THE DEVELOPMENT OF NANOMATERIALS AND AMORPHOUS REFERENCES AND THEIR USE IN PHARMACEUTICAL IDENTIFICATION AND CHARACTERIZATION

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International Centre for Diffraction Data, Newtown Square, PA, USA



This document was presented at PPXRD -Pharmaceutical Powder X-ray Diffraction Symposium

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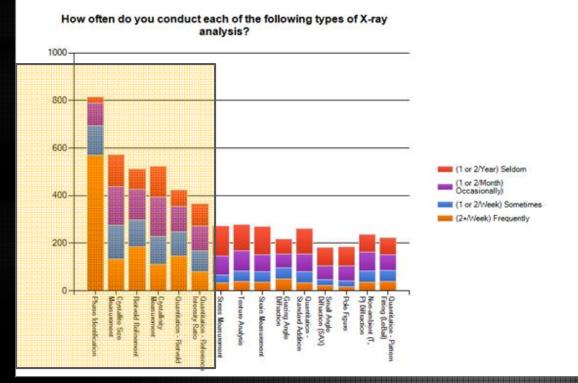
PPXRD Website – <u>www.icdd.com/ppxrd</u>

ICDD Website - www.icdd.com

Why develop nanomaterial or amorphous references

Most common analyses –global user surveys 2006, 2009, 2012

Material Identification Material Quantitation Rietveld RIR Crystallinity Crystallite Size



Most Frequent Downloads

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POWDER X-RAY DIFFRACTION DETECTION OF CRYSTALLINE PHASES IN AMORPHOUS PHARMACEUTICALS

B. A. Sarsfield,¹ M. Davidovich,¹ S. Desikan,¹ M. Fakes,¹ S. Futernik,¹ J. L. Hilden,¹ J. S. Tan,² S. Yin¹, G. Young¹, B. Vakkalagadda, and, K. Volk¹ ¹Bristol Myers Squibb Co., New Brunswick, NJ ²Purdue University, West Lafayette, IN

Advances in X-ray Analysis on-line publications

Basics of Amorphous and Amorphous Solid Dispersions

Ann Newman Seventh Street Development Group PO Box 526, Lafayette, IN 47902 765-650-4462 ann.newman@seventhstreetdev.com <u>www.seventhstreetdev.com</u>

Characterisation and prediction of stability of amorphous materials during pharmaceutical development: Pair-wise Distribution Function

Helen Blade, Steve Cosgrove, Jonathon Booth, Anne Kavanagh

PAR&D, AstraZeneca, Macclesfield, UK

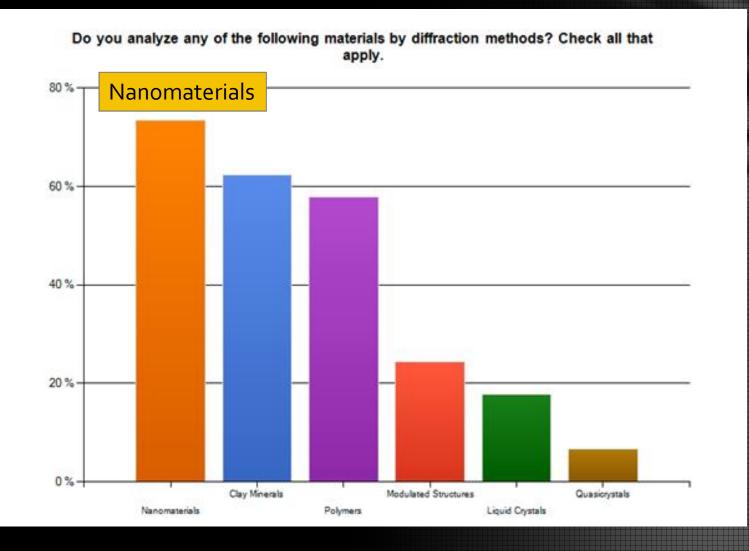
May 2009



PPXRD on-line presentations

PPXRD on-line presentations

PDF-4/Organics



Tools for Nanomaterial and Amorphous Analyses with PDF-4+ Databases

Experimental

Amorphous references

Nanomaterial references

Supporting Data

DSC/TGA

SEM/TEM

Pair Distribution Functions

Calculation

Digital Pattern Simulation

Use 3 basic Algorithms

Applied Instrument Functions

Crystallite Size Modeling

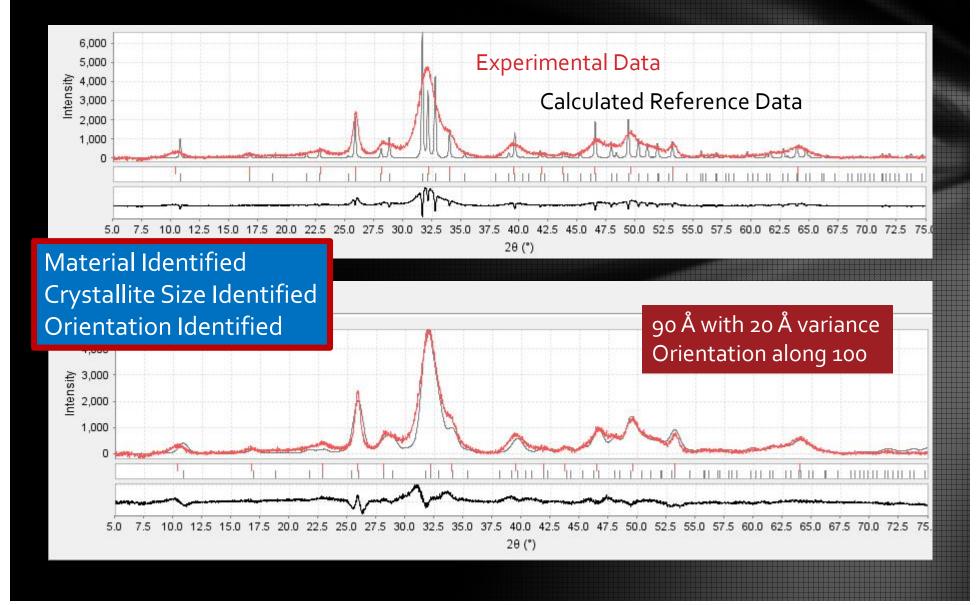
Molecular Orientation Functions

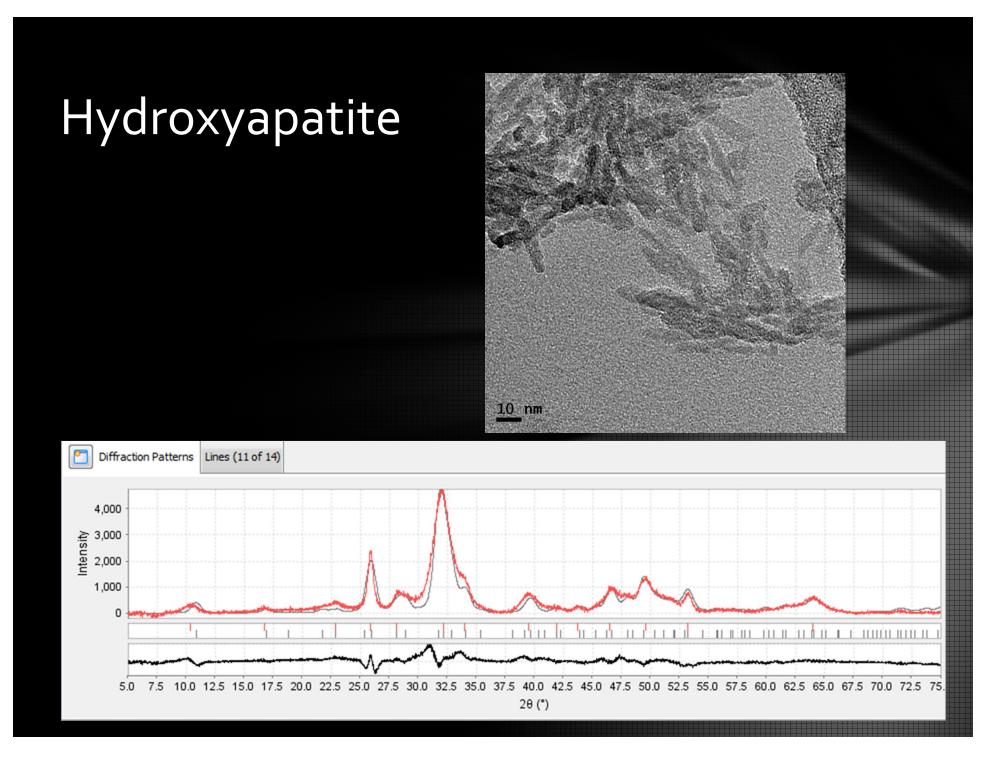
X-ray, neutron and electron diffraction scattering factors

Spot, ring, EBSD pattern simulation

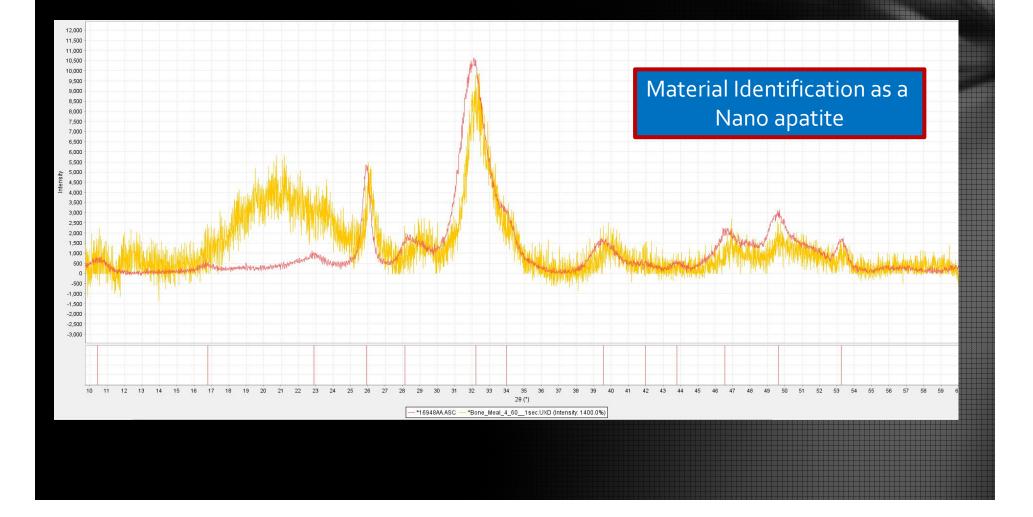
Similarity indices for identification of amorphous materials

Calculation Tools - Hydroxyapatite

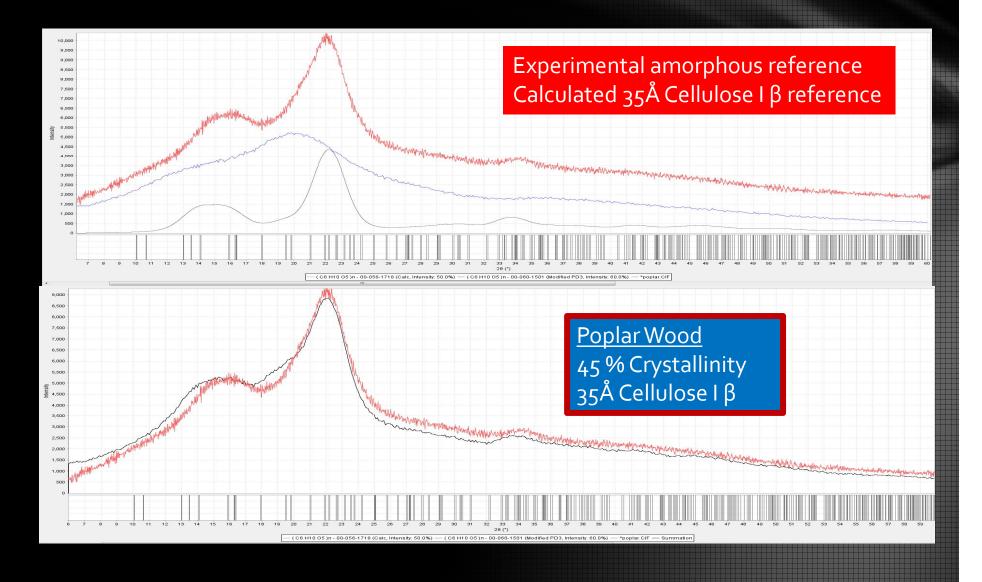




Experimental Tools -Nanoapatite Reference vs Commercial Bone Meal Fertilizer



Combine experimental and calculation tools – Poplar Wood



Reference materials for the study of polymorphism and crystallinity in cellulosics

T. G. Fawcett,^{1,a)} C. E. Crowder,¹ S. N. Kabekkodu,¹ F. Needham,¹ J. A. Kaduk,² T. N. Blanton,³ V. Petkov,⁴

E. Bucher,⁵ and R. Shpanchenko⁶ ¹International Centre for Diffraction Data, Newtown Square, Pennsylvania

²Illinois Institute of Technology, Naperville, Illinois

³Eastman Kodak Company, Rochester, New York

⁴Central Michigan University, Mt. Pleasant, Michigan

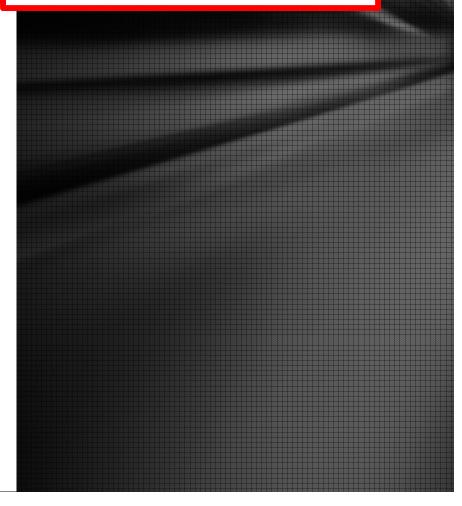
⁵International Paper Company, Loveland, Ohio

⁶Moscow State University, Moscow, Russia

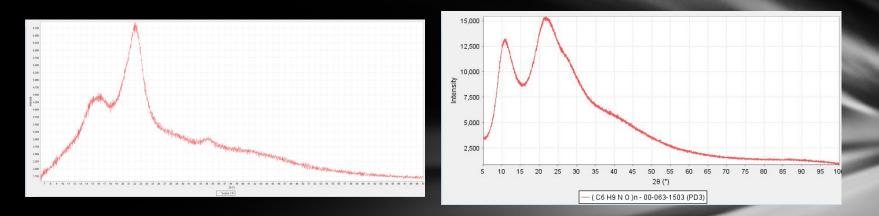
TABLE II. PDF cellulosic reference materials.

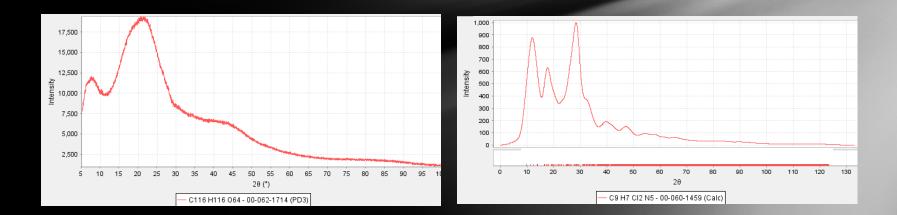
Material	PDF entry	Key attributes/source
Cellulose Ia	00-056-1719	Structural determination derived from fiber
Cellulose Ιβ	00-056-1718	Structural determination derived from filter paper
Cellulose II	00-056-1717	Structural determination derived from fiber
Amorphous cellulose	00-060-1501	Derived from cryogrinding multiple specimens
Microcrystalline cellulose	00-060-1502	Predominately cellulose I β , 40 Å Sigma Aldrich
Amorphous cellulose acetate	00-061-1408	Oriented amorphous
Cellulose acetate, CTA II	00-061-1407	Enhanced crystallinity
Cellulose acetate, CTA II	00-061-1409	Oriented film
Cellulose acetate	00-062-1713	USP grade
Cellulose acetate butyrate	00-062-1712	USP grade
Cellulose acetate pthalate	00-062-1714	USP grade
Methylcellulose	00-062-1290	Production grade
Methylcellulose	00-062-1291	Dehydrated production grade

Powder Diffr., Vol. 28, No. 1, March 2013



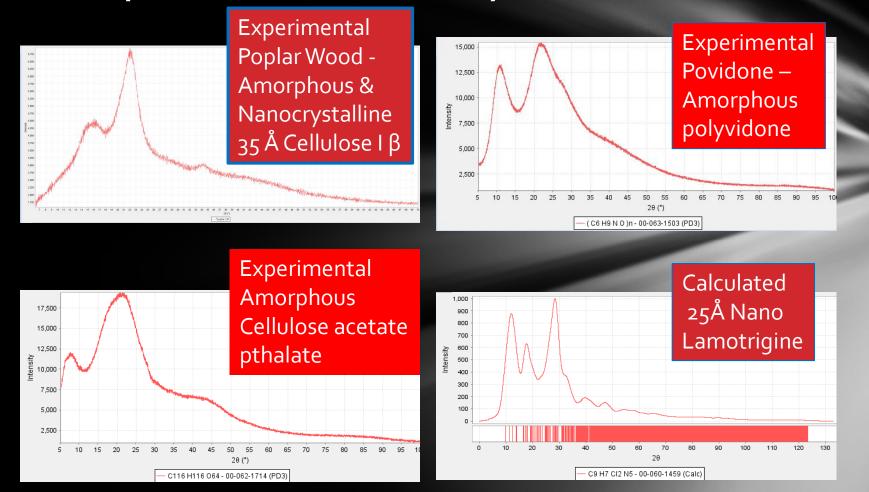
Amorphous or Nanocrystalline ? Incoherent scatter or small (50 Å <) crystallite scatter ?





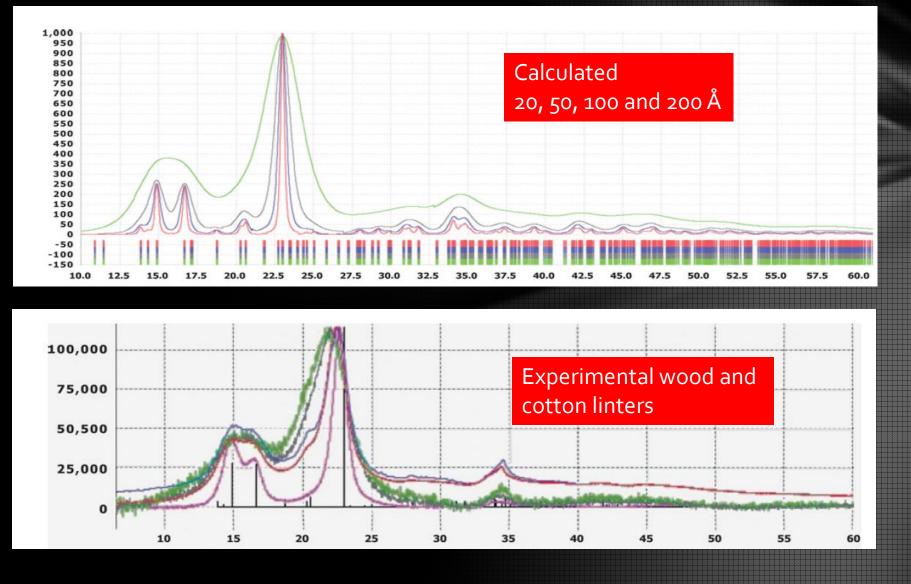
Where is the background ?

Amorphous or Nanocrystalline?



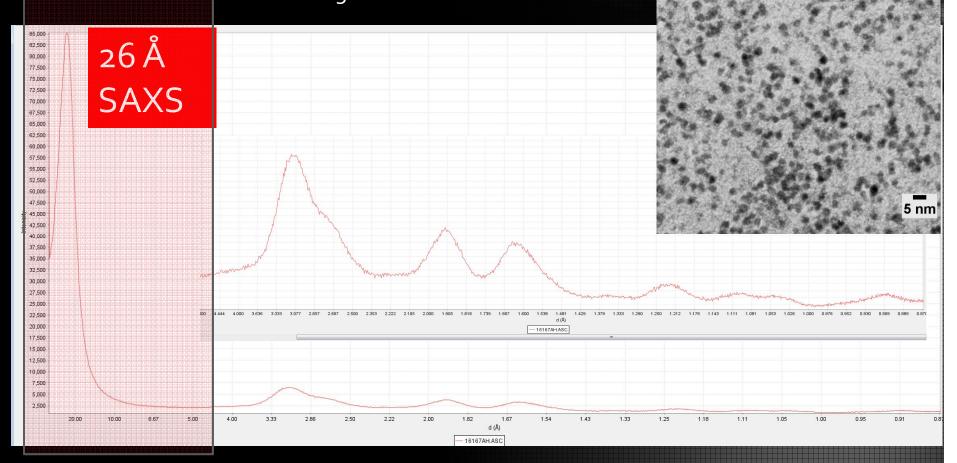
Where is the background ?

Cellulose Iß

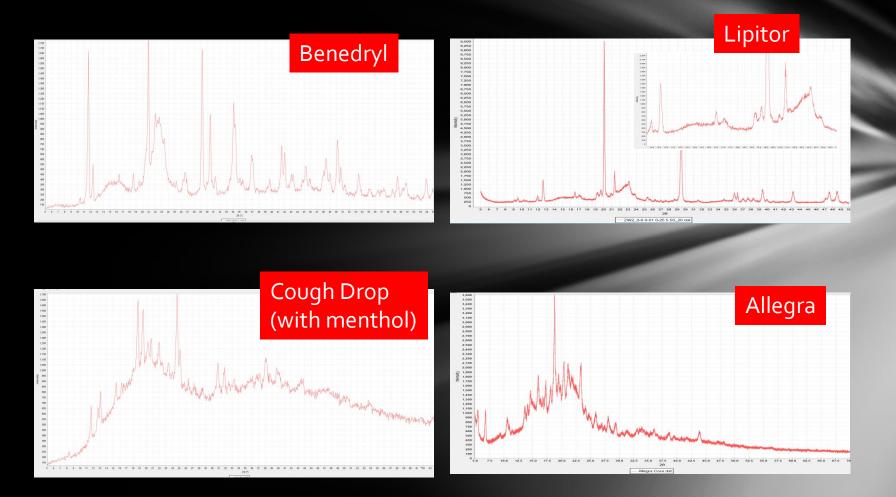


Nano Ceria

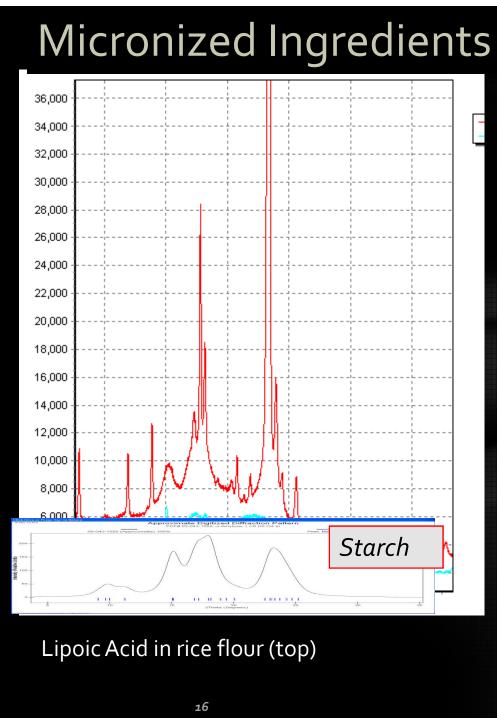
Nanomaterials would be expected to have SAXS at low angle from particle scattering

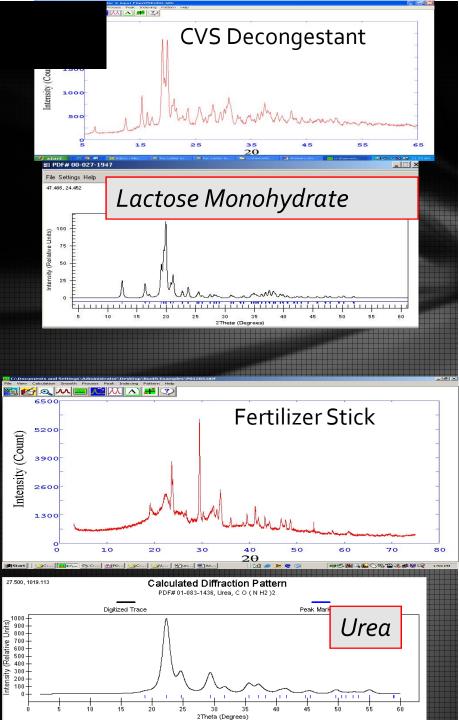


Pharmaceutical Formulations



Nanocrystalline and/or amorphous components are very common

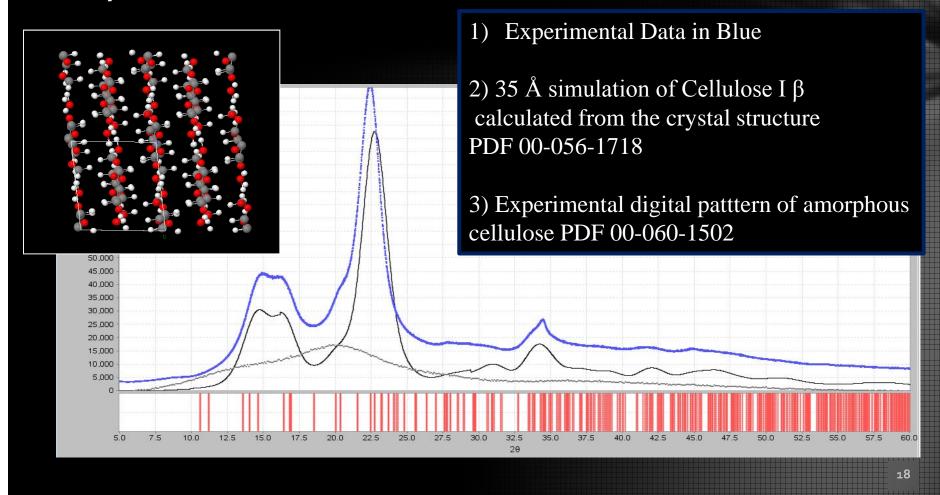




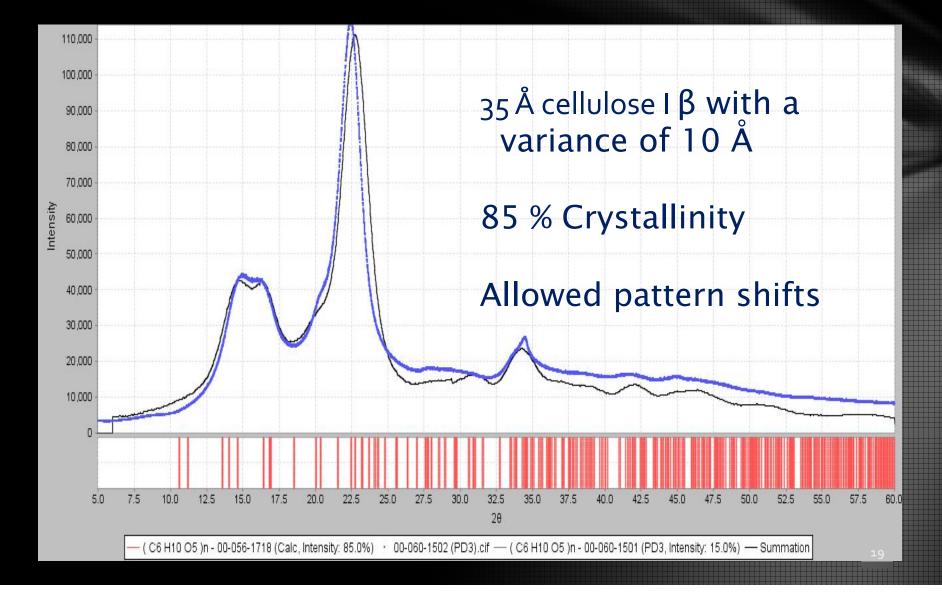
Amorphous or Nanocrystalline ? Incoherent scatter or small (50 Å <) crystallite coherent scatter ?

Use of supplemental analytical data

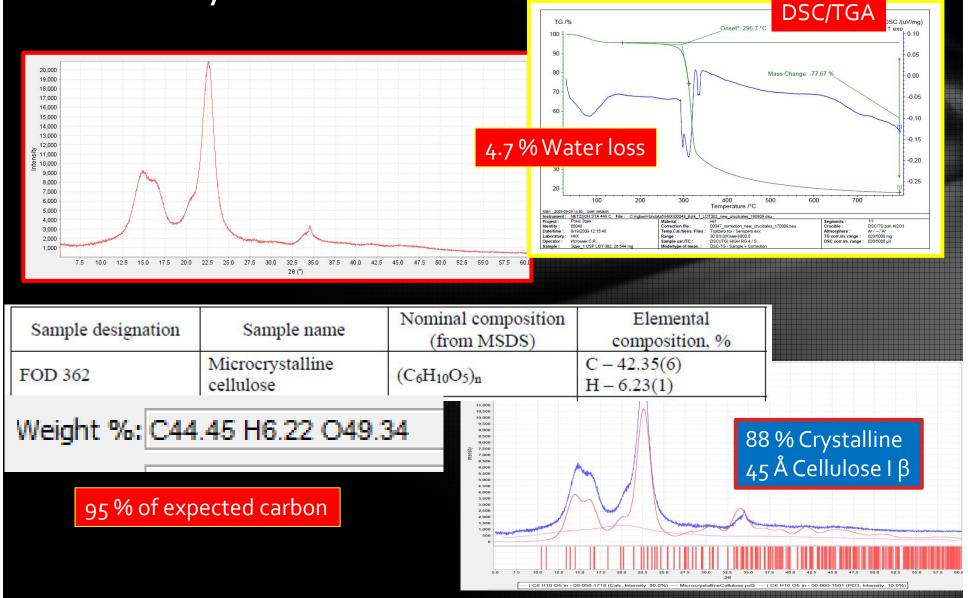
Microcrystalline Cellulose - Digital pattern simulation from the crystal structure

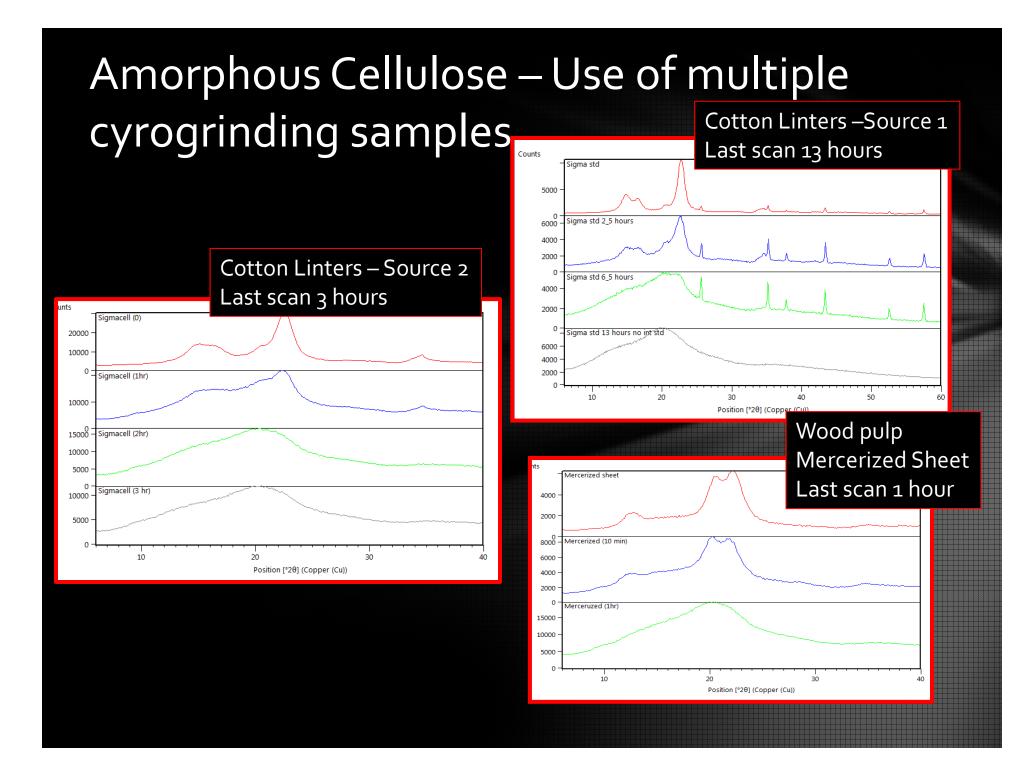


Sigma-Aldrich microcrystalline cellulose – crystallite size and crystallinity

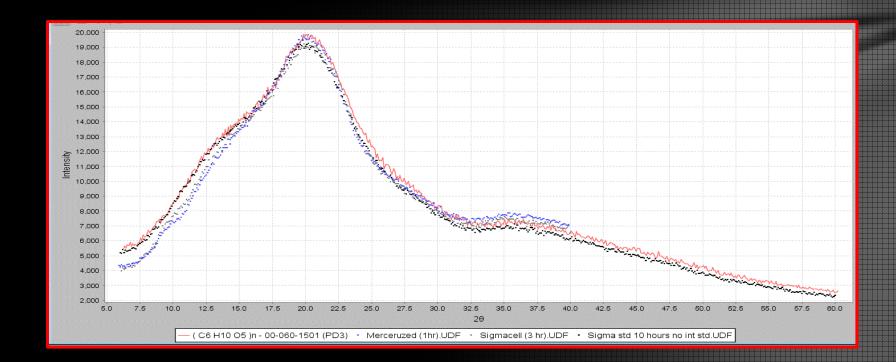


Supplemental Data – USP Microcrystalline Cellulose



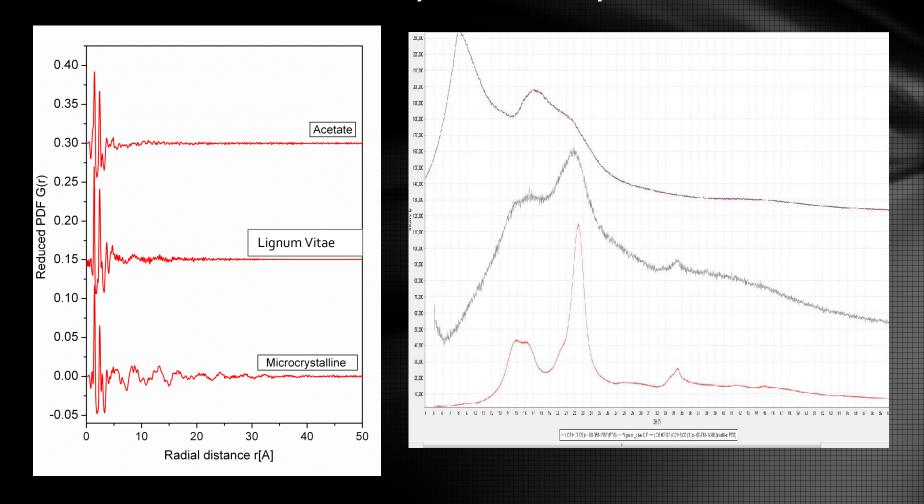


Amorphous cellulose – last data set of the 3 cryogrinding studies



Celluloses - Domain size and crystalliniity

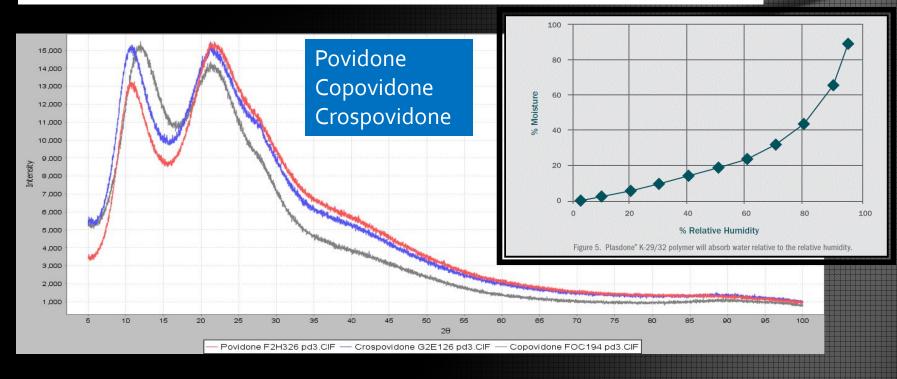
PDF



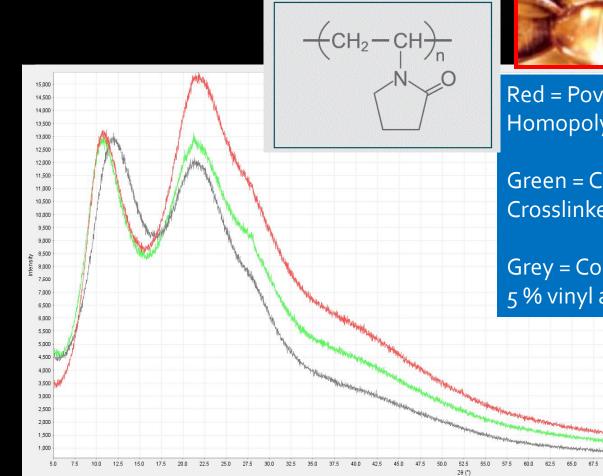
XRD

N-vinyl-2-pyrrolidone (Povidone) Water soluble polymers

Sample	Water absorbed	Decomposition T, °C	Comments (weight loss up to 800°C)
FOC194	-5.3% (30-200C)	276.3 387.2	2 step decomposition: total weight loss - 88.98%
G2E126	-9.51% (30-200C)	373.8	1 step decomposition weight loss - 85.15%
F2H326	-13.51% (30-200C)	377.2	1 step decomposition total weight loss - 80.34%



Povidone Hydration



— (C6 H9 N O)n - 00-063-1503 (PD3) — (C6 H9 N O)n (C4 H6 O2)n - 00-063-1504 (PD3, Intensity: 85.0%) — (C6 H9 N O)n - 00-063-1505 (PD3, Intensity: 85.0%)

Red = Povidone, most water, 13.5 % Homopolymer

Green = Crospovidone, 9.5 % Crosslinked homopolymer

Grey = Copovidone, least water, 5.3 % 5 % vinyl acetate

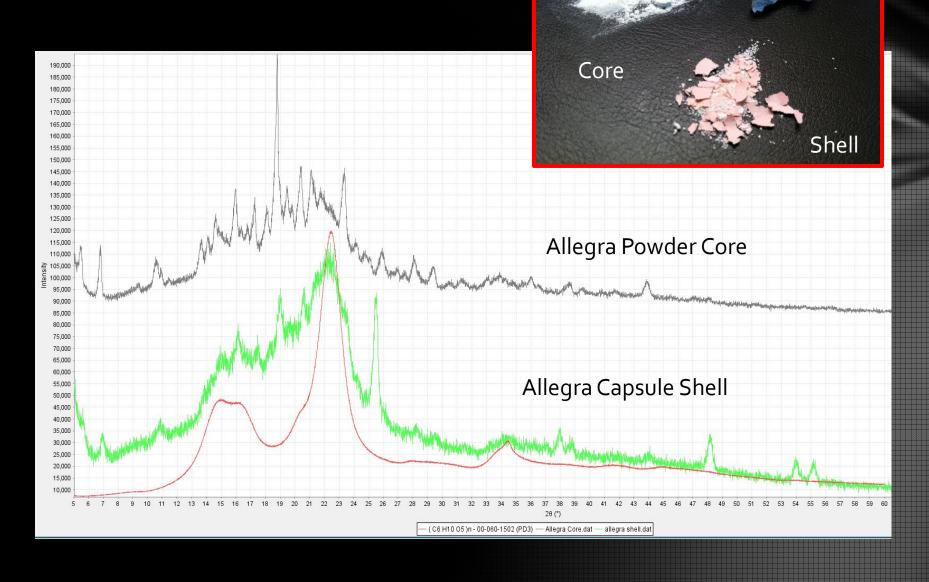
80.0 82.5

77.5

87.5 90.0 92.5 95.0 97.5

85.0

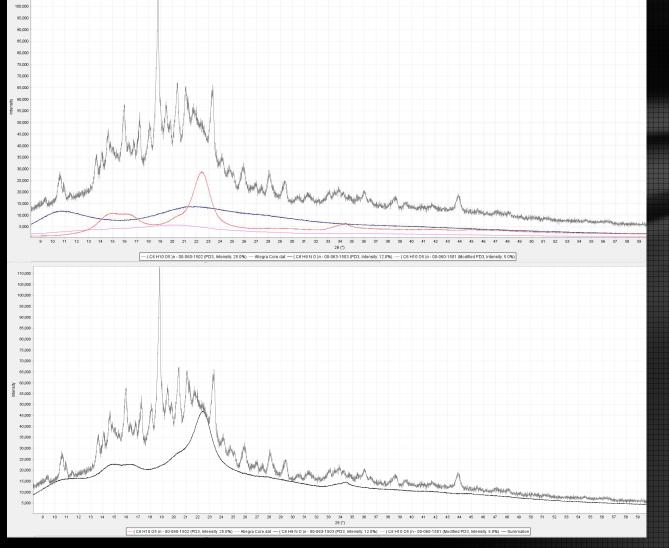
Allegra



Allegra Core

110,000

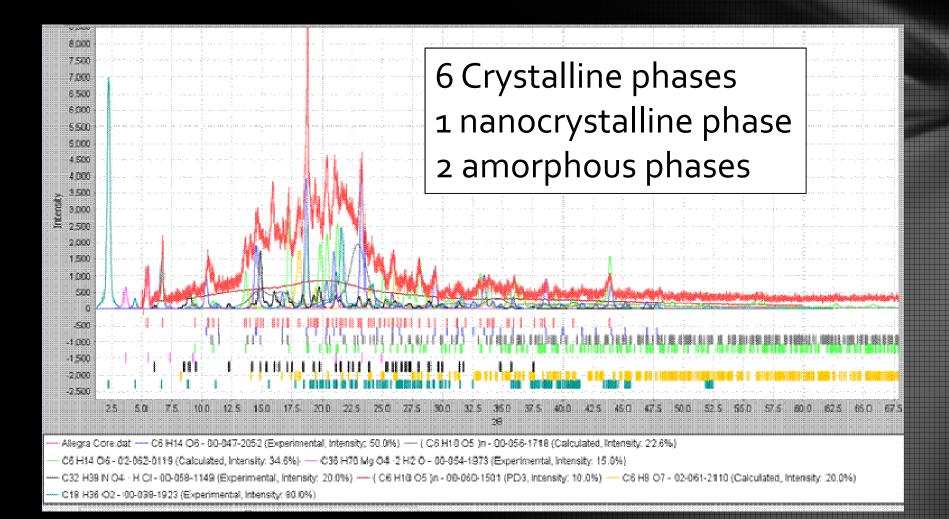
Material Identification Using Amorphous and Nano Crystalline references



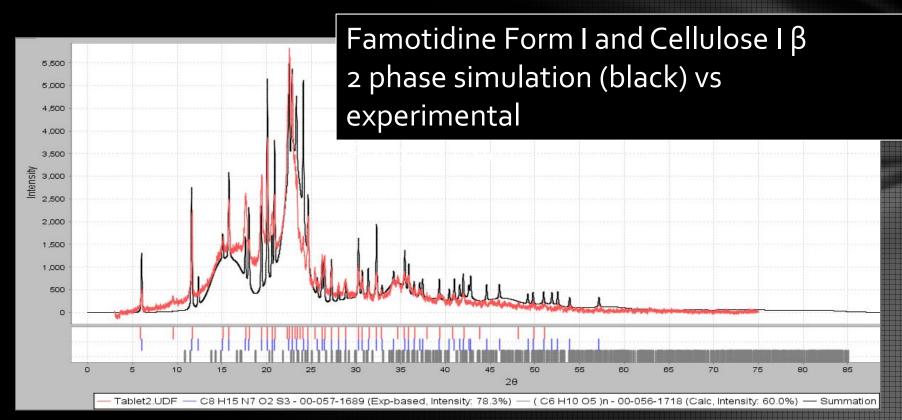
Microcrystalline Cellulose Amorphous Cellulose Amorphous Povidone

Summed Phases

9 phase Pharmacuetical Tablet - Allegra, Uses all Tools



Pepcid AC



Lipitor Using Pharmaceutical and Excipient subfile

📕 Ex d

3.02128

4.15670

6.99331

3.87905

3.94673

1.87431

2.29242

3.81276

.54396

2.28409

2.09512

.46714

.08405

1.86944

4.04027

L.91382

111

🐣 🗖 Ex I

41

15

12

11

🗌 P1 d

3.860000

1.875000

2.095000 8

1.913000

🛛 P1 I

44

🗌 P2 d

4.180000

7.060000

3.893000

4.520000

2.099000

1.867000

10

🗌 P2 I

🗌 P3 d

3.020400

4.181090

3.964080

3.824670

4.533160

4.055780

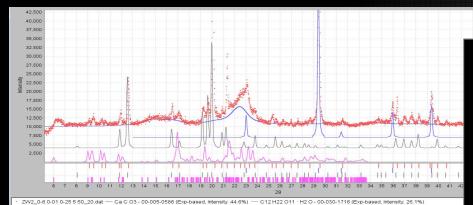
19

🗖 P3 I

Summed Phases 35,00 32,500 30.000 27.500 25.000 22,500 20,000 17,500 15.000 12,500 10,000 7,500 5,000

1 111 1 1 1 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 42 43 44 45 46 4 20 2W2_0-6 0-01 0-25 5 50_20.dat --- Ca C 03 - 00-005-0586 (Exp-based, Intensity: 44.6%) --- C12 H22 O11 - H2 O - 00-030-1716 (Exp-based, Intensity: 26.1%)

4 Individual Phases



Cellulose I β – nanocrystalline
CaCO3
Ca Atorvastatin Trihydrate
Lactose Monohydrate
(oriented 100)

Lipitor

Identification done by using the pharmacuetical and excipient subfile combination

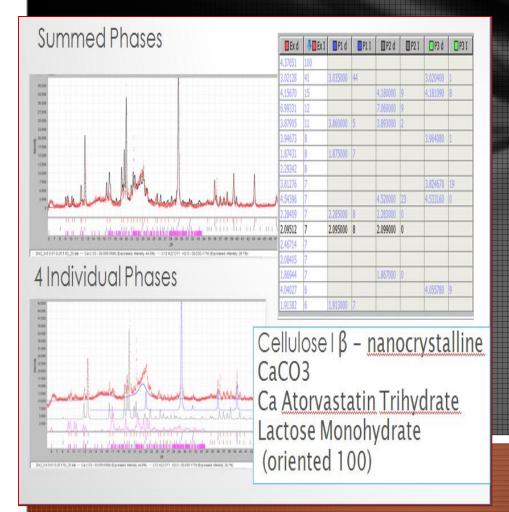
API – *Atorvastatin reference patterns come from donations and patents*

Microcrystalline cellulose identified by PD3 pattern PDF 00-060-1502

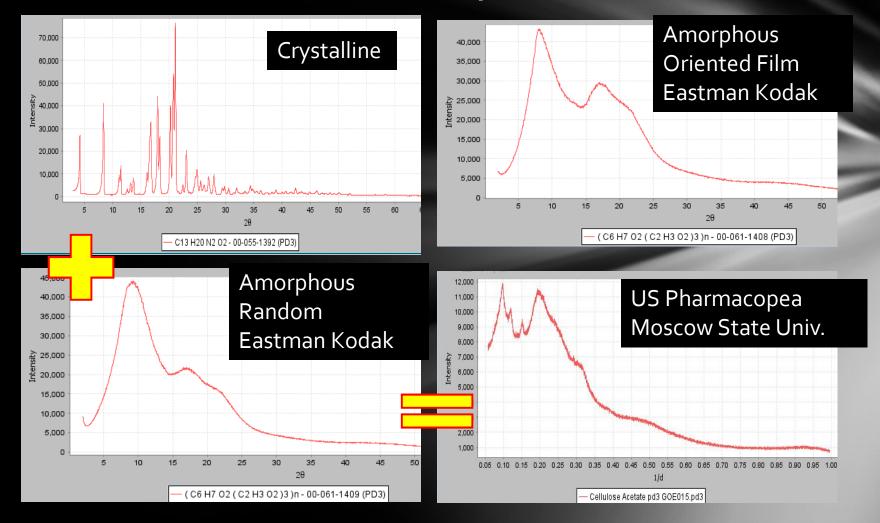
or

Use PDF 00-056-1718 Cellulose I β with a 35 Å crystallite simulation (Faber, Scardi, Leoni)

Lactose monohydrate oriented where we can identify the orientation (March-Dollase)

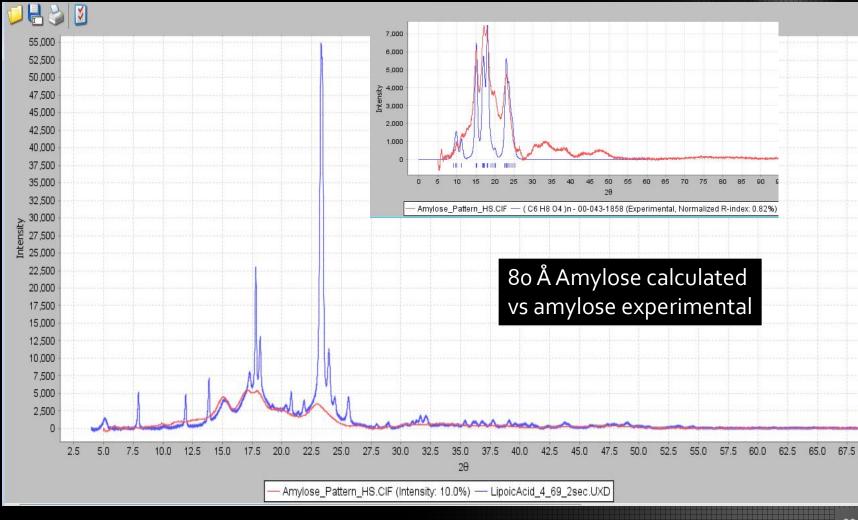


Cellulose Triacetate – Experimental Data

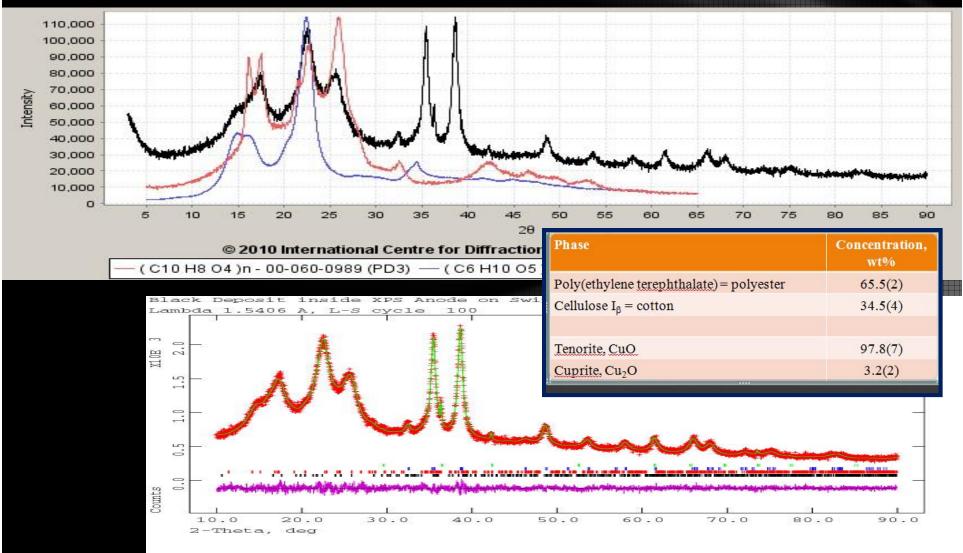


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Amylose with Lipoic Acid



Digital Patterns – Total Pattern Analysis – modified Rietveld refinement



THE DEVELOPMENT OF NANOMATERIALS AND AMORPHOUS REFERENCES

<u>Material</u>

Nano Microcrystalline Cellulose

Nano Apatitie

Substituted celluloses (Methyl, Acetate)

Amorphous Cellulose

N-vinyl-2-pyrrolidone (Povidone)

Polystyrenes, Polyethylene, Polypropylene, Polyvinylalcohol

Dextran

Supporting Data

TGA/DSC, Pair Dist. Function, Multiple samples, C, H, N analyses, SEM analysis

SEM analyses

NMR degree of substitution, Pair Dist. Function Multiple samples under varying process conditions

High purity standards, 3 cryogrinding studies

C, H, N analyses, TGA/DSC, USP references

Multiple data sets taken >40 years apart at different laboratories

C, H, N analyses, TGA/DSC

Amorphous and nanomaterials

"Good Quality Mark"

Marginal Quality Mark

– strong support data

PDF #	QM	Chemical Formula	Compound Name
00-060-1501	G	(C6 H10 O5)n	Cellulose, amorphous
00-062-1701	G	(C6 H7 O2 (O H)z · (C2 H3 O2)x	Cellulose acetate butyrate
00-062-1702	G	(C6 H7 O2 (O H)z · (C2 H3 O2)x	Cellulose acetate butyrate
00-062-1712	G	(C4 H8 O2)n · (C2 H4 O2)n	Cellulose acetate butyrate
00-062-1713	G	C164 H174 O111	Cellulose acetate
00-062-1714	G	C116 H116 O64	Cellulose acetate phthalate, amorph
00-063-1501	G	H (C6 H10 O5)n O H	Dextran-4, amorphous
00-063-1502	G	(C6 H10 O5)n	Dextran-250, amorphous
00-063-1503	G	(C6 H9 N O)n	Povidone, amorphous
00-063-1504	G	(C6 H9 N O)n (C4 H6 O2)n	Copovidone, amorphous
00-063-1505	G	(C6 H9 N O)n	Crospovidone, amorphous
00-063-1508	G		Pepsin A
00-063-1509	G	((C8 H14 O2)0.88 · (C2 H4 O)0	Poly (vinyl butyral), amorphous
00-063-1511	G	C15 H31 C O2 C30 H61	Beeswax

PDF #	QM	Chemical Formula	Compound Name
00-060-1506	М	(C16 H14 O3)n	Polycarbonate, amorphous
00-060-1507	М	(C8 H8)n	Polystyrene, atactic
00-060-1508	М	(C15 H17 N)n	Acrylonitrile butadiene styrene, amor
00-060-1509	М	(C10 H8 O4)n	Poly (ethylene terephthalate), amorp
00-061-1408	М	(C6 H7 O2 (C2 H3 O2)3)n	Cellulose triacetate
00-061-1412	М	(C14 H10 O4)n	α-Poly (ethylene-2,6-naphthalate)
00-062-1288	М	C2.22 H4.22 O0.22	Ethylene vinyl acetate
00-062-1289	М	C2.36 H4.36 O0.36	Ethylene vinyl acetate
00-062-1290	М	(C8.45 H14.9 O5)n	Methyl cellulose, amorphous
00-062-1703	М	(C6 H7 O2 (O H)z · (C2 H3 O2)x	Cellulose acetate butyrate
00-062-1704	М	(C6 H7 O2 (O H)z · (C2 H3 O2)x	Cellulose actetate proprionate
00-062-1705	М	(C4 H8)n	Poly(butene-1)
00-062-1706	М	(O C H2)n	Poly (oxymethylene)
00-062-1707	М	(O C5 H6 Cl4)n	Poly (3,3-bis(chloromethyl)oxetane)
00-062-1708	М	(C4 H8 O)n	Poly(tetrahydrofuran)
00-062-1709	М	(CH2CH((C6H4)CH3))n	Poly (o-vinyl toluene)
00-062-1710	М	(C H2 C Cl2)n	Poly (vinylidene chloride)
00-062-1711	М	(C H2 C H (C6 H5))n (C H2 C H	Poly(styrene-acrylic acid)
00-063-1506	М	(C6 H10 O5)n	Cellulose II, coated
00-063-1507	М	(C6 H10 O5)n	Cellulose II
00-063-1510	М	((C8 H14 O2)0.88 · (C2 H4 O)0	Poly (vinyl butyral), amorphous

Applications

Nanomaterials

Starch in Lipoic Acid

Cellulose in Lipitor, Allegra, Centrum Multivitamins, Pepcid AC, Benzepril, Benedryl, Kroger Decongestant, Promethazine, Donnatal, Effexor

Lactose Hydrate in CVS decongestant (note: many formulations have larger particle sizes of this phase)

Amorphous Materials

Amorphous cellulose in most pharmacueticals with nanocellulose

Povidone in Allegra

Conclusions

The Powder File uses experimental digital patterns and calculated digital patterns to study nanomaterials and amorphous materials

Using both calculation and experimental tools we can identify, quantitate, measure crystallinity and crystallite size.

To distinguish small crystallite (< 50 Å) nanomaterials from amorphous materials often requires additional supplementary information (crystal structures, PDF, SEM, DSC/TGA, NMR etc). This information is being added to the database and used to determine quality.

References and Acknowledgements

Petkov, V., Ren, Y., Kabekkodu, S., Murphy, D., **(2013)**, "Atomic Pair Distribution Functions analysis on low-Z materials of limited degree of structural coherence", an invited article to Physical Chemistry Chemical Physics, *accepted for publication 2013*.

T. G. Fawcett, C.E. Crowder, S. N. Kabekkodu, F. Needham, J. A. Kaduk, T. N. Blanton, V. Petkov, E. Bucher and R. Shpanchenko (2013), "Reference Materials for the Study of Polymorphism and Crystallinity in Cellulosics", *Powder Diffraction*, 28(1), pp 18-32.

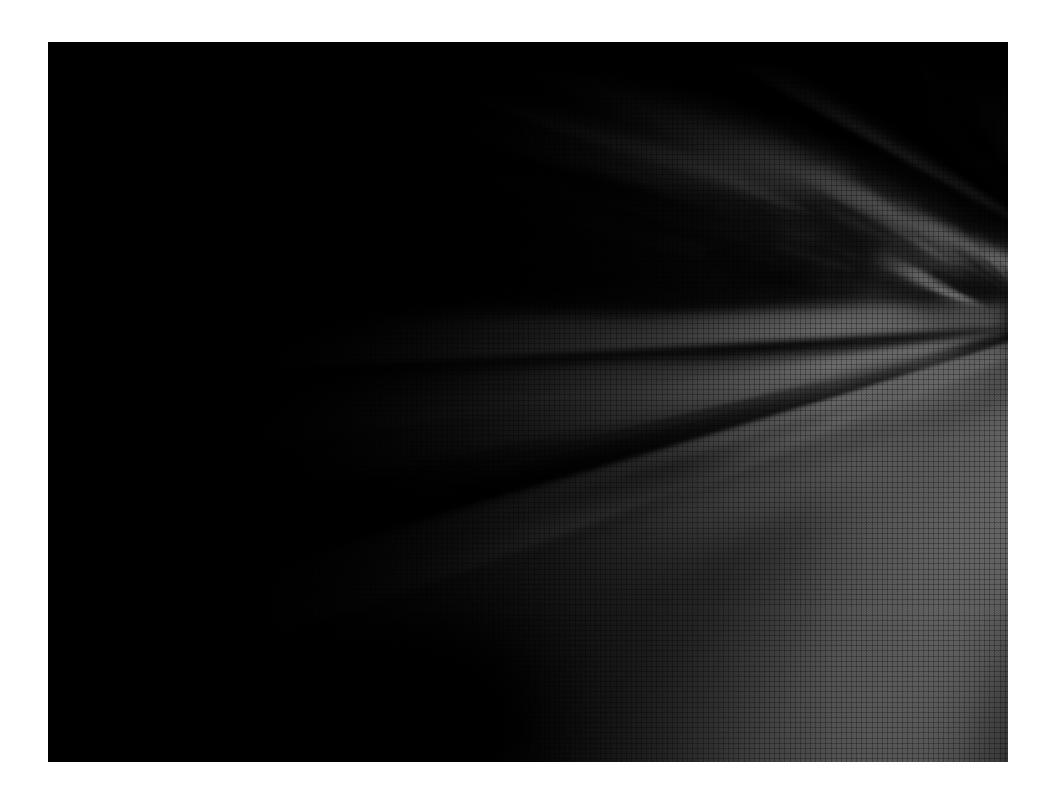
Jim Kaduk, Poly Crystallography Inc., – Amylose, Rietveld refinement

Tom Blanton, Eastman Kodak Co. – SAXS and XRD of nano-Ceria and nano Apatite from Cerion Corp, NMR data on substituted celluloses

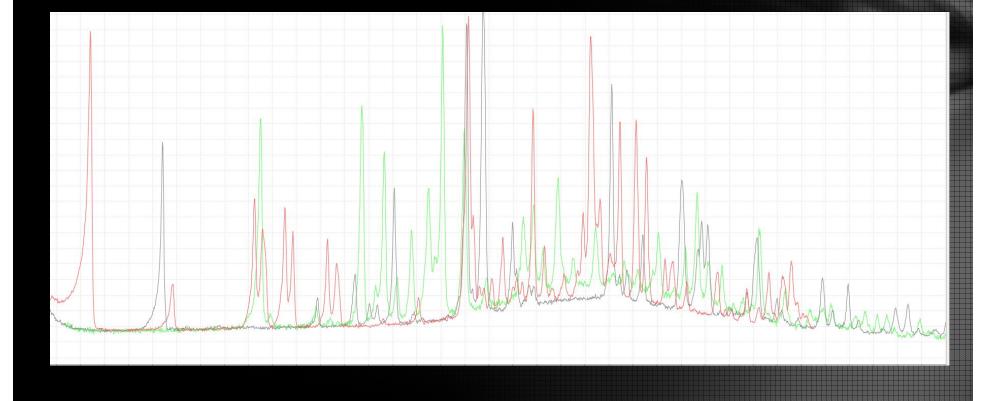
Ewa Bucher, International Paper – Cryogrinding studies of cellulose

Cam Hubbard (retired), Oak Ridge National Laboratory – pharmaceutical samples and data

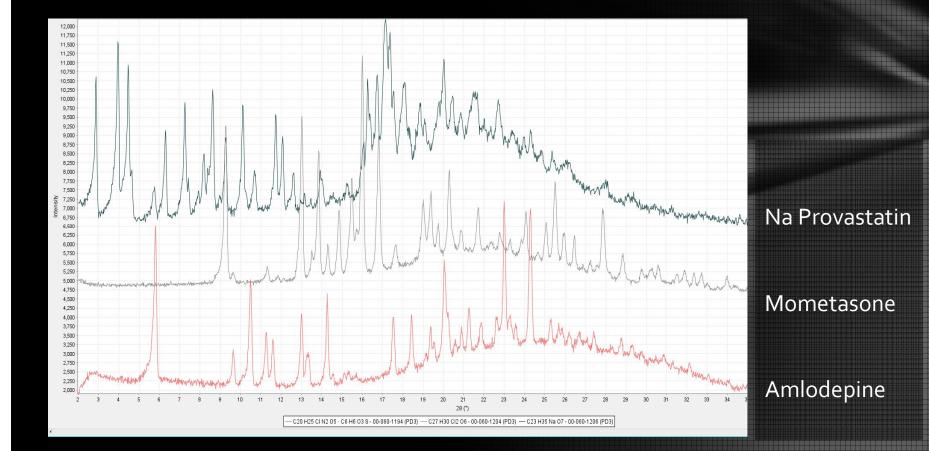
Offices of Nicholas Giuliani, MD – pharmaceutical samples



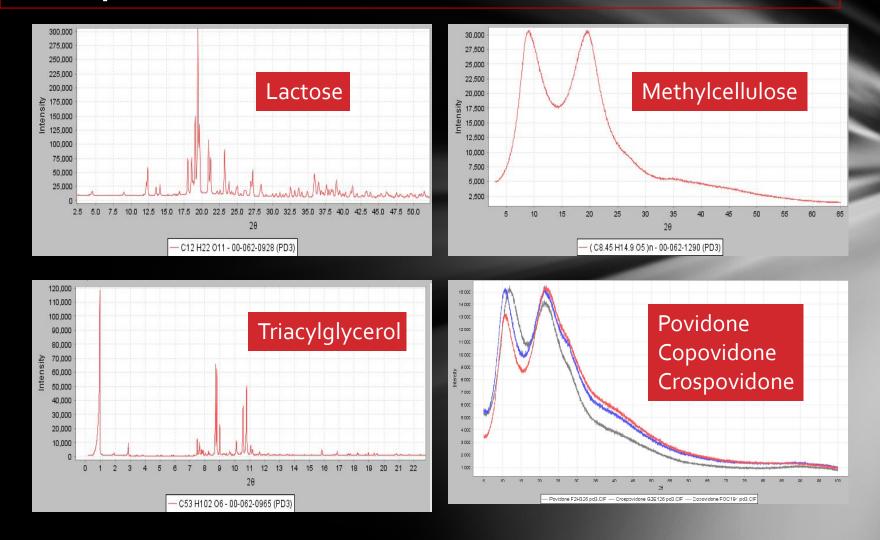
Capillary – Martin Vickers The capillary is amorphous not the drug



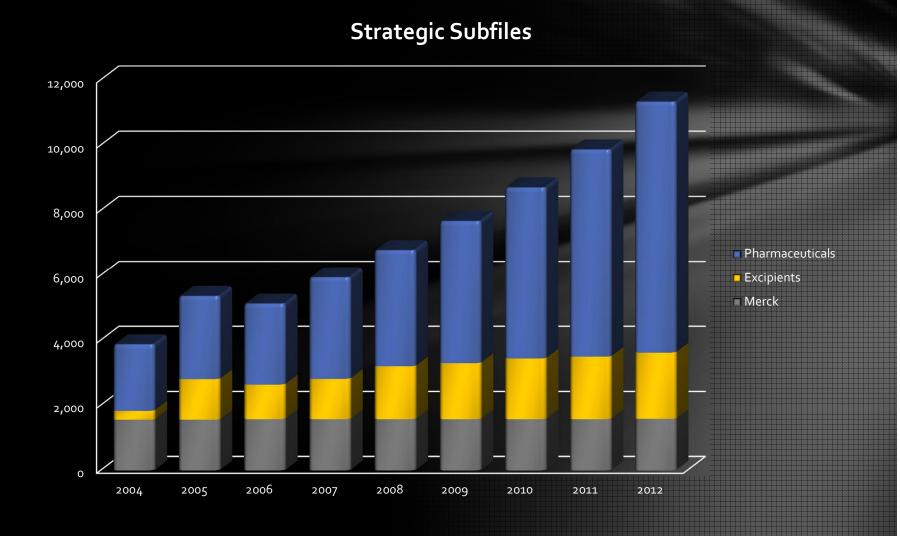
Mixed Crystallinity Pharmaceuticals



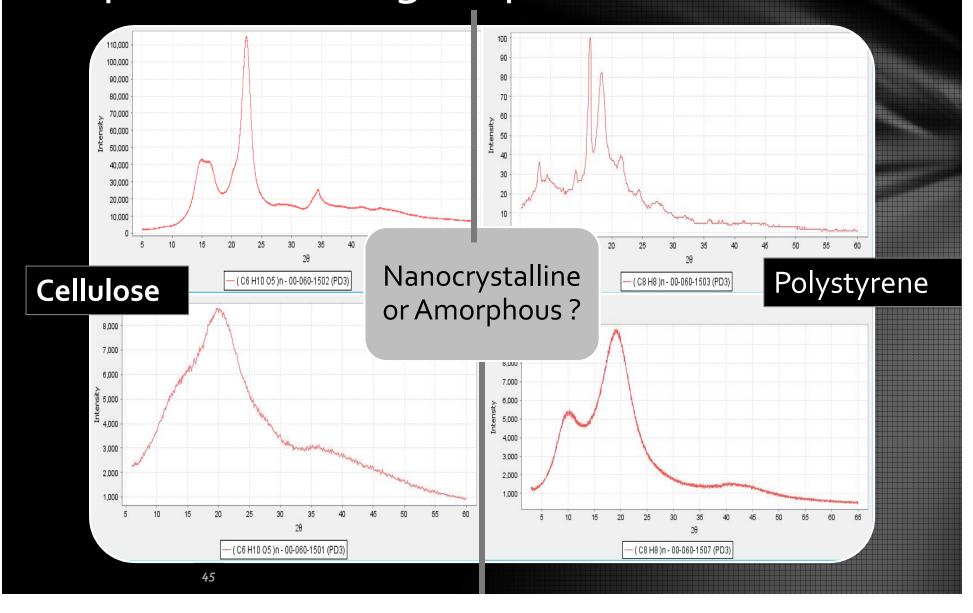
Excipients



Development of Strategic Subfiles



Experimental digital patterns



Polymeric excipients

