Characterization of amorphous pharmaceuticals – what can you do in the home lab?



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Motivation Fingerprinting APIs

- Powder X-ray diffraction (PXRD) is a powerful tool to characterize active pharmaceutical compounds in the solid state
 - This is typically limited to crystalline compounds



Crystalline ketoprofen



Amorphous ketoprofen



Motivation Fingerprinting Amorphous APIs



• The pair distribution function (PDF) provides structural information for amorphous materials, enough to allow for a fingerprint ^[1-3]:



- Measurements carried out at a synchrotron (30 min) ^[1,2], or potentially a lab source, but requiring very long measurement times (hours - days)
- Is it possible to collect PDFs of APIs on a lab instrument, fast enough to allow screening?

^[1] Billinge S.J.L. et al., CrystEngComm, 2010, 12, 1366-1368

^[2] Dykhne T., Taylor R., Florence A., Billinge S.J.L., Pharm. Res., 2011, 28:1041-1048

^[3] Total Scattering Pair Distribution Functions (TSPDF) for Fingerprinting Amorphous Pharmaceuticals S. Billinge, PPXRD-10 (2011)



Si lattice

Start at an arbitrary atom, and look outward

Peak appears at nearest neighbor



Radial atomic **P**air **D**istribution **F**unction (PDF) gives the interatomic distance distribution, or "probability" of finding atomic pairs of distance *r* apart.



Si lattice



Start at an arbitrary atom, and look outward

Another peak appears at the next neighbor Radial atomic **P**air **D**istribution **F**unction (PDF) gives the interatomic distance distribution, or "probability" of finding atomic pairs of distance *r* apart.





Si lattice

Start at an arbitrary atom, and look outward

And so on...

Radial atomic Pair Distribution Function (PDF) gives the interatomic distance distribution, or "probability" of finding atomic pairs of distance r apart.





Si lattice



This process is repeated for all atoms, leading to a probability distribution as a function of interatomic distance In another way, it's a histogram of interatomic distances, normalized by the scattering contribution of each atom (form factor)



PDF data collection Challenges



Some requirements:

High energy X-rays to collect data to high Q



- Good counting statistics
- Low instrument background
- These points (particularly the first 2) are why measurement times are typically so long

Experimental Setup D8 ADVANCE





- Mo radiation is in the 'sweet spot' for PDF on pharmaceuticals
 - $\lambda = 0.7108 \text{ Å}; \text{ Q}_{\text{max}} \approx 17 \text{ Å}^{-1}$ (minimum necessary for PDF analysis)
 - Stronger interaction with weak scatterers (compared to Ag, synchrotron)

Validation of experimental setup Silicon powder



Si powder 1 hour data collection

Measurement strategy:

- frames every 10° 2θ
- count longer at high angles







Information content of the PDF Ketoprofen



- 2 main regions of bonding for molecules
- Small differences between 4 6 Å (cryst. vs amorph.)
- No packing signal (beyond 10 Å) in amorphous





Ketoprofen, nonsteroidal antiinflammatory drug

re-crystallized (from amorphous)

amorphous (melt-quenched)

crystalline (as received)

PDF refinement in TOPAS v6 Crystalline Ketoprofen



- Molecules can be difficult to model (due to overlap region)
- Rigid body for molecule including torsion angles
- Small FWHM < 15 Å (\approx size of molecule)
- Larger FWHM for entire range



PDF fit of crystalline ketoprofen in TOPAS



Packing motif of ketoprofen

'inter and intramolecular interactions prm intra 0.18713` prm inter 0.87715` pdf_for_pairs * * pdf_only_eq_0 pdf_gauss_fwhm = If(X<15, intra, 0); pdf_for_pairs * * pdf_gauss_fwhm = inter;</pre>

Section of TOPAS input file





- Cluster analysis partitions data into sets based on their <u>similarity</u>
 - Similar patterns end up in the same cluster, dissimilar ones in different clusters
 - Quick way to analyze and compare sets of data
 - We looked at several different APIs (and one excipient), and tried to also make the amorphous form of each (8 samples in total)

Compound	Abbrev.	Formula	Form	Amorph?	
Carbamazepine	Carba	$C_{15}H_{12}N_2O$	III	No	
Paracetamol	Para	$C_8H_9NO_2$	I	No	
Ketoprofen	Keto	$C_{16}H_{14}O_{3}$		Yes*	
Indomethacin	Indo	$C_{19}H_{16}CINO_4$	Y	Yes	
Corn-starch	Starch	$C_{27}H_{48}O_{20}$		Nano	

* sample also recrystallized from amorphous

Cluster Analysis on XRD data DIFFRAC.EVA



- As a frame of reference, cluster analysis on the raw XRD data
- Match range: 2 40° 2θ
- Obviously the direct information available in the amorphous patterns is not enough to fingerprint



Cluster Analysis on PDF data DIFFRAC.EVA





- Cluster analysis on 8 PDF datasets
- Match range: 1 15 Å (*intramolecular bonding and* overlap region)
- PDFs of 2 amorphous materials are different enough that they cluster with their parent crystalline PDF

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Cluster Analysis on PDF data Visualizing the results





3D Metric Multidimensional Scaling

- Graphical representation of distances between objects
- Smaller distances → more similar

Correlation coefficients (> 0.80 in red)

	Þ	Rank:	Carba #1	Indo #2	Indo-amorph #3	Keto #4	Keto-amorph #5	Keto-recryst #6	Para #7	Starch #8	
		Carba #1	1	0.7421	0.6389	0.6802	0.5265	0.6593	0.8342	0.4751	
		Indo #2	0.7421	1	0.8029	0.6927	0.644	0.6693	0.6589	0.5114	
		Indo-amorph #3	0.6389	0.8029	1	0.743	0.766	0.7471	0.6133	0.5336	
		Keto #4	0.6802	0.6927	0.743	1	0.8078	0.9707	0.6613	0.568	
		Keto-amorph #5	0.5265	0.644	0.766	0.8078	1	0.8415	0.5762	0.6114	
		Keto-recryst #6	0.6593	0.6693	0.7471	0.9707	0.8415	1	0.6375	0.609	
		Para #7	0.8342	0.6589	0.6133	0.6613	0.5762	0.6375	1	0.5094	
		Starch #8	0.4751	0.5114	0.5336	0.568	0.6114	0.609	0.5094	1	





- The PDF of amorphous molecular materials can provide sufficient structure information for fingerprinting.
- Data suitable for PDF analysis of organic compounds were collected in as little as 1 hour on a D8 ADVANCE diffractometer with a large 2D detector (PILATUS3 R 100K).
- Tests on other materials, particularly those with multiple polymorphs that can also be prepared in the amorphous state would be a next step.



Innovation with Integrity

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