

# BEST PRACTICES IN FORMULATION ANALYSES BY X-RAY DIFFRACTION



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# This document was presented at PPXRD - Pharmaceutical Powder X-ray Diffraction Symposium

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PPXRD Website – [www.icdd.com/ppxrd](http://www.icdd.com/ppxrd)

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# About the Authors

- Suri Kabekkodu – editor in chief, pharmaceutical and excipients subfiles
- Stacy Gates – senior editor, polymers, non-crystalline materials
- Amy Gindhart – senior editor, pharmaceuticals
- Justin Blanton – senior programmer, JAVA, data mining, pattern simulations
- Cyrus Crowder – principle scientist, embedded software and method developments



# Determining Best Practices – Key References

- Fawcett, T. G., McDonnell, H., Higginson, R., (2006, 2009, 2012) “ICDD Global Customer Survey”, private communication, International Centre for Diffraction Data. Three separate surveys taken three years apart with over **2,000 responses** from scientists working in the field of diffraction and material analysis.
- Fawcett, T. G., Needham, F., Faber, J., Crowder, C. E., (2010), “International Centre for Diffraction Data round robin on quantitative Rietveld phase analysis of pharmaceuticals”, *Powder Diffraction*, 25(1), pp 60-67.
- Fawcett, T. G., Hubbard, C. R., (2004), “Formulation Analyses of Off-the-shelf Pharmaceuticals”, *American Pharmaceutical Review*, Vol. 7, pp 80-83.
- Fawcett T. G., Faber, J., Needham F., Kabekkodu S. N., Hubbard C. R. and Kaduk J. A. , (2006), “Developments in formulation analyses by powder diffraction analysis”, *Powder Diffraction*, 21, pp 105-110.





# About the data

- Y. Ososkov and Z. Cherbanyk Beti, Exova, Mississauga, Canada
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- H. Brusova, Zentiva AS, Prague, Czech Republic
  
- C. Hubbard, Oak Ridge National Laboratory, TN, USA
- N. Giuliani, Riddle Primary Care Associates (Riddle Hospital), Glen Mills, PA, USA
- J. Kaduk, Illinois Institute of Technology, Il, USA

# Outline

## Specimen Preparation- Round Robin, Experience

- General issues – low scattering power, absorption difference in ingredients, crystallinity
- Asymmetric Crystals
- Particle size
- Granularity
- Orientation
- Separation Techniques

## Methods of Analysis – Customers Survey Data

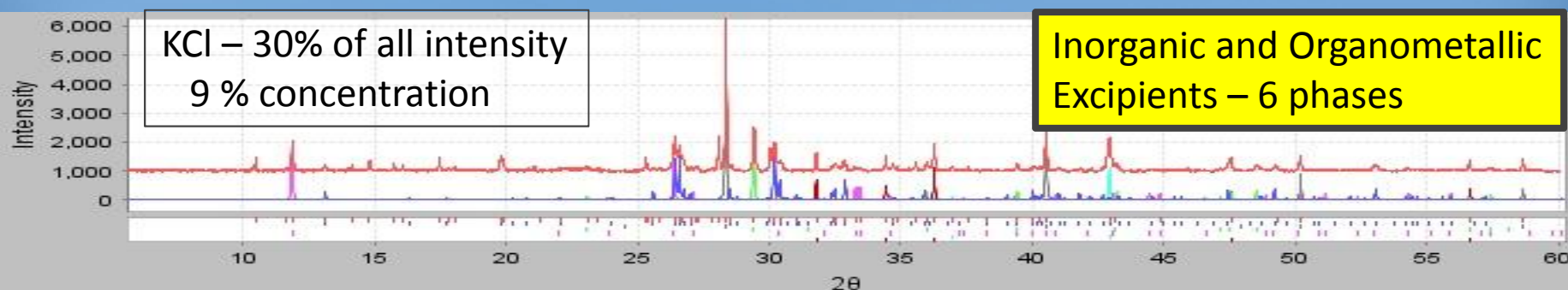
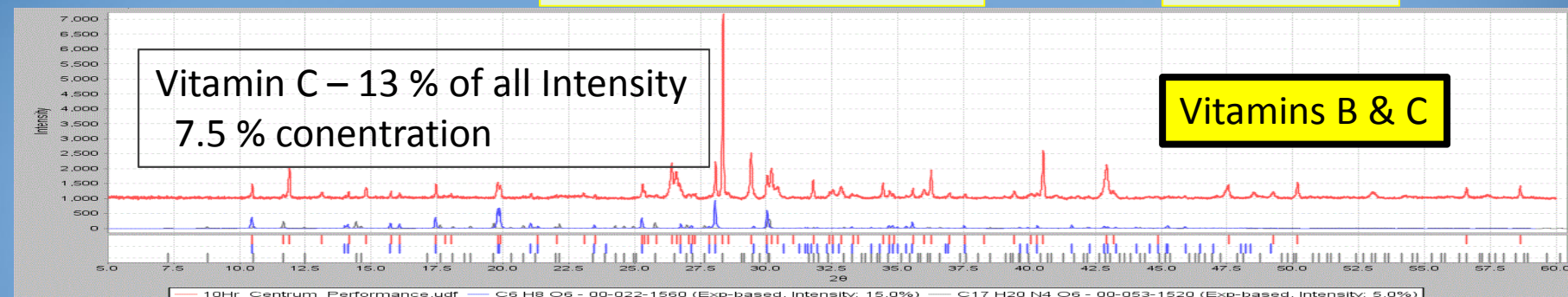
- Experience
- Databases
- Noncrystalline Materials
- Reference Quality
- Elemental Filters
- Subfiles
- Pattern Simulations
- Rietveld Analysis
- Combined Techniques – DXC/XRD
- Complimentary Analytical Techniques

# General Issues

$$I_{(hkl)\alpha} = \left[ \frac{I_0 \lambda^3}{64 \pi r} \left( \frac{e^2}{m_e c^2} \right)^2 \right] \left[ \frac{M_{hkl}}{V_\alpha^2} F_{(hkl)\alpha}^2 \left( \frac{1 + \cos^2 2\theta \cos^2 2\theta_m}{\sin^2 \theta \cos \theta} \right) \right] \left[ \frac{X_\alpha}{\rho_\alpha (\mu/\rho)_s} \right]$$

Low scattering factor (Z)

Low density



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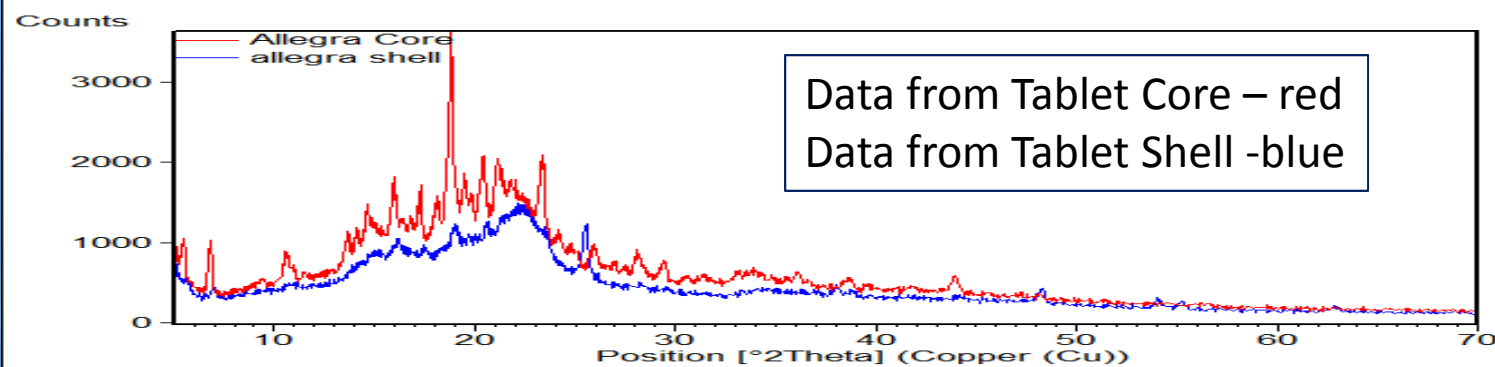
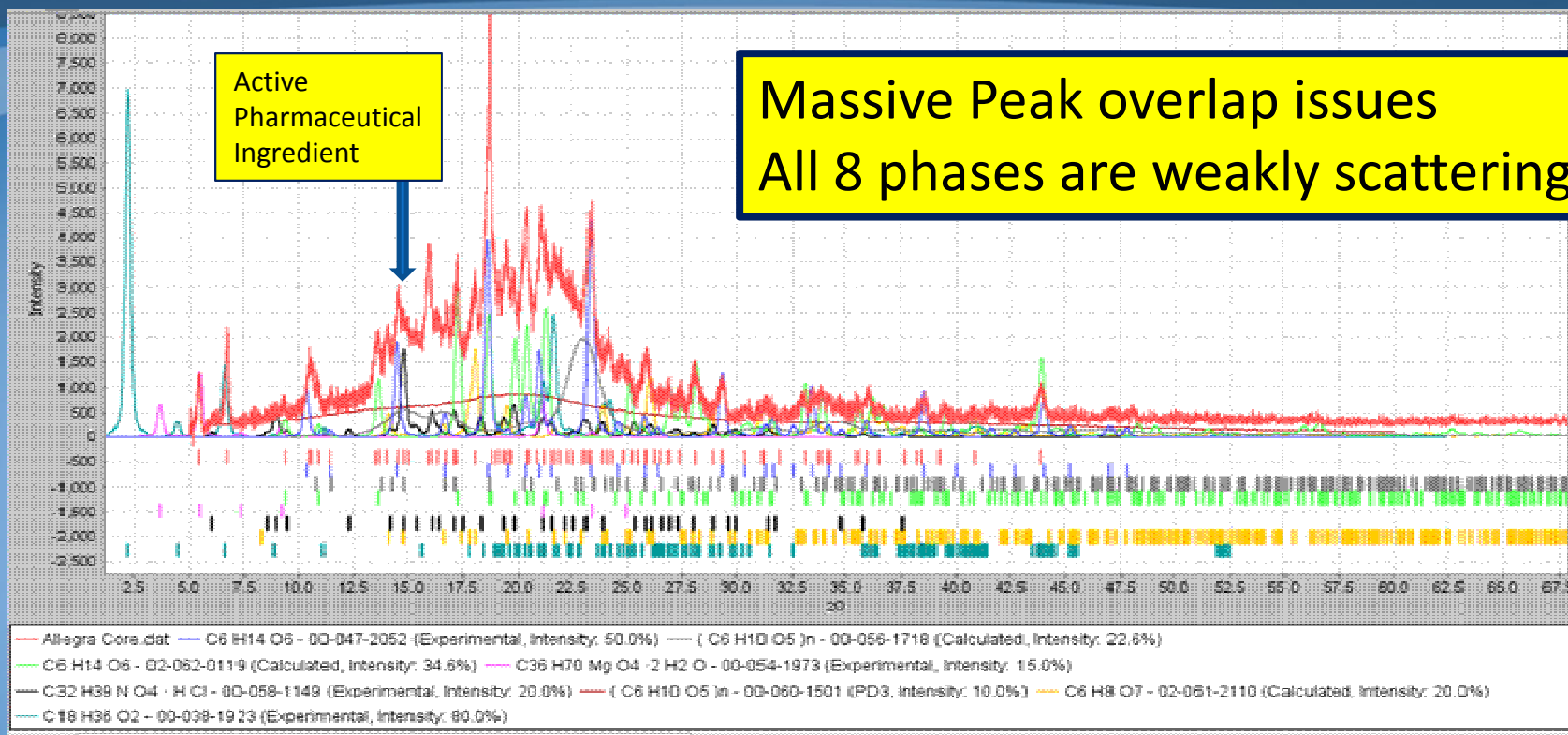
# Consequences

In Formulation analyses the API is often a low scattering low density material that is masked by highly scattering higher density materials often used as excipients

Identify excipients first and then identify the API

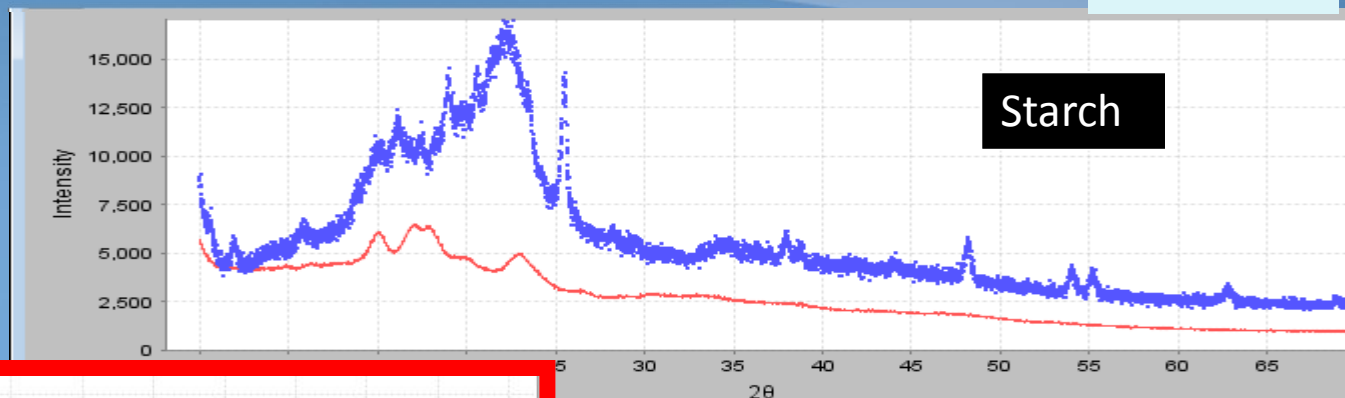


# General Issue - Crystallinity

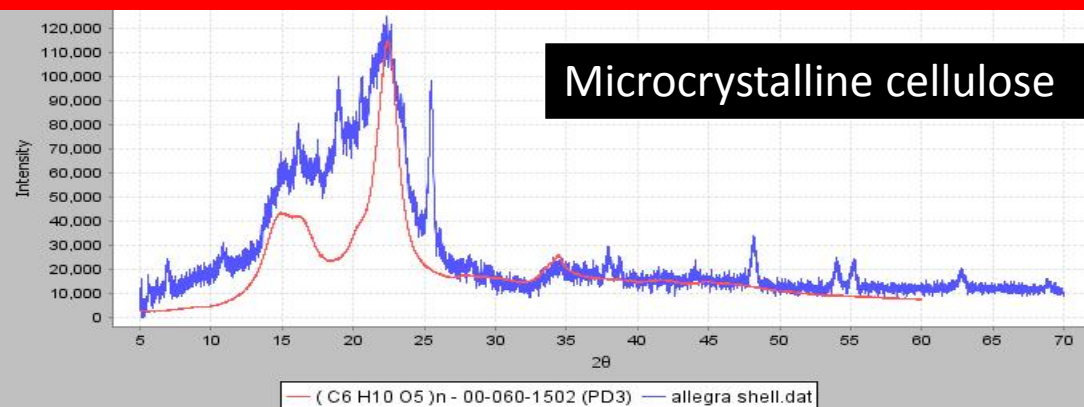


# Experimental digital patterns for Nanocrystalline and non-crystalline materials

Allegra

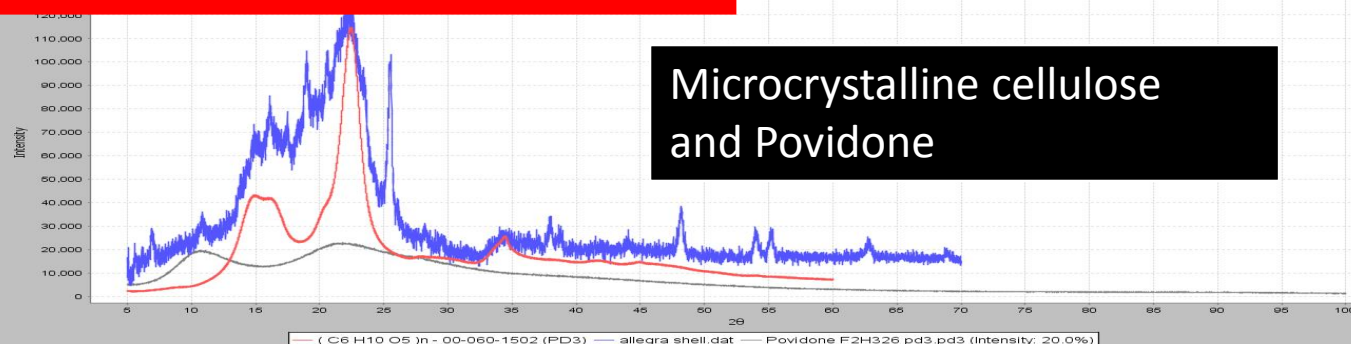


Starch



Microcrystalline cellulose

Microcrystalline cellulose  
Gelatinized Starch  
Povidone

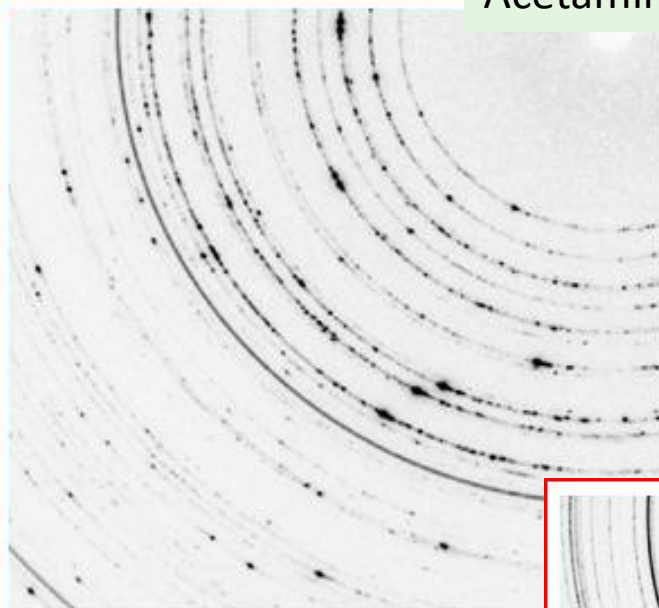


Microcrystalline cellulose  
and Povidone

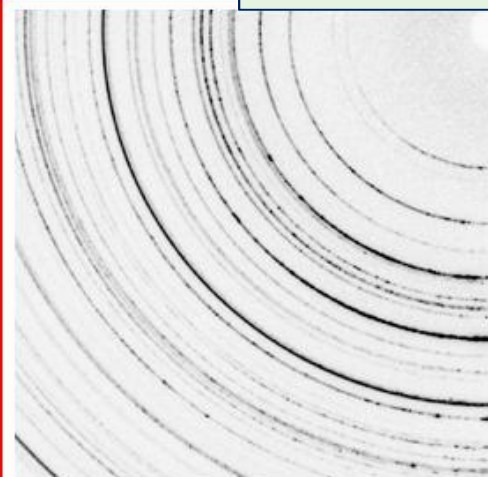
# Assymmetric Crystals

*ICDD Round Robin Data*

Acetaminophen & Si

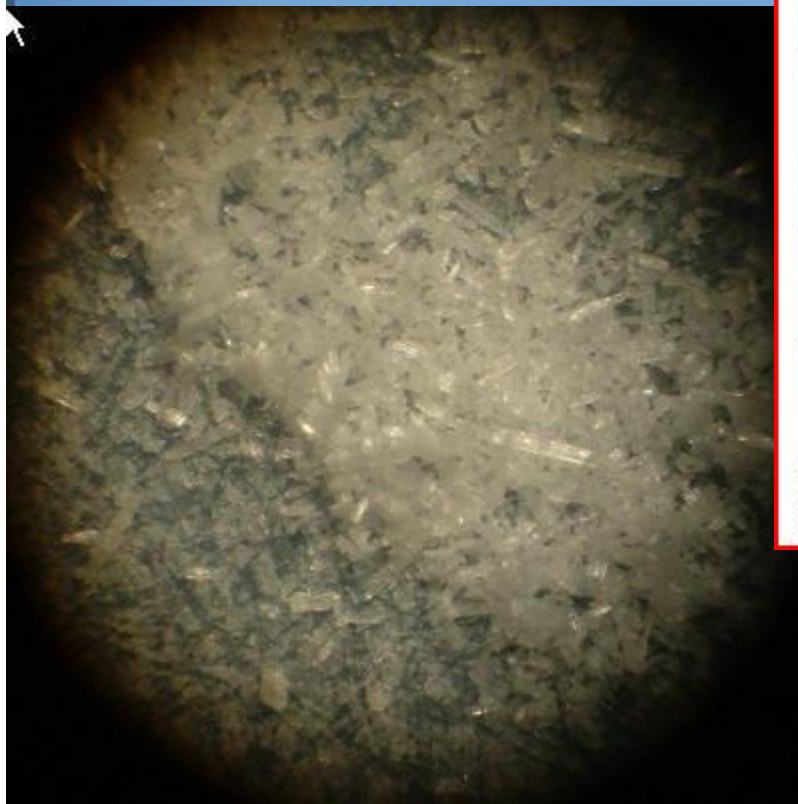


Mannitol & Si

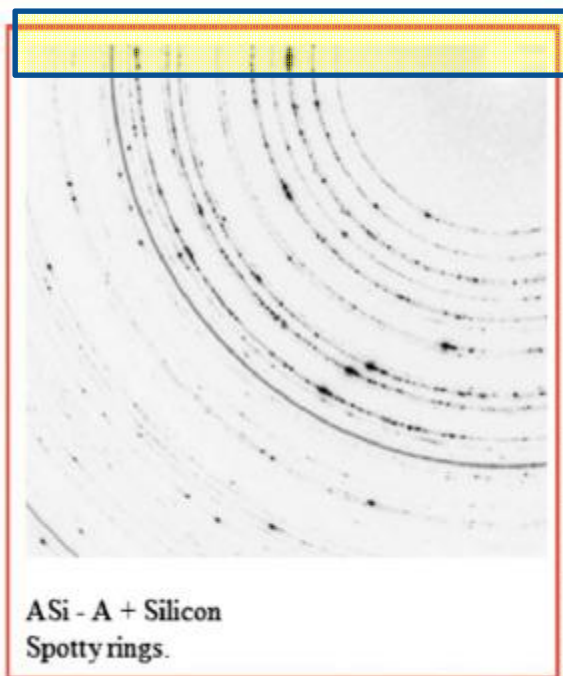
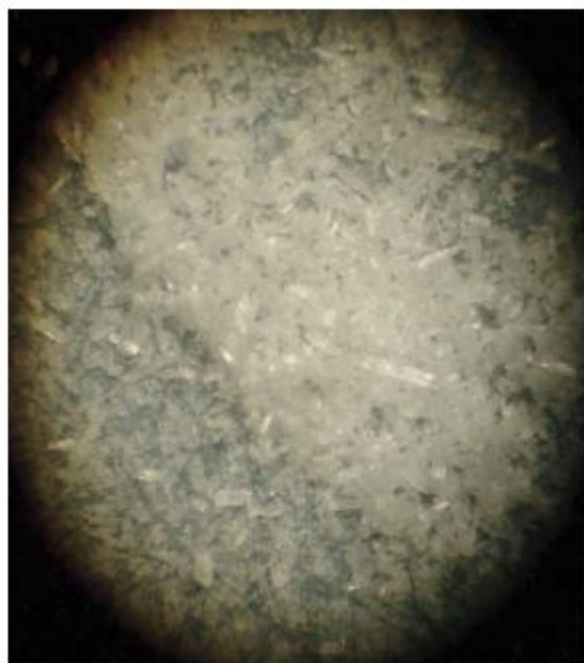


ASi - A + Silicon  
Spotty rings.

MSi - M + Silicon  
Strong texture and some spottiness.



# Large grains – capillary mount



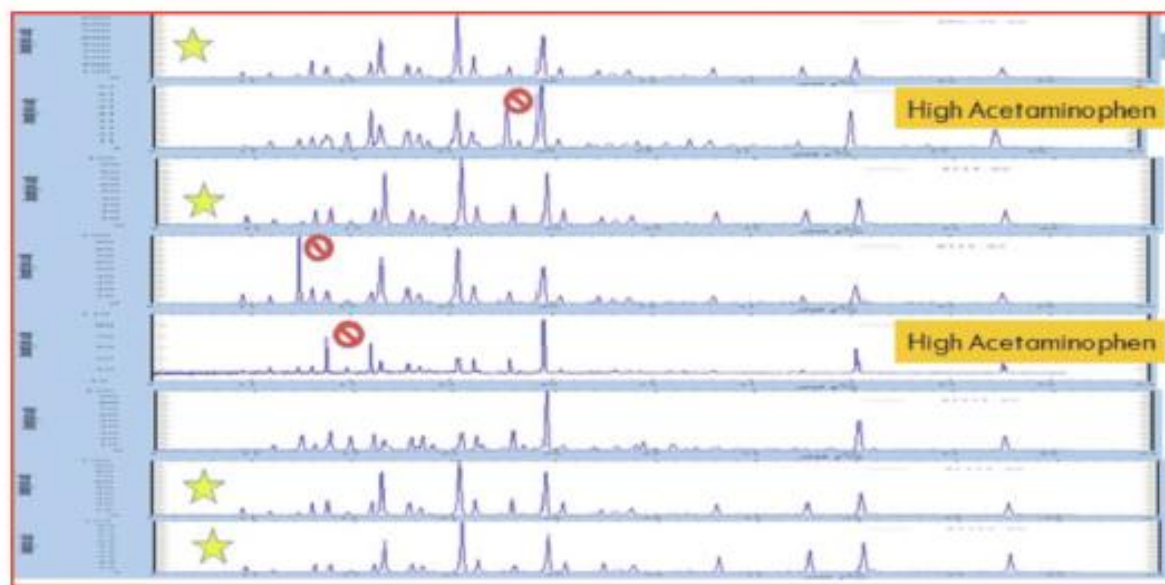
2D Detector



1D Detector

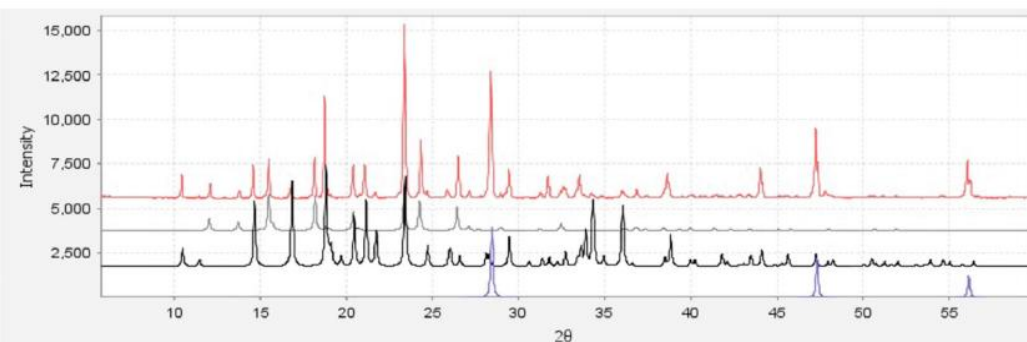


# Particle Size/Granularity



Only 4/8 Laboratories produced a randomly oriented powder – only 3/11 produced the correct quantitative results.

Grind the sample to eliminate the particle size effect in the original material



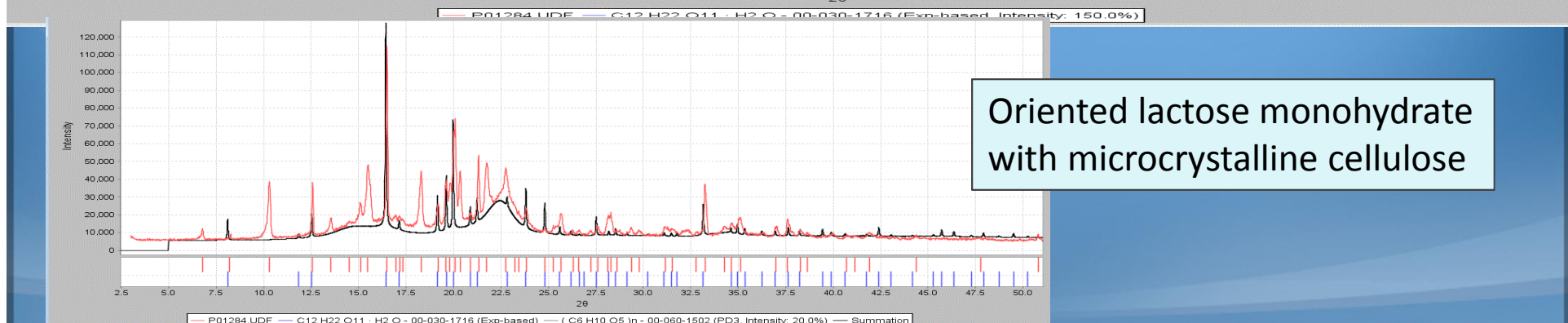
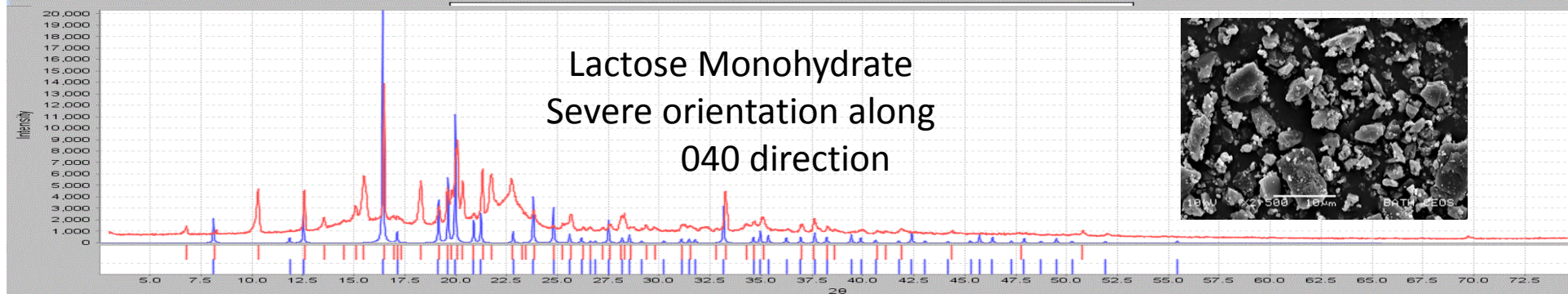
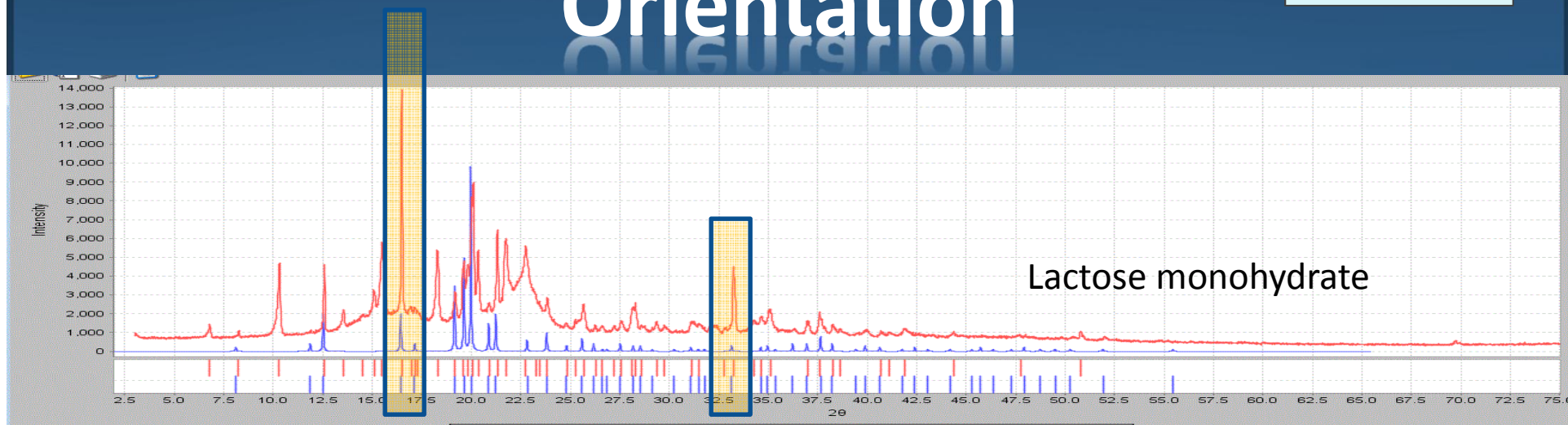
nilar  
and

Acetaminophen, D-mannitol  
and NIST silicon



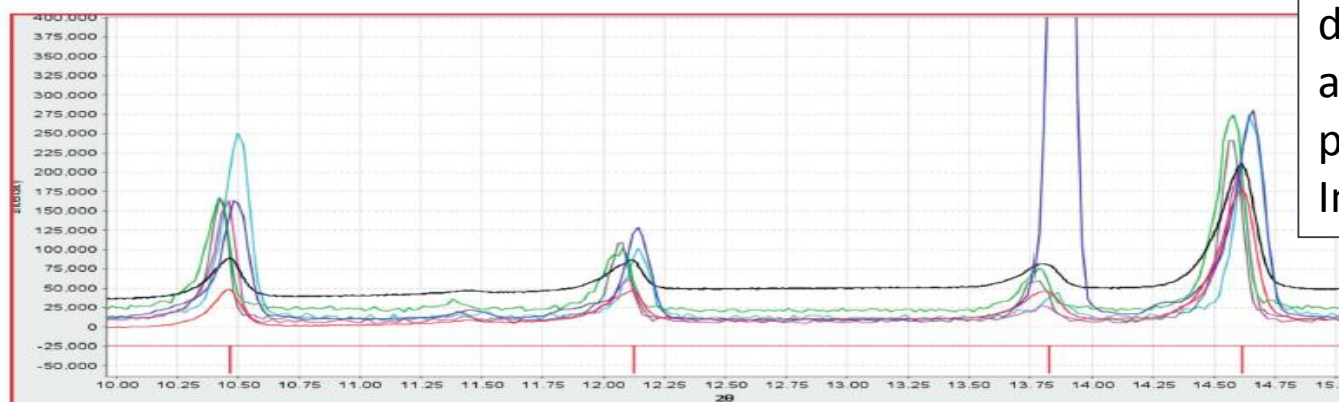
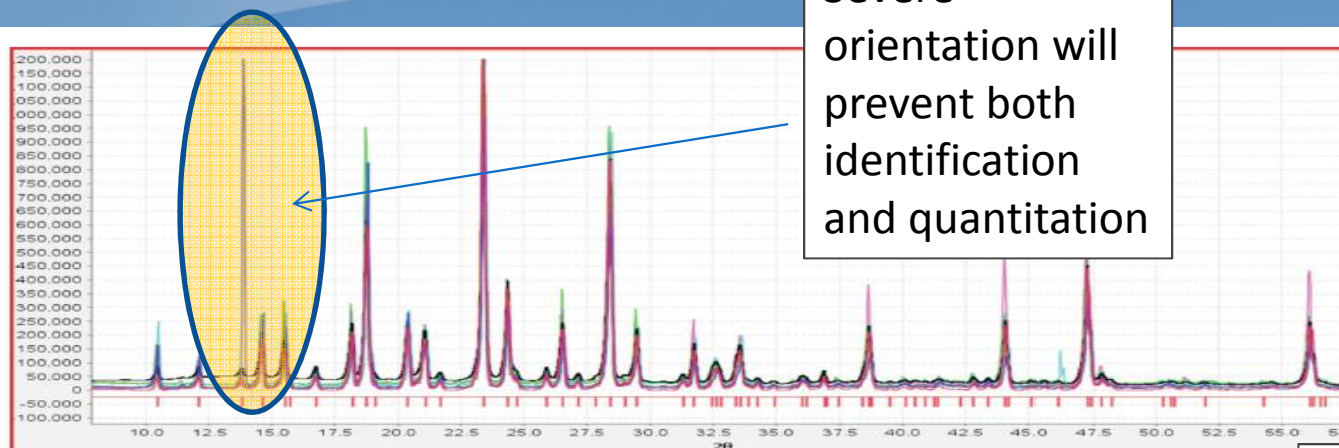
Effexor

# Orientation



# Impact on Intensities

Severe orientation will prevent both identification and quantitation



d-spacings relatively accurate, intensities and profiles vary considerably  
In poorly prepared samples

Figure 8. (Color online) (a) Superimposed data from seven laboratories for the three component mixture that were all scaled to the maximum peak. (b) Data from seven laboratories are superimposed for the three component mix. The outside two peaks are due to mannitol and the inside two peaks to acetaminophen. The off-scale peak is characteristic of severely oriented acetaminophen.

# Best Practices

- Grind or attrite to product random oriented specimens without specimen damage
- Increase collection times and optimize S/N for low Z, low density materials
- Use zero background holders especially if you are analyzing amorphous and non-crystalline components
- Analyze peak shapes to detect granularity (*know your instrumental peak shape by running a NIST standard*)
- Analyze relative peak intensities to detect orientation
- Run multiple specimens and /or separations (compare or cluster the data)



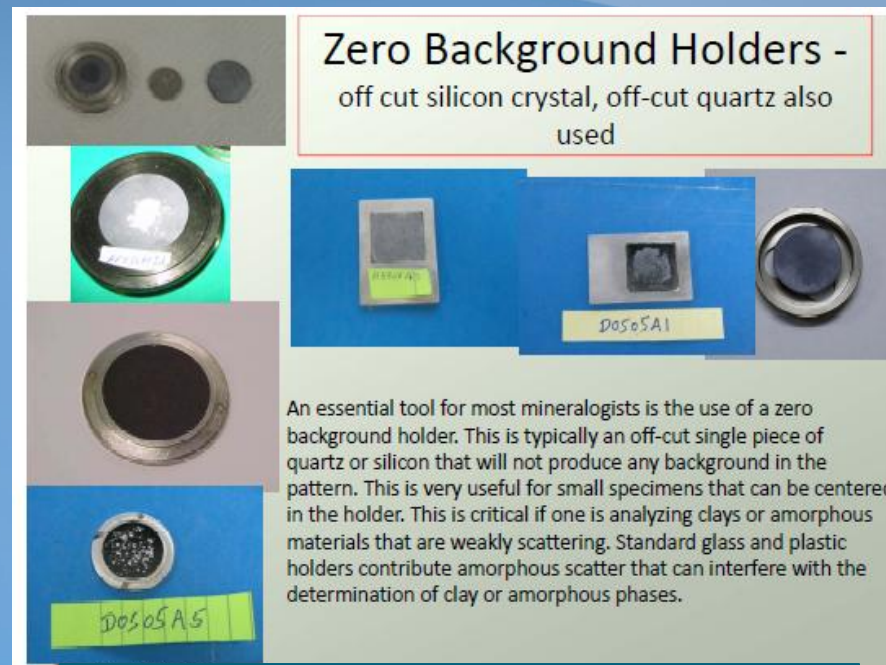
# Tutorial – How to Analyze Drugs ([www.icdd.com/resources/tutorials](http://www.icdd.com/resources/tutorials))



On left: Particle sieves of 100, 200 and 600 mesh

On right: Micronizing chambers from a McCrone micronizing mill using either agate media or alumina media (light pink)

## How to Analyze Minerals



Zero Background Holders -  
off cut silicon crystal, off-cut quartz also  
used

An essential tool for most mineralogists is the use of a zero background holder. This is typically an off-cut single piece of quartz or silicon that will not produce any background in the pattern. This is very useful for small specimens that can be centered in the holder. This is critical if one is analyzing clays or amorphous materials that are weakly scattering. Standard glass and plastic holders contribute amorphous scatter that can interfere with the determination of clay or amorphous phases.

Kaptan capillaries for capillary mounts

For pharmaceutical polymeric attrition media (Polystyrene, Polyethylene) used for soft materials

# METHODS OF ANALYSIS



# ICDD Global User Surveys

	<u>Responses</u>	<u>Queries</u>	
2012 Survey	1031	9956	10.3%
2009 Survey	811	6950	11.6 %
2006 Survey	220	1305	6 %

All survey data are in a relational database format for data mining purposes – We can correlate responses to methods used with responses to success in particular analyses

# Success in Material Identification

	PDF-4+	Good	Average	Bad	Ugly
	<u>&gt;95 %</u>	<u>&gt;95</u>	<u>&gt;90 %</u>	<u>75-90 %</u>	<u>75% &lt;</u>
Number of Reponses	63	132	351	192	61
Age					
Age >50	42.9	43.2%	41.6%	35.9%	41
21-30	4.8	7.6%	11.7%	10.4%	18
Experience					
Primary XRD Analysis	91.8	87.3%	82.9%	78.7%	73.7%
Frequency Phase ID	88.7	81.0%	79.6%	68.6%	70.0%
Tools					
PDF-2 Usage	20.6	48.8%	61.3%	61.8%	61.8%
PDF-4+ Usage	100.0	48.8%	39.6%	34.9%	38.2%
Technique					
Element Filters	74.0	64.6%	65.9%	51.9%	45.0%
Subfiles	53.3	50.0%	48.6%	27.1%	31.3%
GOM	48.7	40.0%	32.9%	24.8%	35.7%
d- Match	37.8	41.1%	31.4%	23.1%	40.0%
Frequently Don't Use Filters	32.4	38.2%	36.7%	34.0%	40.0%
Frequently do pattern simulations	30.0	32.2%	14.8%	12.3%	10.0%

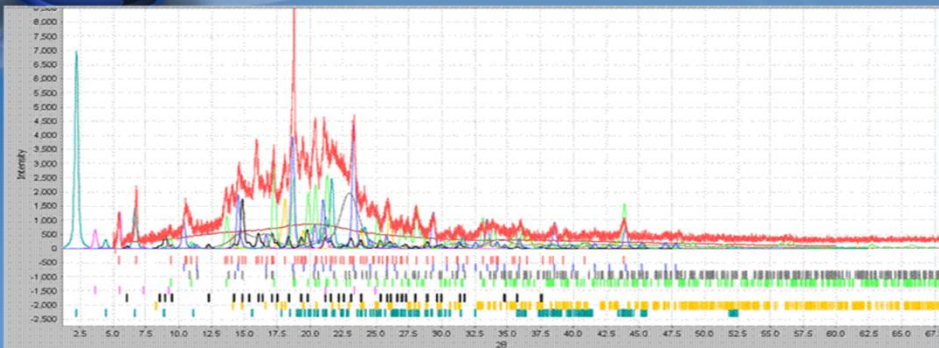
# Database Users

Analyzing Organics  
is more difficult

	PDF-4+	PDF-4/Organics	PDF-4/Minerals	PDF-2		PDF-2 Best	PDF-4+ Best
Total Responses	314	54	94	468		63	63
<b>Q3</b>							
Academic	41.1	24.1	36.1	45.7		41.9	34.9
Industrial/Man/serv.	29.9	38.9	25.5	29.7		36.5	31.8
<b>Q7</b>							
Primary XRD	83.6	87.5	79.5	80.1		85.2	91.8
Primary ED	31.4	22.2	23.4	28.4		27.8	40.0
<b>Q 10</b>							
Phase ID	59.2	54.0	60.9	61		63.8	58.6
Quant Rietveld	9.9	4.0	6.9	6.8		5.2	13.8
Rietveld Structure	5.1	6.0	3.4	6.4		3.4	0.0
<b>Q 11</b>							
Freq XRD Phase ID	73.0	64.0	65.9	69.4		82.3	86.4
Rietveld Quant	35.0	29.3	30.3	22.4		26.7	40.0
Crystallinity	21.8	20.9	13.6	14.8		22.4	36.7
Crystallite Size	24.4	15.2	15.6	18.9		13.5	36.5
<b>Q 18</b>							
Phase ID							
>95%	21.1	18.4	24.2	14.1		100.0	100.0
>90	45.3	40.8	49.5	46.8			
>75	21.8	32.7	18.7	25.7			
75<	7	6.1	3.3	7.6			
<b>Q 19</b>							
Elements	67.7	60.5	54.8	61.3		67.3	74.0
Subfiles	46.4	43.2	58.6	41		55.8	53.3
GOM	35.2	35.5	38.8	30.7		40.0	48.7
Ref qual	39	48.6	40.4	31.3		40.5	37.8

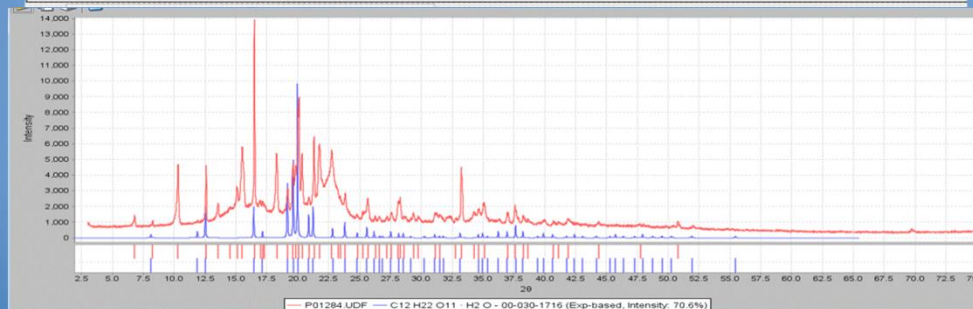


# Difficult Problems



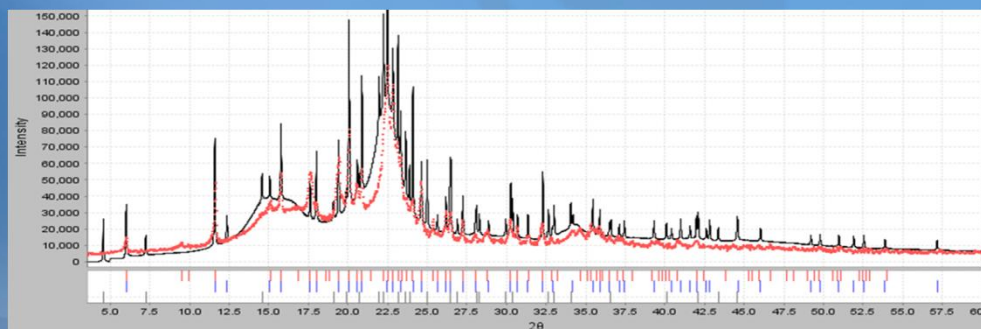
## Allegra

- Complex Formulation
- Heavy Peak Overlap
- Contains amorphous and nano crystalline materials



## Effexor (and Lipitor)

- Major phase (lactose monohydrate) is heavily oriented
- Nanocrystalline material



## Peppid AC

Due to nano and amorphous content this is difficult to analyze with PDF-2, easy to analyze with PDF-4

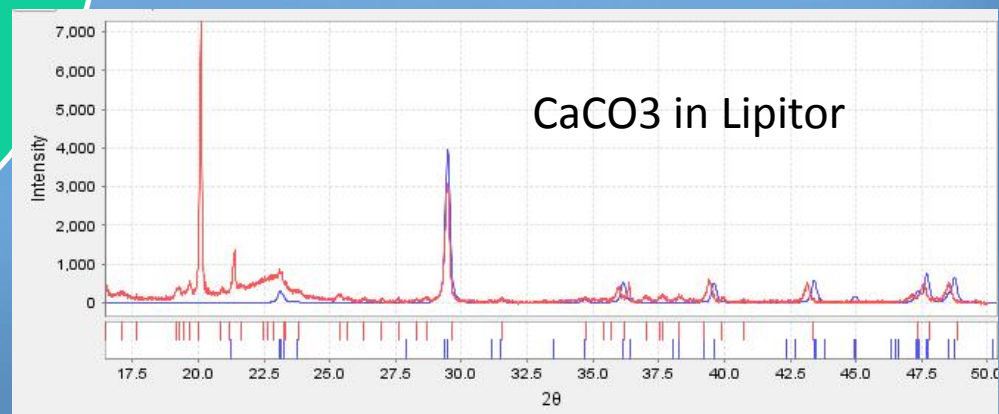
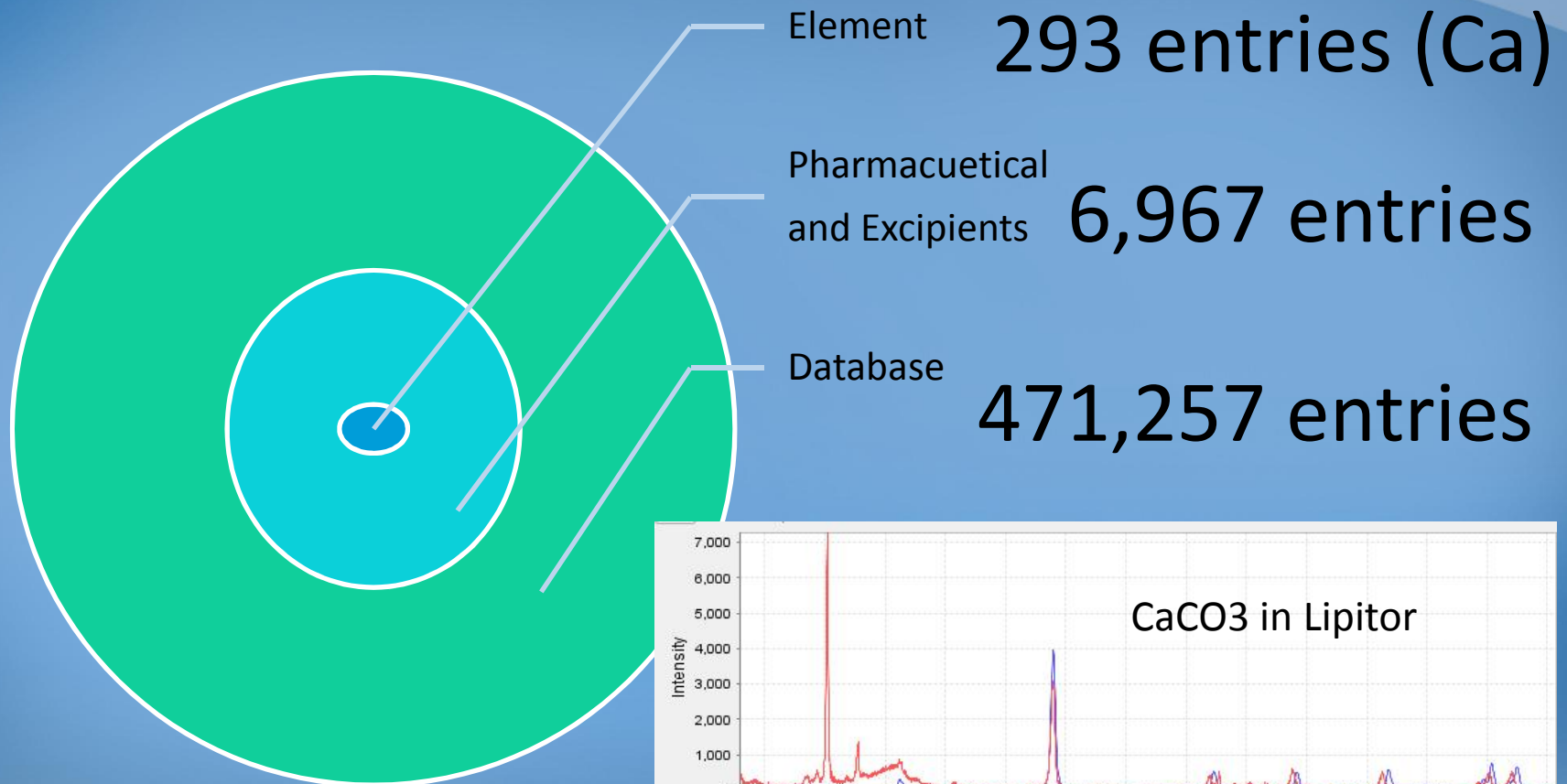


# Database Tools for improved formulation analysis

- *Subfiles on Excipients and Pharmaceuticals*
- Elemental Filters - All data expressed in atomic and weight %'s
- *Full experimental digital patterns for microcrystalline polymers and amorphous materials*
- *Common inorganic salts and oxides*
- *Evaluate and correct disordered structures*
- *Actively collect new pharmaceutical materials through grants*



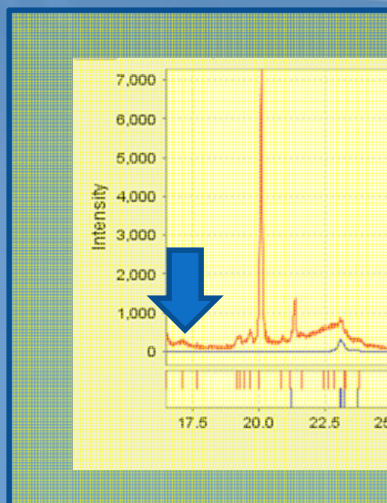
# Use of Filters



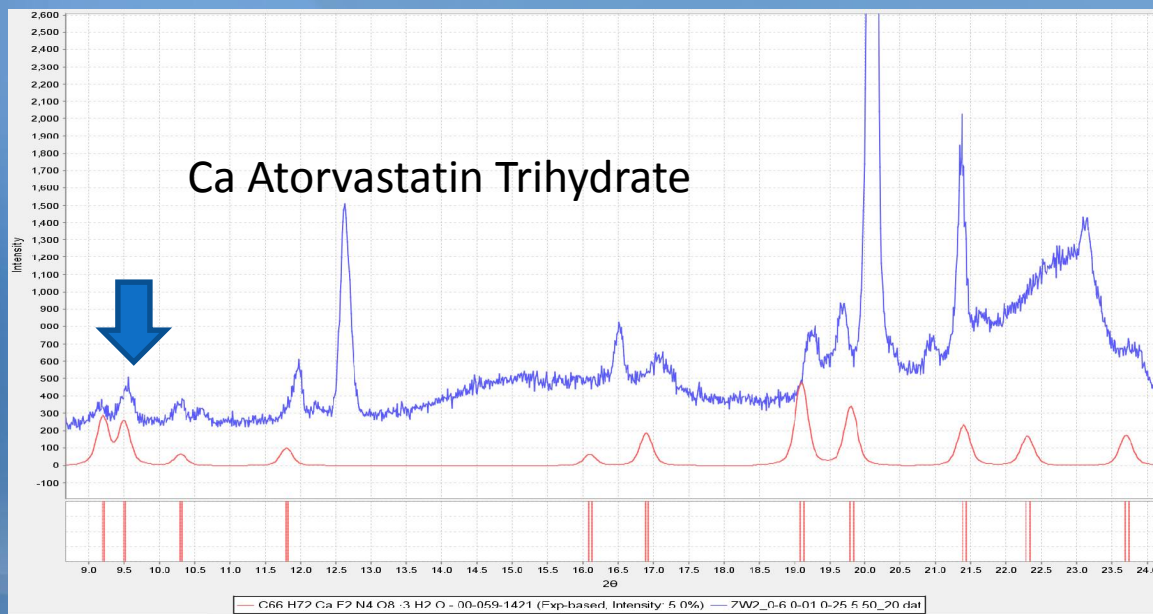
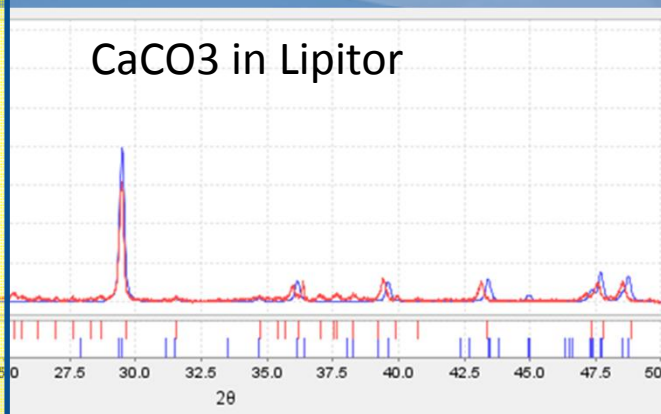


# Use of Filters - Lipitor

Subfile and Element  
Filter



CaCO3 in Lipitor



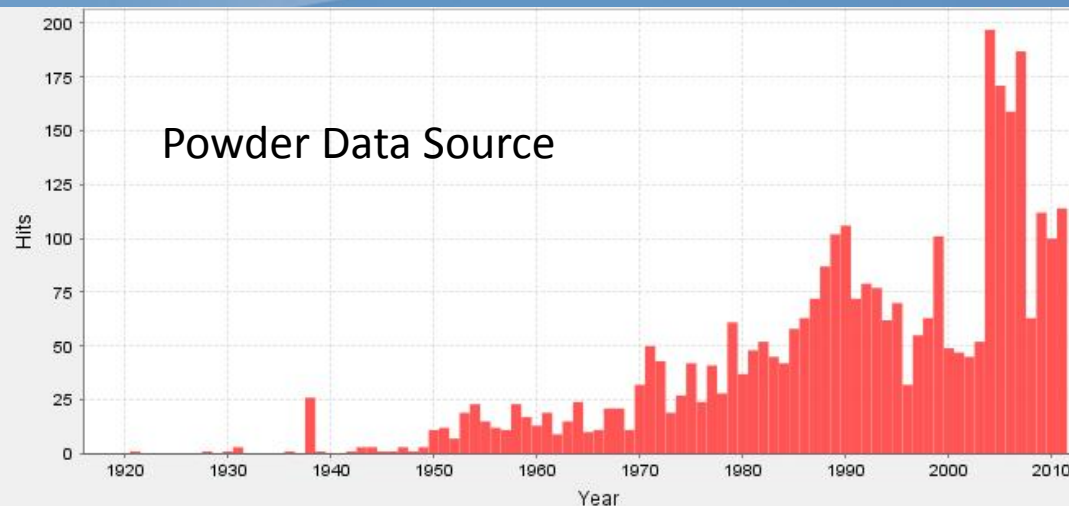
Ca Atorvastatin Trihydrate

d-spacing filter – 9.66 Å  
293 Entries

8 Entries  
4 are Ca Atorvastatin

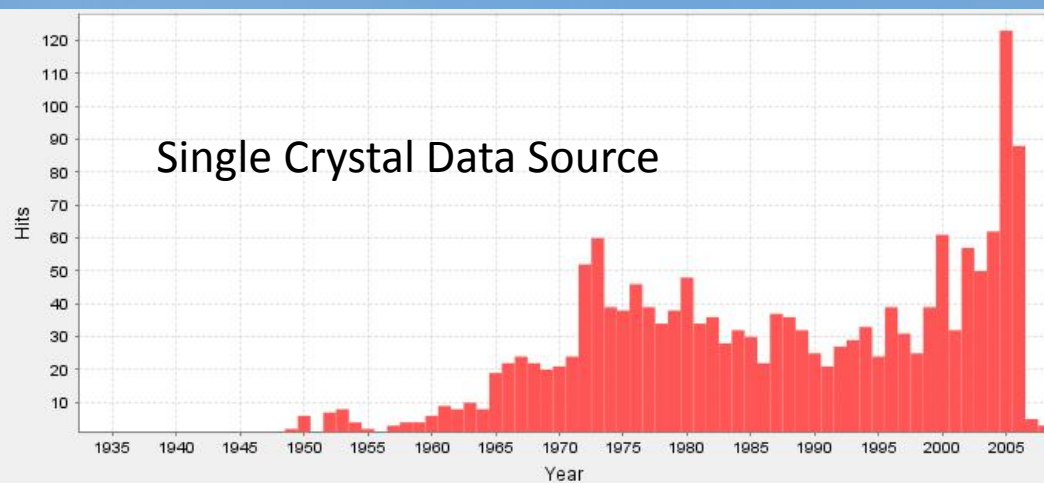
# Subfiles - Pharmaceuticals

Powder Data Source



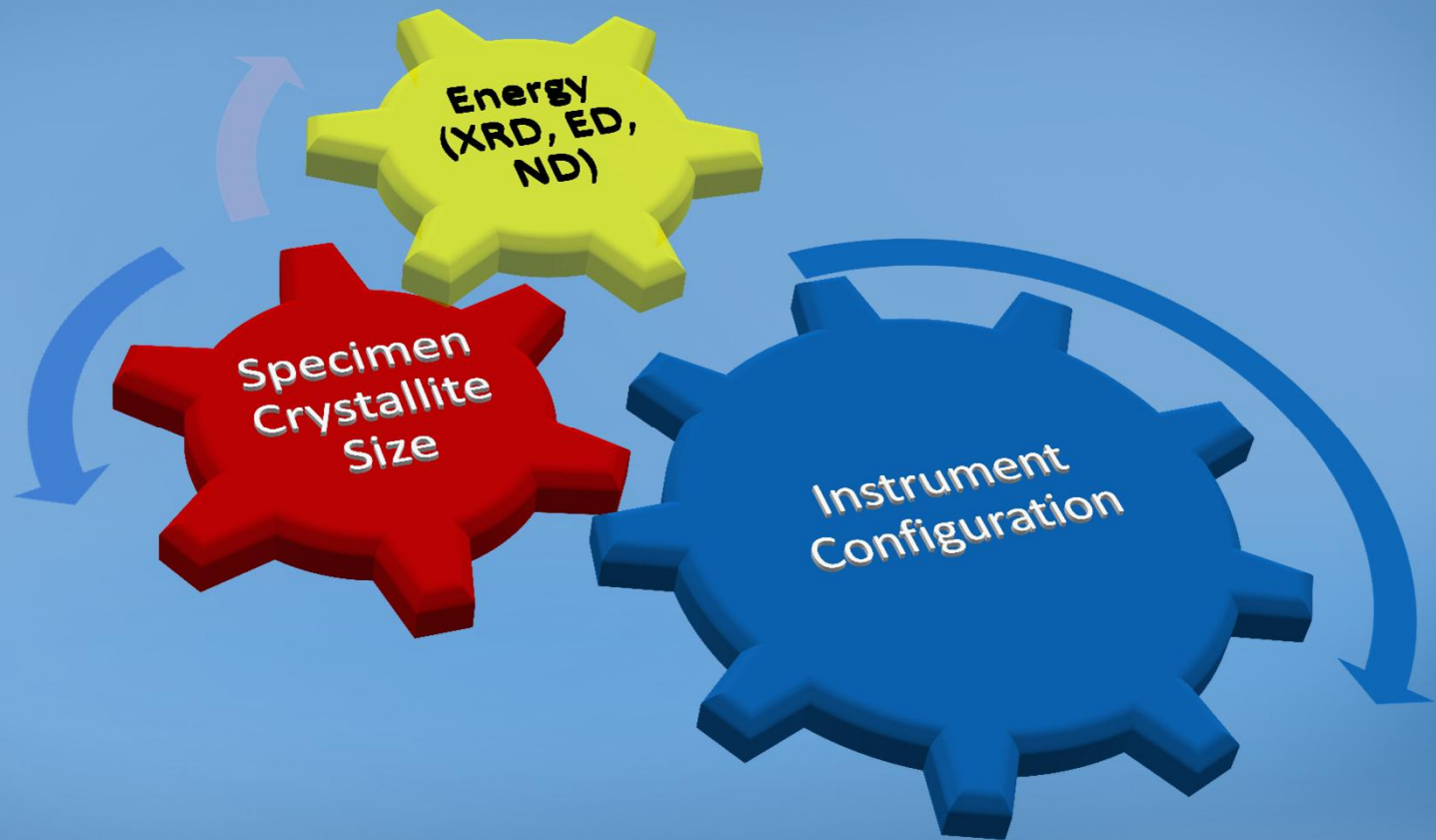
ICDD data sourced from powder diffraction experiments – many grants and donations

Single Crystal Data Source



ICDD data sourced from CSD single crystal data – primarily literature publications

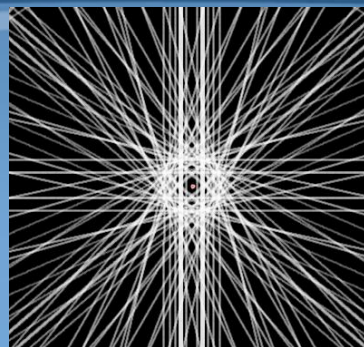
# Pattern Simulation



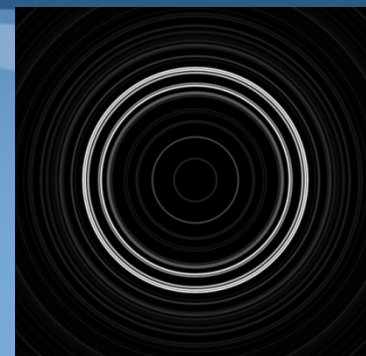


# Shao-Fan Lin Dropropizine

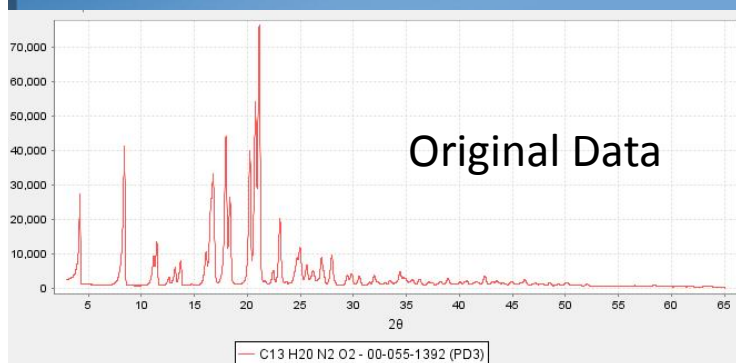
*53 % of PDF-4/Organics users  
have a 2D detector*



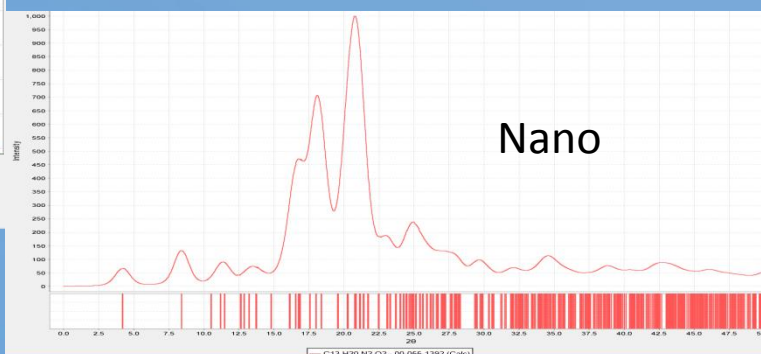
EBSD  
Pattern



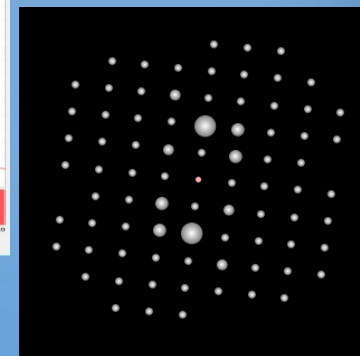
2D XRD



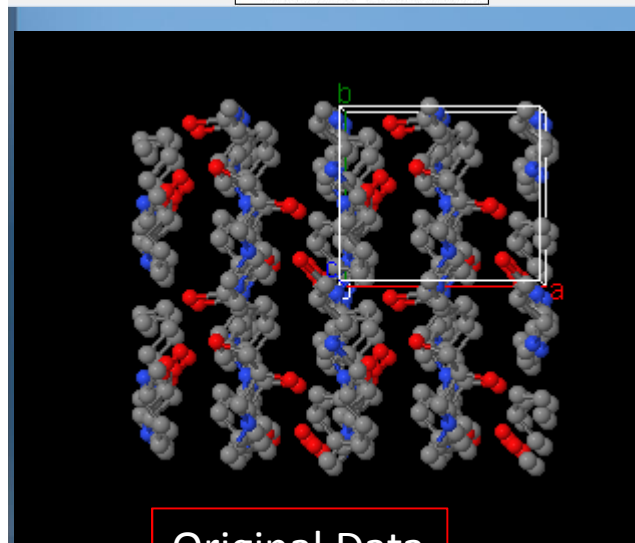
Original Data



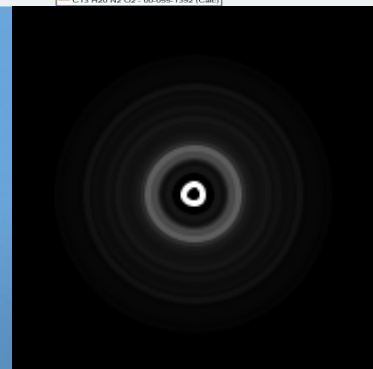
Nano



Electron Diffraction  
Spot Pattern (001)



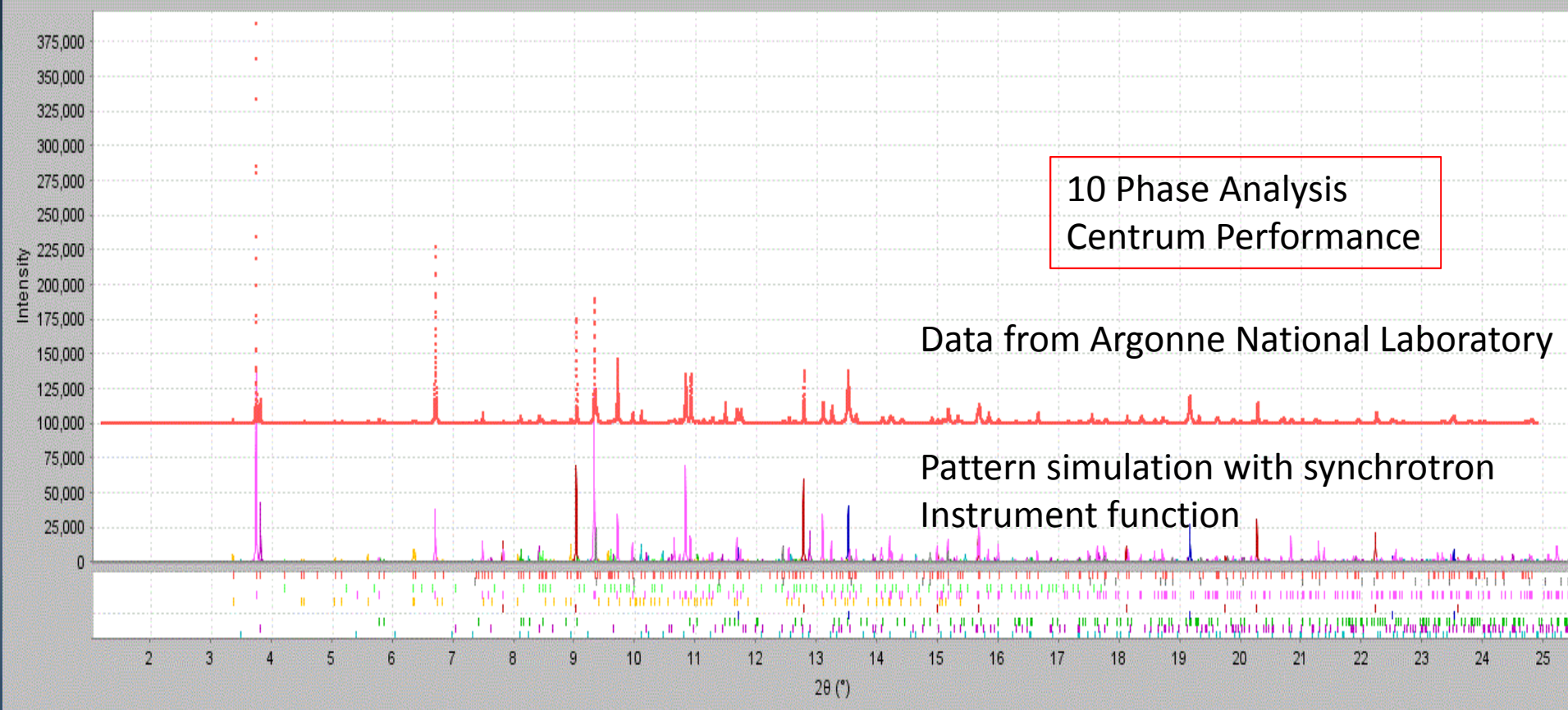
Original Data



Electron Diffraction (100 KeV)



# Synchrotron Analyses



\*centrum.asc — Ca (C O<sub>3</sub>) - 01-072-4582 (Calc, Intensity: 8.1%) — Ca P O<sub>3</sub> (O H) - 00-009-0080 (Exp-based, Intensity: 3.0%) — H Ca (P O<sub>4</sub>) (H<sub>2</sub> O)<sub>2</sub> - 04-013-3344 (Calc, Intensity: 46.0%)  
 C<sub>6</sub> H<sub>8</sub> O<sub>6</sub> - 00-022-1560 (Exp-based, Intensity: 4.0%) — K Cl - 01-076-3363 (Calc, Intensity: 22.7%) — Mg O - 04-005-4394 (Calc, Intensity: 13.2%) — Mn (S O<sub>4</sub>) (H<sub>2</sub> O) - 04-010-4027 (Calc, Intensity: 3.0%)  
 Fe (C<sub>4</sub> H<sub>2</sub> O<sub>4</sub>) - 00-062-1294 (Calc, Intensity: 14.0%) — Ca<sub>5</sub> (P O<sub>4</sub>)<sub>3</sub> (O H) - 04-007-2837 (Calc, Intensity: 4.0%)

~ 30% of PDF-4/Organics users use both a laboratory XRD unit and a synchrotron

# Centrum Performance- Goodness of Merit (GOM)

## High quality lab data

### GOM

Vit C	6159	C	1
Calcite	6145	I	2
Monetite	6143	I	3
KCl	5884	I	4
Fe Fumarate	5334	B	5
Brushite	2640	S	8
Nicotinamide	2517	I	15

95 d-spacings in the pattern

GOM maximum = 8000  
High scores better fit

## Synchrotron Data

### GOM

Calcite	7519	I	1
Monetite	7477	I	2
Brushite	7469	S	3
Vitamin C	7384	C	4
KCl	7166	A	5
MgO	6779	I	12
MnSO4 Hydrate	5230	I	2
Fe Fumarate	4321	R	8
Hydroxyapatite	4892	I	2
Nicotinamide	3161	I	8

More intensity and resolution  
203 d-spacings in the pattern  
Peaks are well separated

# Success in Identification

Technique	Greater than 95 %			Less than 75 %
Element Filters	74.0	64.6%		45.0%
<i>Subfiles</i>	53.3	50.0%		31.3%
GOM	48.7	40.0%		35.7%
d- Match	37.8	41.1%		40.0%
Frequently Don't Use Filters	32.4	38.2%		40.0%
Frequently do pattern simulations	30.0	32.2%		10.0%

## Best Methods

Use elemental filters  
 Use subfiles  
 Use GOM  
 Use pattern simulation

# Rietveld Refinements for improved phase identification

**J. Kaduk**, “A Rietveld Tutorial – mullite”

Powder Diffr., Vol. 24, No. 4, December 2009

T. G. Fawcett, C. E. Crowder, S. N. Kabekkodu, **J. A. Kaduk**,  
“Improved Material Identification through Targeted Data Mining”  
Advances in X-ray Analysis, 2010 (with on-line tutorial, [icdd.com](http://icdd.com))

**J. Kaduk**, “Crystallographic Databases and Powder Diffraction”  
Volume H, International Tables, IUCr, in progress.

Routinely identifies phases in 0.1-5 weight % concentration  
with good quality laboratory data



# Rietveld Refinements for improved phase identification

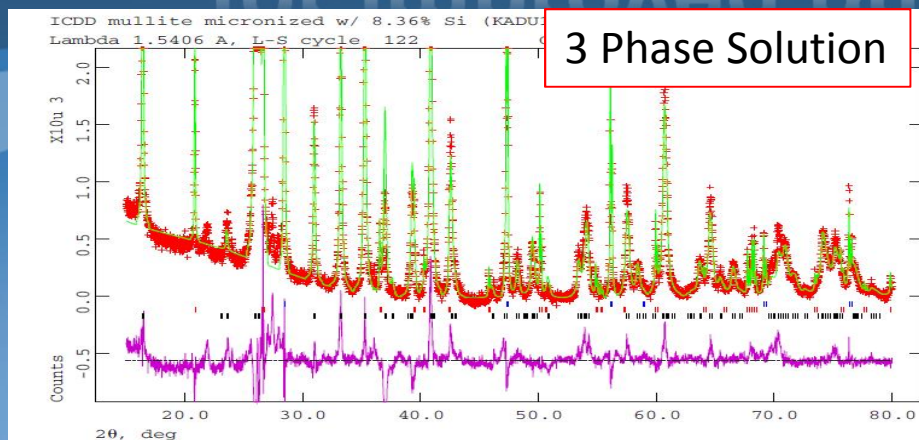
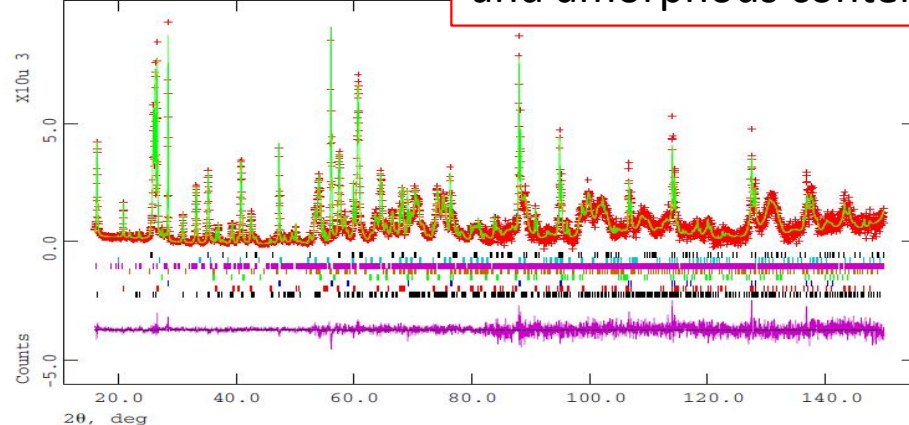
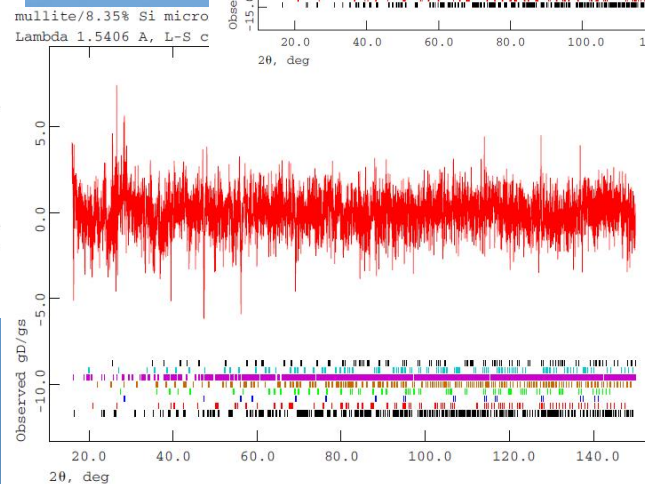
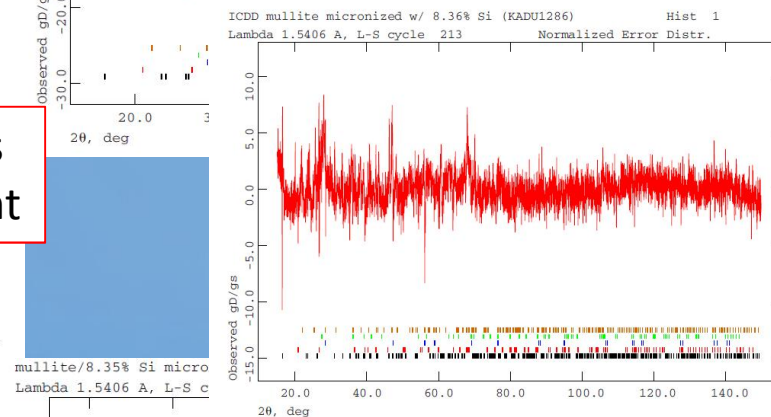
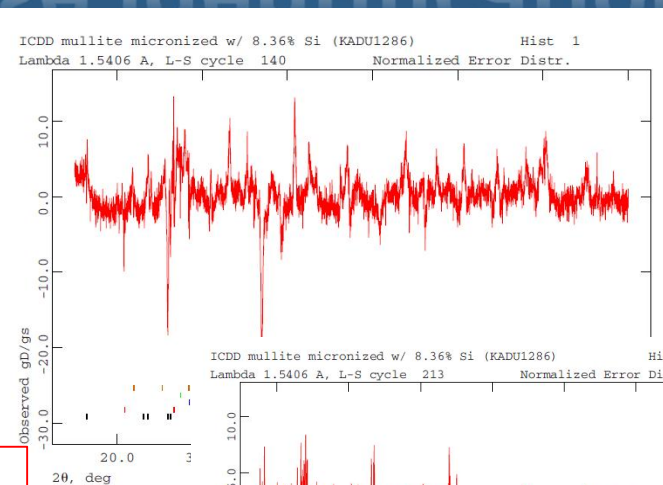


Figure 9. (Color online) A portion of

mullite/8.35% Si micronized blend (KADU1286)  
Lambda 1.5406 Å, L-S cycle 372



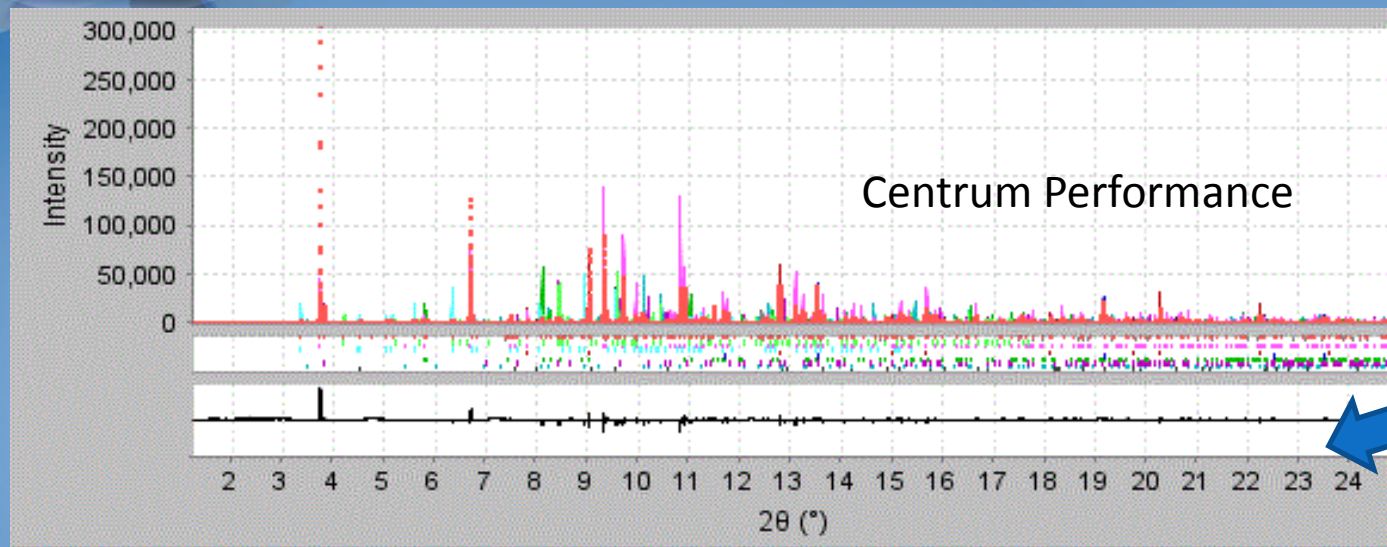
Eight crystalline phases  
and amorphous content



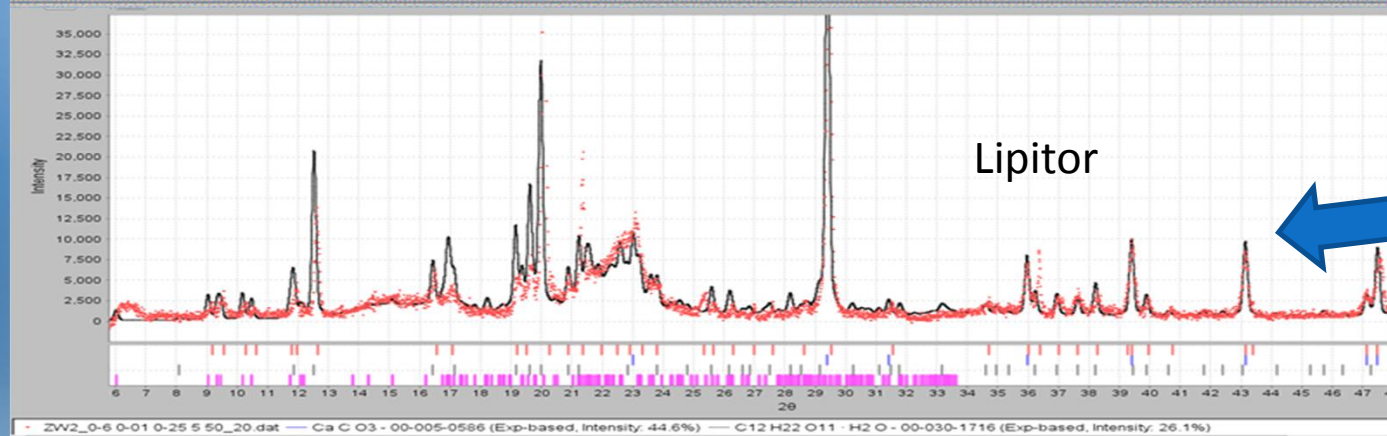
J. Kaduk, A Rietveld Tutorial - mullite  
Powder Diffr., Vol. 24, No. 4, December 2009



# Pattern Simulations (not a refinement)



Difference Plot  
New in 2013



Summed component  
Plot (New in 2012)

# Cluster Analyses

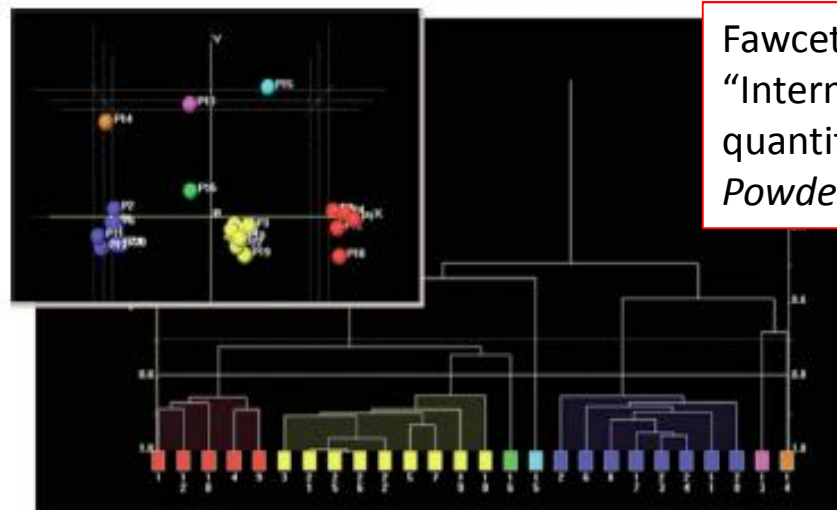


Figure 3. (Color online) Dendrogram and principal component analysis of 21 data sets from seven laboratories.

Fawcett, T. G., Needham, F., Faber, J., Crowder, C. E., (2010), "International Centre for Diffraction Data round robin on quantitative Rietveld phase analysis of pharmaceuticals", *Powder Diffraction*, 25(1), pp 60-67.

## APPLICATION OF HIGH-THROUGHPUT CLUSTER ANALYSIS TO MULTIPLE DATA TYPE – COMBINING DATASETS

G.J. Cunningham, G. Barr, W. Dong, C.J. Gilmore, WestCHEM, Department of Chemistry, University of Glasgow, Glasgow G12 8QQ, Scotland; C. Frampton, Pharmorphix Limited, 250 Cambridge Science Park, Milton Road, Cambridge, CB4 0WE

HIGH-THROUGHPUT POWDER DIFFRACTION - THE SEARCH FOR POLYMORPHS	PPXRD-2	65,348
HOW TO COMBINE PXRD, RAMAN AND OTHER 1-D DATA IN HIGH THROUGHPUT POLYMORPH/SALT/CO-CRYSTAL STUDIES	PPXRD-6	23,954
MANY PATTERNS AND MANY METHODS: NEW METHODS UTILISING MULTIPLE ANALYSIS TECHNIQUES IN POLYMORPH AND SALT SCREENING SYSTEMS	PPXRD-8	6,478
PXRD WITH RAMAN SPECTROSCOPY, DSC AND IR DATA	PPXRD-9	18,973
QUANTITATIVE ANALYSIS OF MIXTURES USING HIGH THROUGHPUT INSTRUMENTATION WITHOUT THE USE OF STANDARDS	PPXRD-6	120,457



# Cluster Analyses

## Reference materials for the study of polymorphism and crystallinity in cellulosics

T. G. Fawcett,<sup>1,a)</sup> C. E. Crowder,<sup>1</sup> S. N. Kabekkodu,<sup>1</sup> F. Needham,<sup>1</sup> J. A. Kaduk,<sup>2</sup> T. N. Blanton,<sup>3</sup> V. Petkov,<sup>4</sup> E. Bucher,<sup>5</sup> and R. Shpanchenko<sup>6</sup>

<sup>1</sup>International Centre for Diffraction Data, Newtown Square, Pennsylvania

<sup>2</sup>Illinois Institute of Technology, Naperville, Illinois

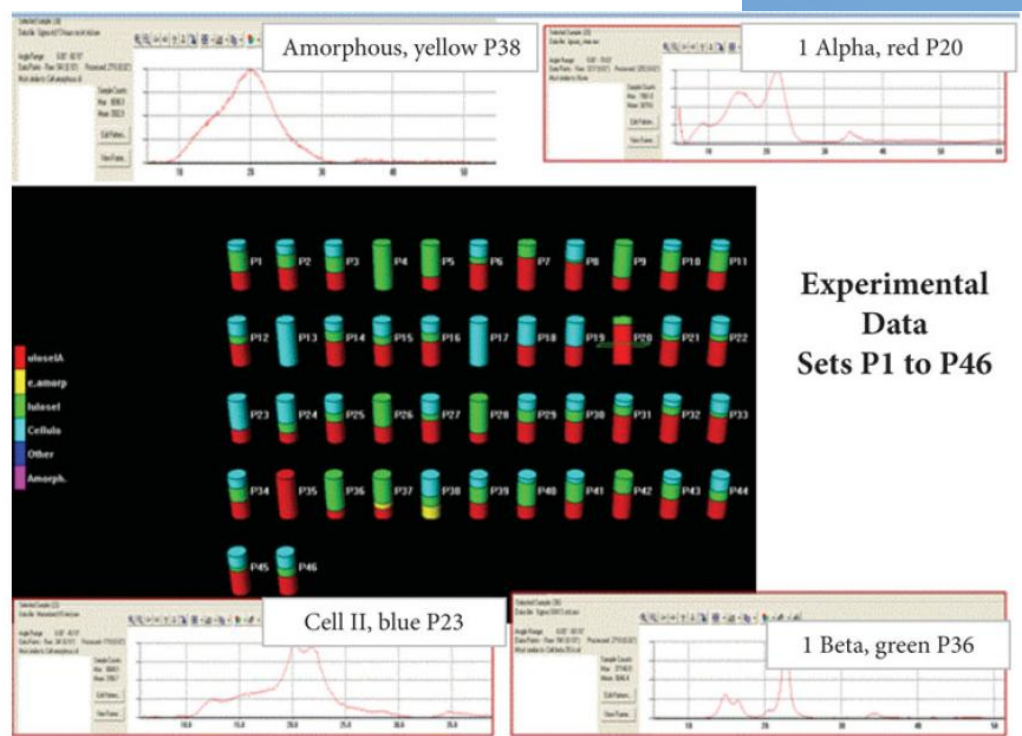
<sup>3</sup>Eastman Kodak Company, Rochester, New York

<sup>4</sup>Central Michigan University, Mt. Pleasant, Michigan

<sup>5</sup>International Paper Company, Loveland, Ohio

<sup>6</sup>Moscow State University, Moscow, Russia

Powder Diffr., Vol. 28, No. 1, March 2013



## Polymorphs Identified

Cellulose  
AZT  
Carbamazepine  
Heroin  
Calcium Phosphates  
Glycine  
Codeine  
Famotidine  
Cocaine  
Cimetidine

*Cluster analysis doesn't depend on the specimen being crystalline !*



# Cluster Analyses

MAPPING DRUG CHEMISTRY FOR AZT AND CARBAMAZEPINE BY  
COMBINING CLUSTER ANALYSIS AND DIFFRACTION DATABASES

Timothy G. Fawcett, Soorya N. Kabekkodu

International Centre for Diffraction Data  
Newtown Square, PA, 19073

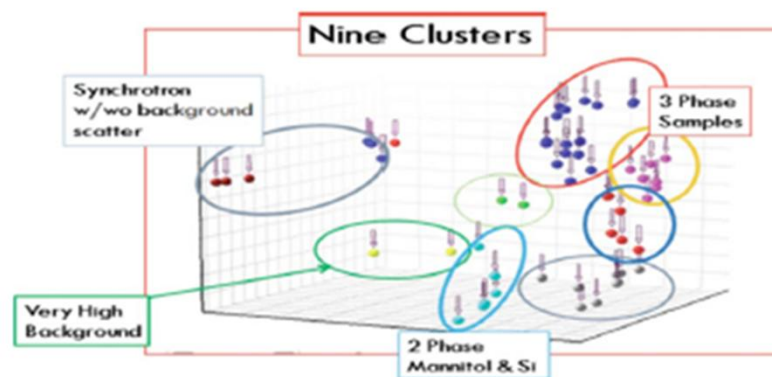


Figure 4. (Color online) PCA of 63 data sets from nine laboratories. The comments were added by the authors.

*Declustering multiple data sets to get component contributions*

**Invited** – CRYSTAL STRUCTURE SOLUTION FROM  
POWDER MIXTURES: THE PXRD-BTEM-RIETVELD  
METHOD

**Marc Garland\***, L.Guo, M. Schreyer, S.  
Thirunahari  
ICES, Jurong Island, Singapore

# Outline – Best Practices

- Specimen Preparation-
- Round Robin, Experience

- General issues – low scattering power, absorption difference in ingredients, crystallinity
- Asymmetric Crystals
- Particle size
- Granularity
- Orientation
- Impact on Intensities
- Separation Techniques

- Methods of Analysis –
- Customers Survey Data

- Experience
- Databases
- Noncrystalline Materials
- Reference Quality
- Elemental Filters
- Subfiles
- Pattern Simulations
- Rietveld Analysis
- Combined Techniques – DXC/XRD
- Complimentary Analytical Techniques



# Best Practices

- Focus on specimen preparation – good data lead to good analyses
- Formulation analyses are difficult – use all your tools
  - Subfiles
  - Complimentary analytical data – Element filters
  - Data analysis tools – Figures of Merit, Reference quality, Intensity and d-spacing matches
  - Data processing techniques (Rietveld reiteration, different search/match algorithms, cluster analyses, simulations especially with non-crystalline materials)







# Ingredients

Dibasic Calcium Phosphate , Potassium Chloride , Ascorbic Acid (Vit. C) , Microcrystalline Cellulose , Calcium Carbonate , DI-Alpha Tocopheryl Acetate (Vit. E) , Magnesium Oxide , Ginkgo Biloba Leaf (Ginkgo Biloba) Standardized Ex , Gelatin , Ginseng Root (Panax Ginseng) Standardized Extract , Ferrous Fumarate , Niacinamide , Crospovidone , Starch , Zinc Oxide , Calcium Pantothenate , Silicon Dioxide , Manganese Sulfate , Pyridoxine Hydrochloride (Vit. B6) , Riboflavin (Vit. B2) , Thiamin Mononitrate (Vit. B1) , Cupric Oxide , Vitamin A Acetate , Beta Carotene , Chromium Chloride , Folic Acid , Potassium Iodide , Sodium Molybdate , Sodium Selenate , Boron , Biotin , Phytonadione (Vit. K) , Sodium Metavanadate , Nickelous Sulfate , Stannous Chloride , Cyanocobalamin (Vit. B12) , Ergocalciferol (Vit. D2) , Acacia Gum , Ascorbyl Palmitate , Butylated Hydroxytoluene (BHT) , Citric Acid , DI-Alpha Tocopherol (Vit. E) , FD&C Red No. 40 Aluminum Lake , FD&C Yellow No. 6 Aluminum Lake , Glucose , Hydroxypropyl Methylcellulose , **Lactose , Magnesium Stearate** , Polyethylene Glycol , Polysorbate 80 , Potassium Sorbate , Sodium Aluminum Silicate , Sodium Ascorbate , Sodium Benzoate , Sodium Citrate , Sorbic Acid , Sucrose , Titanium Dioxide , Tribasic Calcium Phosphate , Water , Maltodextrin , Maltodextrin May Also Contain

Vitamin A (20% as Beta Carotene)	70.0%	%
Vitamin C 120.0 mg	200.0%	%
Vitamin D	100.0%	%
Vitamin E	200.0%	%
Vitamin K 25.0 mcg	31.0%	%
Thiamin 4.5 mg	300.0%	%
Riboflavin 5.1 mg	300.0%	%
Niacin 40.0 mg	200.0%	%
Vitamin B6 6.0 mg	300.0%	%
Folic Acid 400.0 mcg	100.0%	%
Vitamin B12 18.0 mcg	300.0%	%
Biotin 40.0 mcg	13.0%	%
Pantothenic Acid 10.0 mg	100.0%	%
Calcium 100.0 mg	10.0%	%
Iron 18.0 mg	100.0%	%
Phosphorus 48.0 mg	5.0%	%
Iodine 150.0 mcg	100.0%	%
Magnesium 40.0 mg	10.0%	%
Zinc 15.0 mg	100.0%	%
Selenium 70.0 mcg	100.0%	%
Copper 2.0 mg	100.0%	%
Manganese 4.0 mg	200.0%	%
Chromium 120.0 mcg	100.0%	%
Molybdenum 75.0 mcg	100.0%	%
Chloride 72.0 mg	2.0%	%
Potassium 80.0 mg	2.0%	%
Ginseng Root (Panax Ginseng) Standardized Extract 50.0 mg	%	%
Ginkgo Biloba Leaf (Ginkgo Biloba) Standardized Extract 60.0 mg	%	%
Boron 60.0 mg	%	%
Nickel 5.0 mcg	%	%
Silicon 4.0 mg	%	%
Tin 10.0 mcg	%	%
Vanadium 10.0 mcg	%	%

Vitamin C 120 mg  
 Niacin (B3) 40 mg  
 Root Extracts (Celluloses) >100 mg  
 Riboflavin (B2) 5.1 mg  
 Folic Acid 400 mcg  
  
 K,Cl in KCl 147 mg  
 Ca in CaCO<sub>3</sub>, CaHPO<sub>4</sub> 100 mg  
 P in CaHPO<sub>4</sub> 48 mg  
 Fe in Iron Fumarate 18 mg  
 Zn in ZnO 15 mg  
 Mg in MgO, MgStearate 40 mg  
 Si in SiO<sub>2</sub> 4 mg  
 Mn in MnSO<sub>4</sub> 4 mg  
 TiO<sub>2</sub>

4-10 mg  
 Vit B6 HCl  
 Calcium Pantothenate  
 (overlap with Vit C)  
  
 60 mg  
 Boron