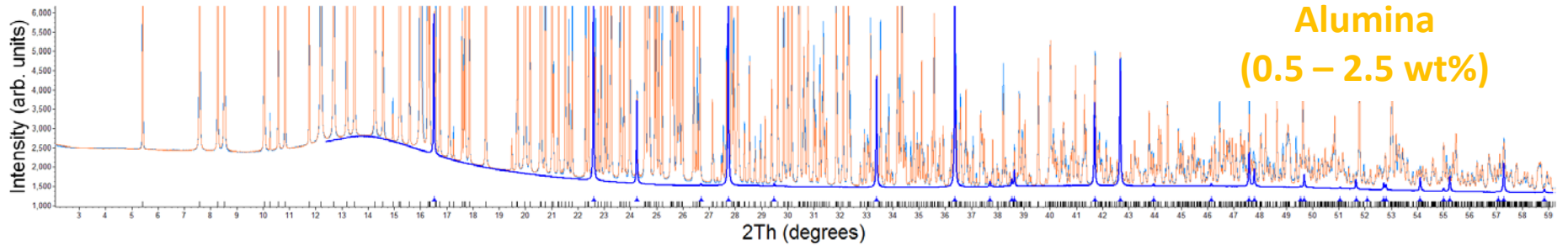


Diamond

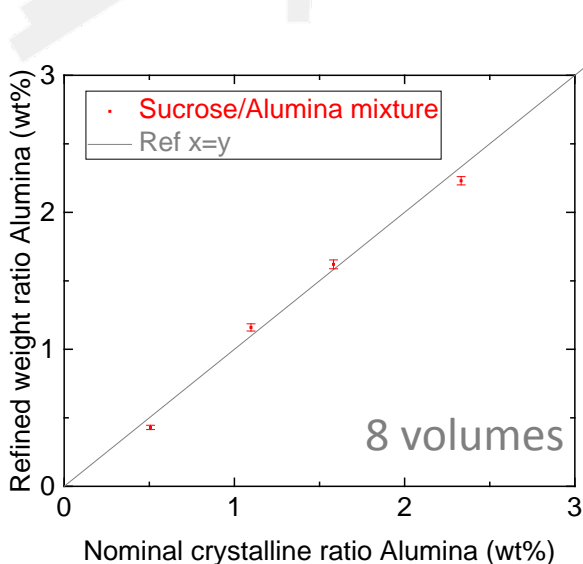
DoC ca. **95.8 wt%**



- ✿ Sucrose/Alumina-hBN-Diamond mixtures
- ✿ Small wt% to improve distribution homogeneity and adapted to scattering power

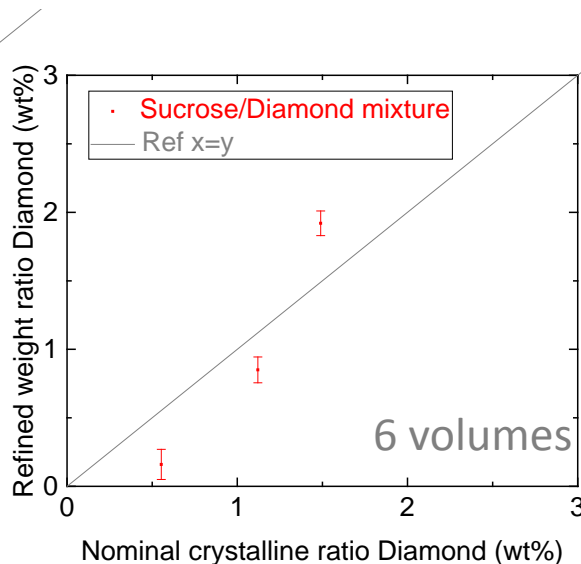


Comparison calibration curves internal standard/sucrose



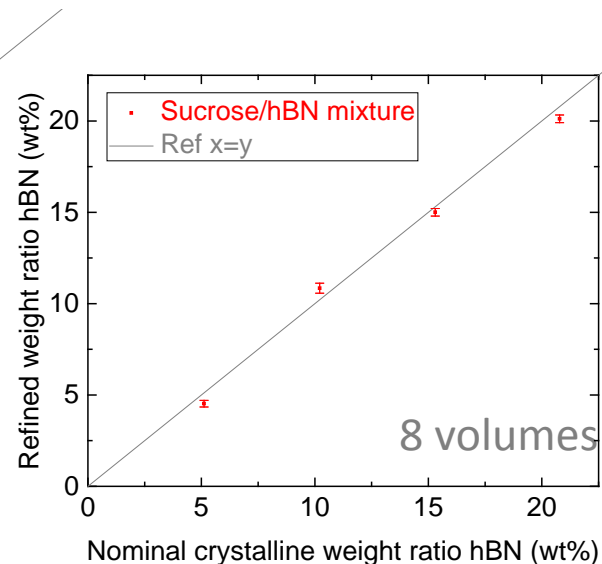
Alumina
(0.5 – 2.5 wt%)

Standard Rietveld
refinement



Diamond
(0.5 – 2 wt%)

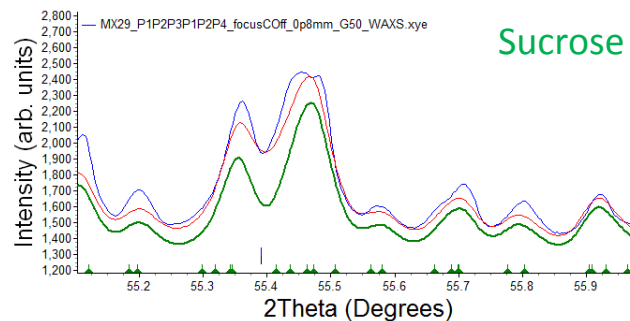
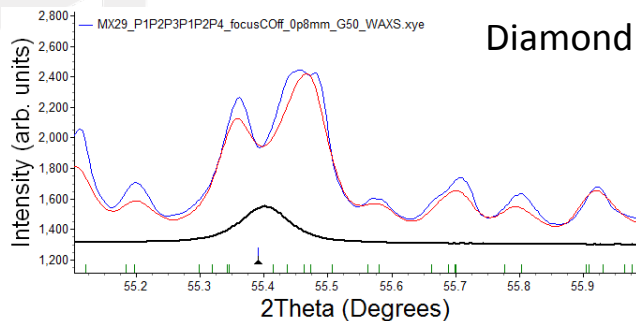
Standard Rietveld
refinement



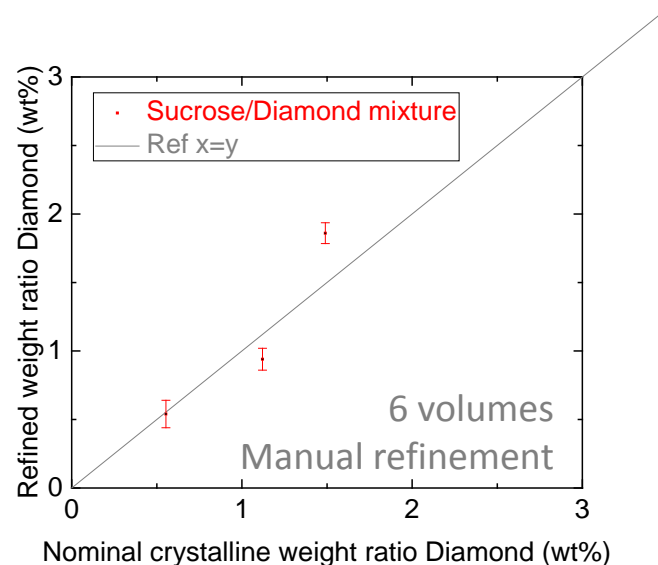
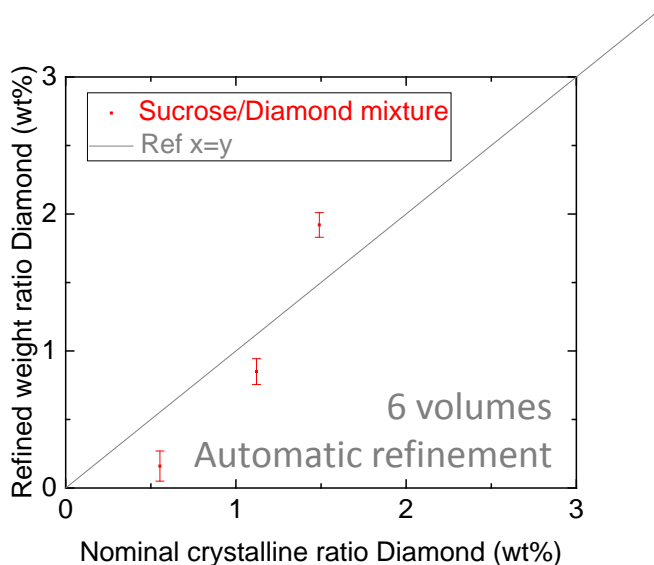
hBN
(5 – 20 wt%)

hBN modelled using
observed data and
calibrated versus Alumina

Strongly correlated Diamond/Sucrose reflections



Limits of the least square optimization



The three candidates can be used under favorable circumstances:

Alumina: restrained wt%

Diamond: if peak overlapping allows

hBN: if peak overlapping allows and intensity (not more than 20wt%)

- ❖ Validate DoC method
- ❖ Try DoC using the scaled amorphous phase
- ❖ Assess the lower limit of amorphous-QPA

- ❖ Ternary mixtures with each of the standards (increased Diamond wt%)
- ❖ Test of alternative analysis methods (PONKCS, 'Siroquant'-like, proper model of hBN distortions, 'Principal Component Analysis'-like)

My thanks go to:

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Claire Villevieille & Cyril Marino
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Danysz Eustache & coworkers
***Saint-Gobain, Saint-Gobain Advanced
Ceramics LLC***

Ian Madsen
CSIRO, Australia

Useful references

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Hyderabad – India

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Thank you for your attention
Questions?

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